

Insights in Emerging $Ti_3C_2T_x$ MXene-Enriched Polymeric Coatings for Metallic Surface Protection: Advancements in Microstructure, Anti-Aging, and Electrochemical Performance

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Abstract

This study provides new insights into the development of high-performance MXene-reinforced coatings to strengthen polymeric nanocomposites by enhancing microstructure, anti-aging properties, corrosion resistance, and robustness. MXene nanoparticles, labeled 25C and 80C, were synthesized using two different methods and incorporated at concentrations ranging from 0.1 to 2.0 wt.% into epoxy composites. The results demonstrated that 80C MXene, characterized by its finer morphology and superior dispersion, significantly improved the composite's performance compared to 25C. Electrochemical Impedance Spectroscopy (EIS) tests, along with long-term exposure assessments, suggested that incorporating both types of MXene nanoparticles enhances the corrosion protection performance of epoxy coatings over time. Micro-CT analysis revealed that both types of MXene substantially reduced defects and voids in the polymeric matrix, resulting in enhanced protective performance. This void reduction confirms that the incorporation of both 25C and 80C MXene improves microstructural integrity by filling voids and creating a more continuous, uniform structure, particularly in samples with 0.1% to 1.0% MXene flakes. The findings also highlighted MXene's potential in modifying the anti-aging properties of epoxy by inhibiting free radical generation and enhancing the composite's resistance against corrosion. Both 25C and 80C MXene-epoxy groups exhibited a clear trend of diminishing free radical intensity with increasing MXene concentration up to 1.0%, with free radical intensity reduced by over 40% compared to neat epoxy. The relationship between MXene concentration and reinforcement was also investigated, revealing superior corrosion protection properties at concentrations of 0.5-1.0 wt.%. This research offers a profound understanding of MXene's potential in polymer-based composites, laying a foundation for future investigations aimed at utilizing MXene to achieve superior material properties for a wide range of applications, particularly in the realm of metallic surface protection.

Keywords: MXene, High-performance, Polymer-matrix composites (PMCs), Microstructures, Anti-aging,

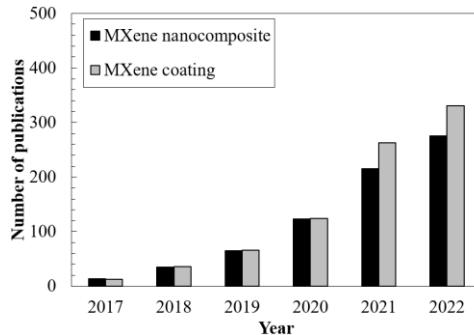
1 Introduction

Despite significant efforts in advancing polymeric coatings, particularly for corrosion protection and durability, premature coating failures are frequently encountered, leading to early-stage structural damage.

38 induced by corrosive attacks. MXene, a 2D graphene-like nanomaterial, exhibits exceptional physical and
39 chemical properties that have garnered significant attention from researchers and industries in recent years [1–
40 4]. With the exceptional 2D flake structure and chemical functionality, MXene and MXene-based materials can
41 be integrated into polymeric composites to fabricate high-performance materials that meet the requirements of
42 cutting-edge applications [5].

43 MXene-based nanomaterials are transition metal carbides and/or nitrides, which were discovered by
44 Naguib et al., in 2011 from Ti_3AlC_2 MAX phase, and have attracted more attention as a new material in the
45 recent decade [6,7]. MXenes were exfoliated from MAX phases, which not only inherited properties from
46 parent materials, but also simultaneously demonstrated other intriguing properties [1,8]. The MAX phase, which
47 combines metal and nonmetal elements under high-temperature conditions, shows high thermal and mechanical
48 stability, as well as electrical conductivity from the metallic bonding [1]. On the other hand, MXenes are layered
49 materials that have a metallic or semi-conducting behavior, and are known for their excellent electrical
50 conductivity, mechanical strength, and chemical stability. If used as nanofillers, MXene nanoparticles offer
51 several advantages over traditional nanoparticles in reinforcing polymeric coatings. Firstly, the 2D structure
52 and high aspect ratio of MXene can contribute to the formation of a robust network within the polymer matrix,
53 resulting in enhanced mechanical properties and barrier performance [9,10]. Secondly, the presence of diverse
54 surface functional groups on MXene enables stronger interfacial interactions between the nanofiller and the
55 polymer, which has the potential to improve the overall performance of the composite coatings [11].

56 Owing to their distinctive properties and wide-ranging potential applications, MXene-based polymeric
57 composites have drawn significant research attention in recent years, and many scientists are working to explore
58 their potential and develop new technologies based on these materials. MXenes are arguably the fastest growing
59 2D materials in the post-graphene era. Since their discovery in 2011 [12,13], they have attracted widespread
60 research interest due to their intriguing electrical, thermal, mechanical, and chemical properties [14]. Based on
61 the data retrieved from the Web of Science, the terms "MXene nanocomposite" and "MXene coating" in
62 scientific publications have seen a remarkable increased in recent years, showcasing an exponential growth of
63 the field, as evidenced by the yearly increase in the number of related papers published (Fig. 1).



64

65 Figure 1. The scientific publications with the keywords of “MXene nanocomposite” and “MXene coating”,
66 results were obtained from the Web of Science.

67 The findings in researchers’ previous investigations agreed that, 2D MXenes became an ideal nanofiller
68 for fabricating high-performance polymer composites that can meet the electrical, thermal, mechanical, flame-
69 retardant and other intriguing properties [15,16]. On the other hand, one thing that should be noticed is that
70 MXenes have trouble in the application process by itself, as they are easy to be oxidized in the air, such as the
71 formation of TiO_2 ; another challenge is that the MXenes are easy to restack. Both of the challenges can be
72 overcome when using MXene as a nanofiller and directly dispersed into polymeric resin, as 1) the oxidization
73 of MXene will be inhibited as they are covered by polymeric matrix, and 2) the high viscosity of polymeric
74 resin will mitigate dispersed MXene from restacking. Therefore, the application of MXene into polymeric
75 coating shows great potential. Researchers agreed that, among the numerous MXene materials, $Ti_3C_2T_x$ is the
76 most widely used due to its excellent chemical and mechanical properties [17,18]. Compared with other
77 traditional two-dimensional materials, the active surface groups of $Ti_3C_2T_x$, such as F, O, OH, and Cl, make it
78 possible to well interacting with the organic polymers through covalent bonds, electrostatic interactions, van
79 der Waals forces or hydrogen bonds [19]. Therefore, MXene-based nanocomposite, as new-generation coatings,
80 are designed to provide high-performance, and multi-functional protective films that have excellent properties
81 in corrosion-resistance, mechanical properties, antimicrobial, durability, and etc.

82 Along the research going with MXene, the application of MXene in areas such as water filtration,
83 electromagnetic shielding, sensors, energy storage and catalyst, etc have been attracted ever-increasing
84 attention and has also been extensively studied [20–24]. For example, Chen et al. summarized the current
85 situation of MXene/polymer nanocomposites and reviewed the applications of MXene/polymer composites in
86 energy, sensing, electromagnetic shielding and biomedical fields [25]. However, the research on MXene-based
87 polymer composites is still in the early stages, and diverse performances need to be explored further [1]. Keep

88 exploring the development of MXene nanocomposite will undoubtedly discover and unlock the full potential
89 of MXene-based composite coatings.

90 In recent years, several studies have been performed to investigate the utilization of MXene or MXene-
91 based composites as coatings to provide protection to infrastructures. Qiang et al. investigated the inclusion of
92 graphene oxide (GO)/ $Ti_3C_2T_x$ nanosheets into epoxy coatings, and the results indicated that GO- $Ti_3C_2T_x$
93 nanosheets acted as effective nanofillers, providing a stronger physical barrier against corrosion for steel in 3.5
94 wt.% NaCl solution at different pressures. The impedance value for GO- $Ti_3C_2T_x$ /EP remained at $1.84 \times 10^8 \Omega$
95 cm^2 after 8 days of immersion at 5 MPa, which was two orders of magnitude higher than that of pure epoxy
96 (EP). This study highlighted the enhanced anti-corrosion performance of GO-functionalized MXene coatings;
97 however, a reduction in corrosion resistance was observed after 8 days of exposure [26]. Cao et al. developed
98 a $Ti_3C_2T_x$ /silane composite coating using a solution-based process and examined its corrosion resistance
99 performance and mechanism. The $Ti_3C_2T_x$ nanosheets effectively minimized micro-voids and extended the
100 diffusion pathway for corrosive media. After 30 days of immersion in a 3.5 wt% NaCl solution, the
101 $Ti_3C_2T_x$ /silane composite coating maintained excellent anti-corrosion performance with a high protection
102 efficiency of 98.55%. This study demonstrated the remarkable improvement in corrosion resistance provided
103 by $Ti_3C_2T_x$ MXene in silane composite coatings on copper [27]. Moreover, Lou et al. also employed
104 electrochemical methods to evaluate the anticorrosion properties of nanocomposite coatings on copper
105 substrates. Their findings revealed that the polyurethane sample containing $Ti_3C_2T_x$ MXene and functionalized
106 carbon nanotubes exhibited a significant reduction in corrosion rate. This study underscores the improved anti-
107 corrosion properties of polyurethane nanocomposites through the incorporation of $Ti_3C_2T_x$ MXene and
108 functionalized carbon nanotubes [28].

109 While previous studies have extensively explored the integration of various nanomaterials into epoxy
110 matrices, there remains a notable gap in research that not only expands on the conventional understanding of
111 polymer reinforcement but also introduces new dimensions related to microstructural stability and anti-aging
112 properties under accelerated aging conditions. Furthermore, there appears to be a lack of comprehensive
113 evaluation of nanofillers, particularly concerning particle size, dispersion, and their impact on the viscosity and
114 microstructure of the composite. Additionally, most existing research has predominantly focused on a single
115 MXene nanoparticle fabrication method. In contrast, our study delves into comparing two distinct MXene

116 synthesis methods and examines their impact on the performance of the resultant nanocomposites. By
117 addressing these gaps, our research offers a nuanced understanding of how different synthesis conditions affect
118 the dispersion quality, structural integrity, and functional performance of MXene within an epoxy matrix, thus
119 providing significant insights into advanced materials engineering.

120 In this manuscript, the presented research aims to provide novel insights into the fabrication of high-
121 performance MXene-reinforced coatings, in order to enhance the corrosion and durability of metal substrates.
122 The MXene-reinforced epoxy coatings were evaluated by a comprehensive experimental study, which
123 incorporated MXene nanoflakes into a solvent-free nanocomposite coating for corrosion protection on metallic
124 surfaces. Two methods were used to fabricate MXene nanoparticles, and the products were labeled as 25C and
125 80C; additionally, both of them were employed as nano-fillers to fabricate epoxy-based nanocomposites. In
126 order to find out the influence of MXene nanofiller concentration on the performance of the coating,
127 nanocomposites were fabricated with varied concentrations of MXene (0.1, 0.5, 1.0, and 2.0 wt.%). Both types
128 of MXene nanoparticles were homogeneously dispersed in the epoxy matrix, and the protective capabilities of
129 developed MXene-epoxy composite coatings were investigated. This study employed micro-computed
130 tomography (Micro-CT) to investigate the internal structure of MXene-epoxy nanocomposite, particularly
131 evaluating void reduction by MXene. The corrosion resistance of MXene-based coatings was assessed using
132 the Electrochemical Impedance Spectroscopy (EIS) test, and the durability was determined by employing the
133 ASTM B117 salt spray accelerated weathering test. Additionally, Electron Spin Resonance (ESR) was utilized
134 to track the production of free radicals and assess the anti-aging capabilities of the MXene-reinforced epoxy.

135 The obtained findings from this study, focusing on the development and analysis of MXene-enriched
136 nanocomposites, marks a significant advancement in the field of materials science, particularly in the domain
137 of polymer composites. The necessity of this work stems from the urgent need for materials that exhibit
138 enhanced durability, structural integrity, and aging resistance under various environmental conditions. By
139 integrating MXene into epoxy composites, the study not only addresses the prevalent issue of rapid degradation
140 in traditional polymeric materials but also introduces a novel approach to enhancing protection properties. The
141 investigation into the microstructural enhancements, anti-aging capabilities, and superior electrochemical
142 behavior of MXene-epoxy composites underlines the potential of these materials to industries such as aerospace,
143 automotive, and civil engineering. The groundbreaking insights provided by this research could lead to the

144 development of more resilient materials capable of withstanding harsh environmental challenges, thereby
145 extending their lifespan and reducing maintenance costs.

146 **2. Materials and experimental design**

147 **2.1. Synthesis of $Ti_3C_2T_x$ MXene**

148 ***Preparation of MAX powder:***

149 The MAX phases Ti_3AlC_2 particles were synthesized by ball milling TiC , Ti , and Al powders for 2 hours.
150 The resulting powders were then pressed into a pellet and sintered at 1350 °C for 4 hours under argon flow. The
151 collected pellets were then milled back into powder and sieved through a 160-mesh sieve. The MAX powder
152 was then collected and stored in a small glass tube.

153 ***Preparation of MXene nanoparticles:***

154 *1) For 25C MXene:* The minimally intensive layer delamination (MILD) method was used at a room
155 temperature without any application of heat. Fluoride-based salt etchants were used to prepare in-situ HF. The
156 etchant was prepared by adding 1.6 g of LiF to 15 mL of HCl and 5 mL of DI water, and then left under
157 continuous stirring for 5 min. A total of 1.0 g of Ti_3AlC_2 powder was gradually added to the etchant above, and
158 the reaction was allowed to run for 48 hours at room temperature of 25°C.

159 *2) For 80C MXene:* Hydrothermal approach was used to prepare 80C MXene nanoparticles. Initially, the
160 etchants were weighed out for LiF , HCl , and MAX using the same stoichiometric ratio for in-situ HF and were
161 stirred on a stir plate using a Teflon bottle at room temperature for 1 hour. After that, the mixed solution was
162 transferred into an autoclave and then placed in a tube furnace to a final heat up at 80°C for 48 hours.

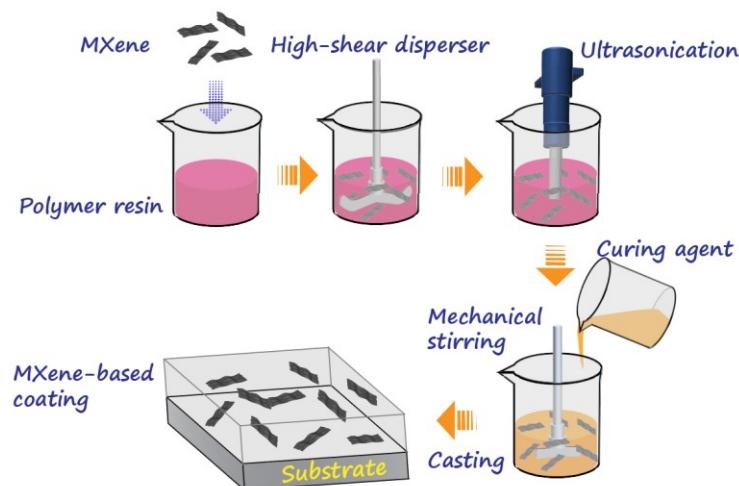
163 *3) Final washing step for both 25C and 80C:* Once the etching steps were completed after 48 hours, the
164 washing steps were performed. During the washing process, the acidic mixtures were washed with DI water via
165 centrifugation (5 min per 6000 rpm) for multiple cycles. The washing process continued until a pH of 4-5 was
166 achieved.

167 **2.2. Fabrication of MXene-epoxy nanocomposites**

168 In this study, EPON 828 resin (Hexion Inc., USA), a bisphenol resin, was used as a liquid epoxy resin to
169 fabricate the polymer to incorporate with MXene nanoparticles. EPIKURE Curing Agent 3175 (Hexion Inc.,
170 USA) was applied as the cross-linking agent to harden the epoxy resin. The process of fabricating

171 nanocomposites can be seen in Fig. 2. The MXene particles were dispersed into the epoxy resin using a
172 combination of a high-shear disk mixer and ultrasonic dispersion. The MXene epoxy resin slurry was subjected
173 to high-shear mixing at a controlled speed of 4000 rpm for 30 minutes. Subsequently, the mixture was sonicated
174 with a Misonix S1805 ultrasonic system, following a 15-second "on/off" cycle at 100% amplitude for an overall
175 duration of 30 minutes. To prevent overheating during the dispersion process, a water bath was employed. It is
176 important to highlight that the MXene nanoparticles were dispersed into the epoxy resin immediately after
177 fabrication. This method effectively mitigates the oxidation of MXene by ensuring that the nanoparticles are
178 completely enveloped by the epoxy resin following the dispersion process. This encapsulation isolates the
179 MXene from air exposure, significantly minimizing oxidation and preserving the integrity of the nanoparticles.

180 The curing agent was added to the mixture after dispersion, with a 1:1 mole ratio to the epoxy resin, and
181 the mixing was carried out mechanically for 10 minutes at a speed of 600 rpm. The experimental investigation
182 in the study involved the MXene-epoxy nanocomposites containing different weight contents of MXene,
183 specifically 0.1, 0.5, 1.0, and 2.0 wt.% for both 25C and 80C MXenes. All the sample groups are denoted by
184 the type and quantity of nanoparticles; for instance, 0.1%25C-E and 0.1%80C-E denote composites reinforced
185 with 0.1% by weight of 25C and 80C MXene, respectively. Furthermore, a pure epoxy sample was utilized as
186 a reference.



187

188 Figure 2. Schematic of the fabrication process of MXene-polymer nanocomposite.

189 **2.3. Characterization of MXene nanoparticle and MXene-epoxy composite**

190 Powder X-Ray Diffraction (XRD) patterns were captured by a Bruker AXS' D8 Discover multifunctional
191 X-Ray Diffractometer, and the results were utilized to examine the structure of the MXenes. The particle size
192 distribution MXenes was examined using the dynamic light scattering (DLS) method, which was carried out
193 using a Nicomp 380 submicron particle sizer. The micrographs that were used in this study were produced with
194 a field emission scanning electron microscope (FE-SEM), which a JSM-7600F Schottky was used. In addition,
195 high-resolution analytical TEM equipment, JEM-2100, was employed to analyze the morphology and structures
196 of the MXene nanoparticles.

197 **2.4. Microstructure of nanocomposites using Micro-CT**

198 Micro-CT, or micro-computed tomography, is a nondestructive imaging technology used to examine and
199 analyze the internal structure of composite materials with a high degree of resolution. A micro-CT scan was
200 used to assess the defect/voids in the MXene nanocomposite and to examine the interactions between MXene
201 nanofiller and polymeric resin as well as the mechanisms of nanoparticle reinforcement. The investigation was
202 conducted based on the void content and size of the voids; and the acquired data were then discussed
203 together with the coating performance.

204 **2.5. Electrochemical impedance spectroscopy test**

205 The protection properties of the developed MXene nanocomposite coatings were investigated by
206 Electrochemical impedance spectroscopy (EIS) test. The barrier properties of coating film and degradation of
207 coated system were assessed by interpreting the EIS data via Bode and Nyquist plots. The EIS measurement
208 was performed by a three-electrode cell, where the steel substrate (Q-panel) acts as the working electrode; and
209 a platinum mesh and a saturated calomel electrode were employed as the counter electrode and reference
210 electrode, correspondingly.

211 **2.6. Accelerated durability test**

212 Salt spray tests were used as an accelerated durability test to assess the developed coatings' corrosion
213 resistance performance over a longer period of time. The ASTM B117 salt spray test is a widely accepted
214 technique for assessing the corrosion resistance of materials or coatings, by subjecting the material or coating
215 to a NaCl salt spray solution. This evaluation allows for the determination of a material's or coating's ability to

216 withstand corrosive environments. Consequently, the coating's long-term durability in a corrosive environment
217 was assessed based on the EIS and Electron spin resonance (ESR) data gathered after varying hours of exposure.

218 **2.7. Electron spin resonance (ESR) test**

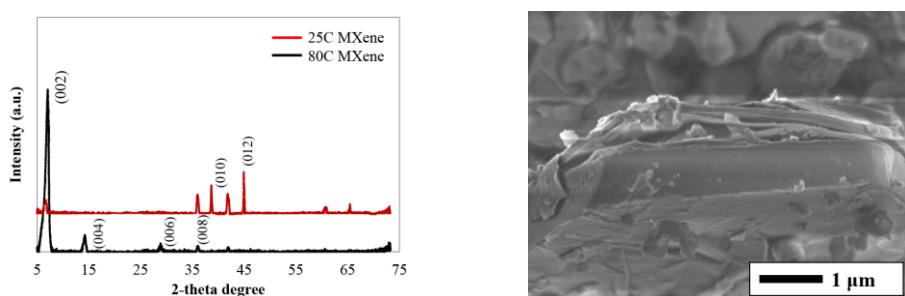
219 The capacity of MXene to delay the epoxy's aging progression was investigated using Electron Spin
220 Resonance (ESR) analysis, performed on both neat epoxy and MXene-epoxy nanocomposites post exposure to
221 salt spray. The ESR spectra were collected at ambient conditions using an ESR spectrometer, and this analysis
222 was essential for identifying the formation of unpaired electrons, which were generated during the aging process.

223 **3. Result and discussion**

224 **3.1. XRD patterns and morphologies of fabricated MXene**

225 The XRD (X-ray diffraction) data presented in Fig. 3(a) shows two distinct patterns for MXene synthesized
226 at 25C (red line) and 80C (black line). The produced 80C MXene exhibits a prominent (002) peak at 7.087°,
227 indicative of a high-purity MXene phase, with additional peaks at 14.311° and 28.851° corresponding to (004)
228 and Li_3AlF_6 phases, respectively. This suggests a successful etching process, corroborated by the minimal
229 presence of Al and TiC impurities, leading to a conclusion that the etching of the MAX phase is more complete
230 at this higher temperature. The presence of -OH groups as the surface-terminating groups further confirms the
231 cleanliness of the MXene surface region [29].

232 In contrast, the 25C MXene displays a much weaker (002) peak, with discernible peaks at 35.985° and
233 44.96° suggesting the presence of TiC impurities, and peaks at 38.715° and 41.835° indicative of Ti_3AlC_2 , a
234 constituent of the original MAX phase [29,30]. This implies incomplete etching at the lower temperature,
235 resulting in a less pure MXene with residual MAX layer components. The 80°C sample, in agreement with
236 SEM results, shows the highest quality MXene, with peaks that confirm the formation of Li_3AlF_6 during the
237 experiment, likely contributing to the debris observed in SEM image (Fig. 3 (b)).



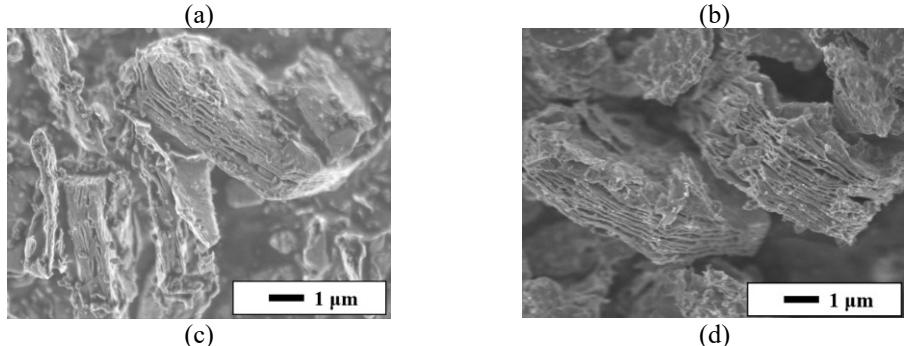


Figure 3. (a) XRD results of pure MXene samples, (b) MAX residuals found in 25C MXene, (c) exfoliated MXene sheets in 25C sample, and (d) exfoliated MXene sheets in 80C sample.

238
 239 The SEM images in Fig. 3(c) and (d) displayed the exfoliated MXene sheets found in 25C and 80C MXene
 240 specimens. As demonstrated in the images, the etching procedure resulted in the removal of the Al layer from
 241 the MAX phase, and formed stacked MXene sheets. Notably, both 25C and 80C MXene flakes exhibited
 242 lamellar structures with lateral dimensions of approximately 5 μm . Both 25C and 80C appeared as evenly
 243 formed 2D nanostructured materials with a smooth surface, which offers a large surface-to-volume ratio;
 244 materials with these unique morphologies can have novel chemical or physical properties, particularly when
 245 these structures fall within the nanoscale scale [31]. On the other hand, the 25C MXene reveals a higher
 246 concentration of MAX phase particle residue than 80C MXene nanoparticles. As Fig. 3(b) highlights, the
 247 distinctive architecture of residual MAX phase particles was more evident in the 25C sample, hinting at a lower
 248 purity level than that of the 80C MXene; furthermore, this observation aligned with the conclusions drawn from
 249 the XRD results.

250 **3.2. Particle size dispersion by DLS and imaging techniques**

251 The dispersion characteristics of the MXene were conducted using Dynamic Light Scattering (DLS) by
 252 identifying the size distribution of the nanoparticles, and the results are presented in Fig. 4(a). The dispersion
 253 level of nanoparticles is a critical factor as it significantly influences the properties of the developed
 254 nanocomposite. As presented in the results of DLS particle size analysis, both of the 25C and 80C MXene have
 255 a size range between 10^2 to 10^5 nm, and most of the particles around 10^3 to 10^4 , which confirmed the
 256 observations in FESEM images (Fig. 4(b) and (c)).

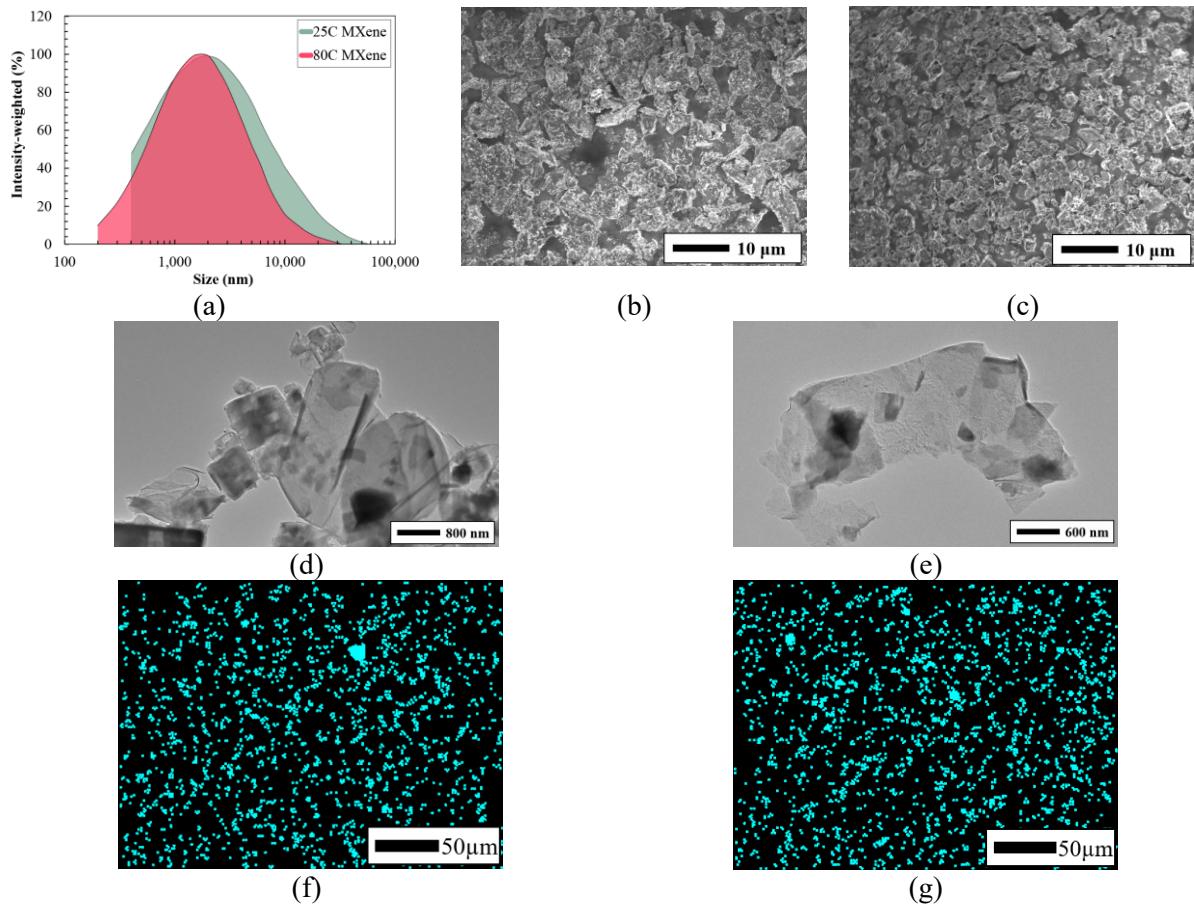


Figure 4. (a) Particle size distribution of 25C and 80C MXene. SEM images of (b) 25C MXene particles and (b) 80C particles. TEM images of (d) 25C and (e) 80C MXene nanosheets after ultrasonication. EDS images of a fracture surface of epoxy reinforced by (f) 25C and (g) 80C MXene.

257

258 However, it is worth noting that the 80C MXene nanosheets were generally smaller in size than the 25C
 259 MXene, and the mean diameter of the 80C is around 2,600 nm while it is 4,000 nm for 25C. Additionally, the
 260 maximum diameter of 25C reached 40,000 nm while the maximum diameter of 80C is 30,000 nm. This finding
 261 can be attributed to the higher amount of MAX phase residuals in 25C MXenes which led to a higher tendency
 262 to form large agglomerates. Additionally, the smallest size of both 25C and 80C showed a good agreement, as
 263 the minimal particles in 80C reached 200 nm, while the minimal particles in 25C was 400 nm. Field Emission
 264 Scanning Electron Microscopy (FESEM) findings reinforced the results from Dynamic Light Scattering (DLS)
 265 particle size distribution. Fig. 4(c) shows the 80C sample having smaller particles and a higher level of
 266 homogeneity compared to the 25C sample. Generally, the smaller and well-distributed nanoparticles improve
 267 dispersion and enhance the properties of composites. However, poor nanoparticle dispersion can lead to
 268 agglomerations, impacting the material's overall performance.

269 The MXene particles that were subjected to ultrasonication were also investigated by TEM, as presented
270 in Fig. 4(d) and (e). The observations revealed that ultrasonication successfully separated stacked MXene
271 nanosheets into ultrathin flakes with large surface areas, enhancing dispersion and reducing agglomeration risks.
272 The 80C sample exhibited better dispersion with fewer residuals and thinner layers than the 25C sample, which
273 was likely due to fewer MAX phase particle residuals as indicated by X-ray Diffraction (XRD) and SEM
274 analyses (Fig. 3).

275 The Energy-dispersive X-ray Spectroscopy (EDS) images in Fig. 4(f) and (g) presented for the 25C and
276 80C MXene samples in epoxy show distinct differences in particle distribution, which align with previous
277 observations from XRD and SEM analyses. The first image, representing the EDS of 25C MXene in epoxy,
278 reveals a less uniform distribution with a tendency towards larger particle agglomerations. This is consistent
279 with the SEM images and XRD data that indicated a lower crystallinity and purity for the 25C MXene. The
280 presence of larger agglomerates could be indicative of incomplete etching, as also suggested by the XRD
281 analysis, which revealed the presence of TiC impurities and other residual phases from the MAX phase. In
282 contrast, in the EDS image of 80C MXene in epoxy, where the particles are more finely dispersed throughout
283 the matrix. This finer dispersion supports the XRD findings of a more crystalline structure and higher purity for
284 the MXene synthesized at 80°C. A uniform dispersion is often correlated with smaller particle sizes and a more
285 complete etching process, which removes impurities and results in a purer MXene product, as the XRD peaks
286 suggested.

287 The results of this study elucidate the contrasting dispersion behaviors and subsequent impacts of two
288 distinct types of MXene nanoparticles, labeled as 25C and 80C, which are synthesized under different
289 conditions to yield varying particle morphologies and sizes. The 25C MXene, synthesized using a less intensive
290 processing method, exhibits a coarser morphology with larger and less uniform particles. This morphology
291 leads to heterogeneous dispersion within the epoxy matrix, resulting in areas where the matrix is not uniformly
292 reinforced. Such inconsistencies can adversely affect the mechanical properties of the composite, potentially
293 increase its porosity, and lead to agglomeration. Additionally, the presence of TiC impurities in 25C contributes
294 to these dispersion challenges, further complicating the integration of these nanoparticles into the epoxy matrix.
295 On the contrast, 80C MXene nanoparticles are synthesized under conditions that foster a finer particle size and
296 more uniform morphology. This refined synthesis results in a more even dispersion throughout the epoxy matrix,

297 which is crucial for achieving consistent material behavior across the composite. The uniform dispersion of 80C
298 MXene will improve the interaction between the nanoparticles and the matrix, and the finer particles of 80C
299 contribute to a denser packing within the matrix, significantly reducing porosity and enhancing the barrier
300 properties of the coating.

301 The comparison between these two images highlights the significant effect that synthesis conditions have
302 on the quality and distribution of MXene particles within an epoxy matrix. The 80C MXene's finer dispersion
303 is likely to contribute to a more consistent reinforcement in the composite, potentially leading to better
304 properties and a more reliable performance in practical applications.

305 ***3.3. Influence of the MXene on the microstructure of the nanocomposites***

306 Micro-computed tomography (Micro-CT) is a non-destructive detection method, that provides detailed
307 information into a composite material's internal structure by using cross-sectional X-ray images; the obtained
308 images can be used to identify voids or flaws within the composite. Voids are major defects in nanocomposites
309 and are commonly found in the polymeric composite matrix. These micro-pores form during the mixing and
310 application process, where the viscous polymeric resin restricts the air release process [32]. Consequently, these
311 produced voids could be large enough to degrade the performance of coating, for example, reduce the barrier
312 properties and accelerate damage development. The interaction between the nanofiller and polymer can lead to
313 either a reduction or an increase in voids; therefore, it is crucial to assess the impact of incorporating MXene
314 on the microstructure of the nanocomposite [33–35].

315 As presented in Fig. 5 and 6, the investigations through Micro-CT scanning on pure epoxy and MXene-
316 epoxy nanocomposites demonstrated intriguing findings regarding the effect of MXene concentration on the
317 voids within the composites. In Fig. 5, the pure epoxy sample revealed nonporous microstructures with low
318 void content; however, these voids were notably large in size. This large void formation in epoxy networks
319 could be attributed to the reduced air release due to the epoxy's viscosity during mixing. Therefore, the large
320 voids can be detrimental to the integrity of the composite, as they may reduce the barrier properties, initiate
321 stress concentrations, and potentially lead to material failure for both coatings and substrates when exposed to
322 corrosive conditions.

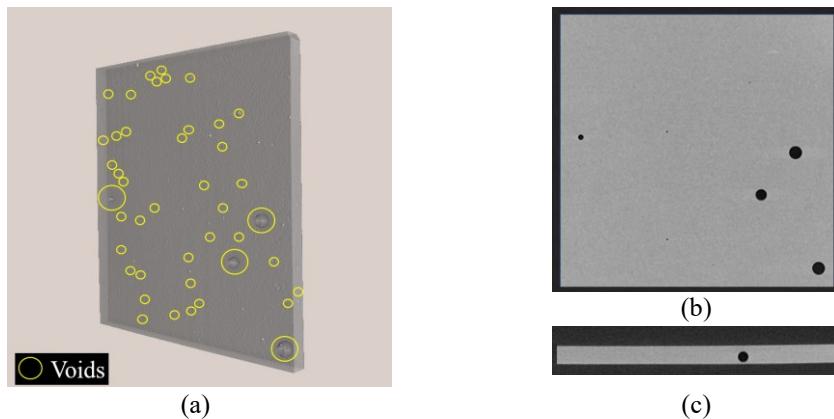


Figure 5. Micro-CT images of neat epoxy (a) transparent 3-D image with areas containing voids highlighted, (b) (c) cross-sectional images.

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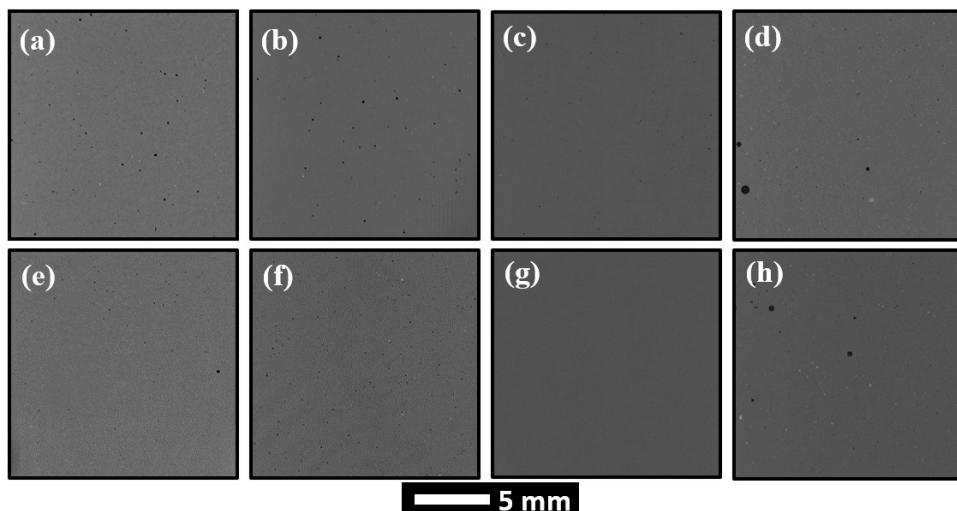


Figure 6. Cross-sectional images of MXene epoxy obtained by Micro-CT, (a) to (d) samples with 25C MXene at 0.1, 0.5, 1.0 and 2.0 wt.%. (e) to (h) samples with 80C MXene at 0.1, 0.5, 1.0 and 2.0 wt.%.

324

325

The incorporation of MXene into the epoxy matrix was found to significantly diminish both the size and numbers of the voids in the MXene-epoxy nanocomposite, implying a beneficial interaction between the MXene and the epoxy matrix. The void reduction for samples confirms the incorporation of both 25C and 80C MXene, improving microstructural integrity by filling in the voids and creating a more continuous, uniform structure, especially for the sample with 0.1% to 1.0% of MXene flakes. However, an interesting trend is observed at higher MXene concentrations, as the samples with 2.0% MXene showed an increase in both void size and amount, which is likely attributable to the elevated viscosity and agglomeration at higher nanofiller contents. This finding emphasizes that the proper amount of MXene can enhance material properties by reducing defects; excessive addition of nanoparticles can lead to the opposite effect.

334 The results also indicated a slight superiority in void mitigation for 80C MXene-epoxy samples compared
335 to the ones with 25C MXene. This observation showed good agreement with the previous findings that 80C
336 MXene is more exfoliated, which could influence the dispersion of MXene in the epoxy matrix, thus affecting
337 the void formation. Overall, these findings contribute significantly to the understanding of defect formation in
338 MXene-epoxy nanocomposites, and the results are essential for the optimization of nanocomposite performance
339 through defect control.

340 **3.4. Corrosion resistance and durability of the MXene-epoxy nanocomposites**

341 As one of the most common and effective methods to evaluate the coating corrosion protection ability,
342 Electrochemical Impedance Spectroscopy (EIS) measurement was employed to identify the barrier properties
343 of the developed MXene-epoxy coatings before and after exposure to a corrosive environment. In Fig. 7, the
344 impedance and phase angle plots for the steel substrates coated with MXene-epoxy coatings at their initial stage
345 are illustrated, while the results after 250 and 500 hours of salt spray exposure are presented in Fig. 8. The
346 impedance modulus at the low-frequency region (usually 0.01 Hz) in the Bode plots can be used to evaluate the
347 corrosion barrier properties of the coatings [36,37]. Furthermore, the phase angle is a vital parameter for
348 assessing the integrity of corrosion protection coatings. Ideally, the phase angles for an intact coating would
349 approach 90 degrees over a wide frequency range, and a reduction in the phase angle value along with an
350 increase in the corresponding frequency suggests the development of coating delamination [38,39].

351 Initially, although the impedance value $\log|Z|$ at 0.01 Hz was high (above 10), the data collected suggested
352 that the neat epoxy coating was inadequate for providing effective corrosion protection, as evidenced by the
353 noticeable bend in the phase angle curve at the low-frequency range. On the other hand, both 25C and 80C
354 MXene at concentrations of 0.1, 0.5, 1.0, and 2.0 weight percent exhibited high impedance values at the 0.01
355 Hz point, with $\log|Z|$ values exceeding 10. Furthermore, the impedance curves of all the MXene-epoxy coatings
356 were linear, with phase angle values closely approaching 90 degrees across all tested frequencies. This
357 observation indicates that all the MXene-epoxy coating films acted as intact protective layers for the underlying
358 substrates, regardless of the type and concentration of MXene particles. In summary, the obtained results
359 demonstrate that improved corrosion resistance was offered by the addition of both 25C and 80C MXene. The
360 enhanced performance is proved by the high impedance values and near 90 degrees of phase angle values in the
361 collect Bode plots.

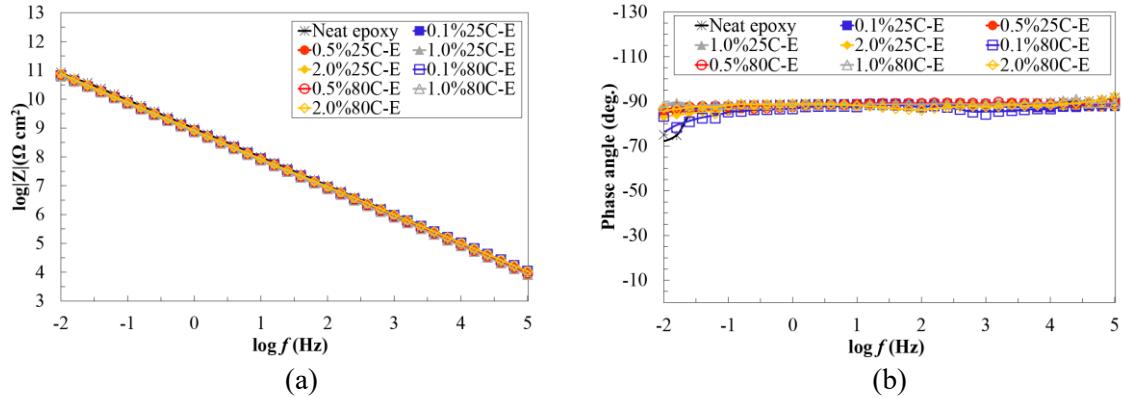


Figure 7. Bode plots of the MXene-epoxy coatings at the initial stage, (a) impedance plot, and (b) phase angle plot.

As presented in Fig. 8, the long-term corrosion resistance of the neat epoxy coating aligned with previously predicted protection performance. The EIS data obtained after exposure to salt spray for 250 and 500 hours suggested a significant decline in the corrosion protection properties of the neat epoxy coating. As exposure duration increased under the accelerated salt spray durability test, the impedance value at the low-frequency region decreased substantially. The initial value of impedance ($\log|Z|$) was above 10, which dropped to 7.9 after 250 hours and further reduced to 5.3 after 500 hours of exposure, indicating a significant reduction in the coating's barrier properties. Consequently, the minimum region in the phase angle diagram shifts toward higher frequencies, suggesting the barrier performance of the pure epoxy coating degrades under accelerated environmental attacks.

On the other hand, the results highlight the reinforcement of MXene addition on the corrosion resistance of epoxy coatings after exposure to salt spray. The impedance values at 0.01 Hz for both 25C and 80C MXene-epoxy samples generally increased with the concentration of MXene, and these values were higher than the neat epoxy coating at both 250 and 500 hours of exposure. More importantly, the MXene nanocomposite coatings containing 1.0 wt.% MXene showcased the greatest improvement, with no apparent changes in impedance and phase angle. Although material degradation was detected in the other sample groups, their corrosion resistance still surpassed that of the unmodified epoxy. This suggested that the incorporation of both types of MXene nanoparticles, especially with 1.0 wt.%, enhances the corrosion protection performance of epoxy coatings over time, as demonstrated by the improved impedance values in the EIS tests.

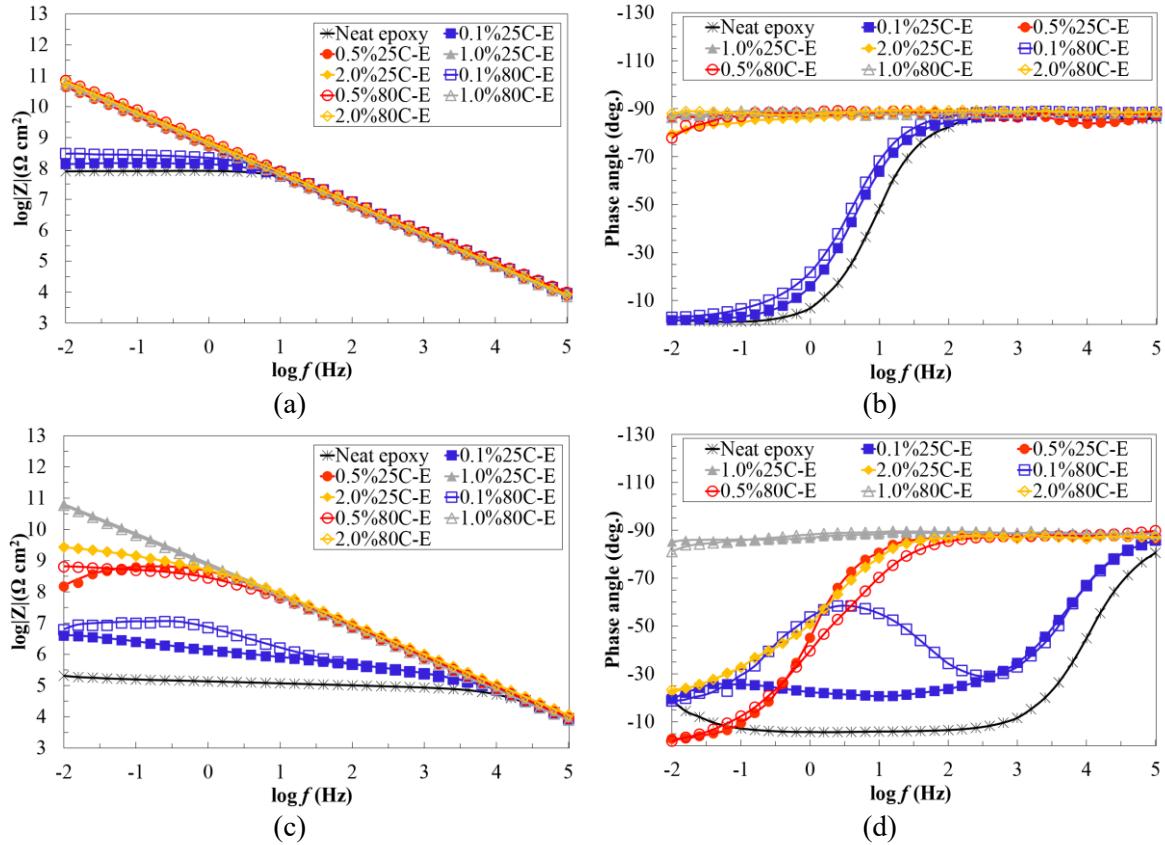


Figure 8. Bode plots of the MXene-epoxy coatings after exposure, (a) impedance plot after 250 hours, (b) phase angle plot after 250 hours, (c) impedance plot after 500 hours, (d) phase angle plot after 500 hours.

381
 382 The effects of MXene concentration on the corrosion resistance of epoxy coatings can be conducted by
 383 considering various mechanisms. The unique two-dimensional structure of MXene nanoparticles contributes to
 384 the development of a complex, tortuous pathway throughout the coating to the substrate. This tortuous pathway
 385 hinders the penetration of corrosive agents through the coating and reduces the rate for them to reach the
 386 substrate. This phenomenon may explain the increased impedance values observed when incorporating MXene
 387 nanoparticles into the epoxy. Additionally, the presence of MXene nanoparticles can enhance the coating by
 388 creating a composite material with fewer and smaller pores or defects, thereby inhibiting the infiltration of
 389 corrosive substances. This improvement might be due to MXene's ability to reduce voids in the composite,
 390 which was observed in Micro-CT test. The improved corrosion resistance associated with increasing MXene
 391 concentrations, up to 1.0% for both 25C and 80C MXene-epoxy specimens, supported this assumption. On the
 392 other hand, when the MXene concentrations increased, such as 2.0%, corrosion resistance appeared to decline
 393 compared to samples containing lower concentrations of MXene (1.0%). This reduction might be attributed to

394 the formation of large-sized agglomerates and voids within the coating, which function as defects and assist the
395 penetration of corrosive media.

396 In summary, both 25C and 80C MXene enhance corrosion resistance compared to the neat epoxy, with the
397 80C MXene-epoxy samples generally exhibiting slightly superior performance. The findings suggested that an
398 optimal MXene concentration exists for maximizing corrosion resistance, potentially attributable to the balance
399 between the advantageous effects of MXene's 2-D structure and reduced voids.

400 ***3.5. Evaluation of corrosion damages over time using EEC modeling***

401 The electrochemical equivalent circuit (EEC) model, comprising a finite set of elementary (resistors,
402 capacitors, and inductors), is generalized to match the collected EIS data, so the corrosion stage of the coating-
403 substrate system can be characterized. The electrical equivalent circuit (EEC) model is a mathematical model
404 that represents the corrosion process of a coated metal as an electrical circuit. In the EEC model, the coated
405 metal is represented by a series combination of an electrical resistor and a capacitor. In order to assess the
406 coating deterioration under increased environmental stress, four types of EEC models are generally incorporated
407 to characterize the corrosion stage. As depicted in Fig. 9, from Model A to Model C, the state of the coating
408 system ranges from undamaged to severely corroded.

409 Model A describes the coatings remaining intact in the initial stage (Stage I), and the EEC model is
410 constructed by solution resistance (R_{sol}), coating resistance (R_c), and constant-phase element of the coating
411 (CPE_{po}). Model B examines the coating with imperfections, in which the electrodes being able to break through
412 the polymer film and contact the metallic surface, therefore triggering corrosion in the substrate. Under this
413 scenario, the corrosion activity is in Stage II; hence, charge transfer resistance (R_{ct}) and constant-phase element
414 of double-charge (CPE_{dl}) should be introduced to model the corrosion behaviors at the occurs at the coating-
415 substrate interface. In the third phase, corrosion products diffuse via coating defects, and this activity begins to
416 predominate the corrosion reaction in the coated system. Therefore, the Warburg diffusion element (W_s)
417 is adopted to the EEC model in order to clarify the diffusion effect, which denotes that the corrosion process
418 has reached Stage III. Model C is employed to represent coating systems that under severe corrosion damage,
419 suggesting that a thin layer of rust has formed in the coating-substrate interface. In order to simulate the layer
420 of corrosion products, constant phase element of diffusion capacitance (CPE_{diff}) and diffusion resistance (R_{diff})
421 are substituted for Warburg diffusion at this stage (Stage IV).

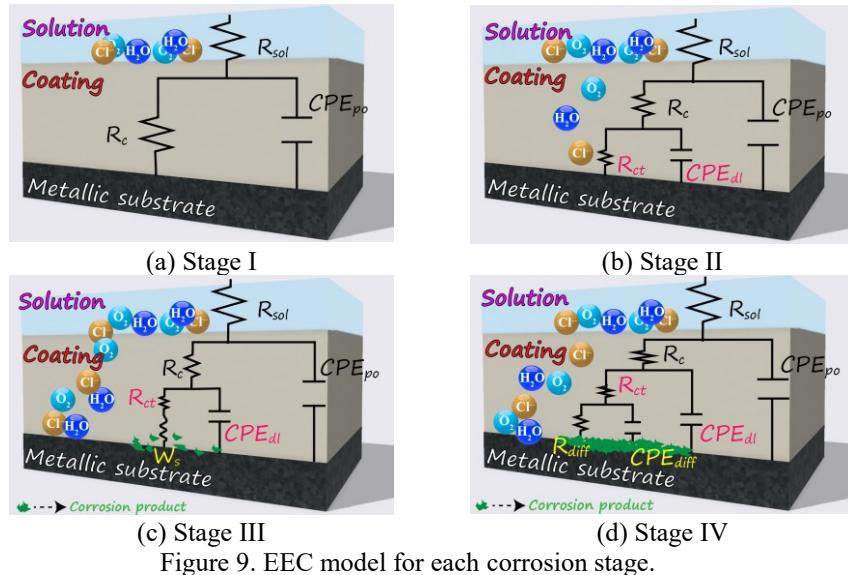


Figure 9. EEC model for each corrosion stage.

422
 423 The following discussion expands on the corrosion stages of each developed MXene nanocomposite
 424 coatings, providing a more detailed analysis of their performance under B117 salt spray exposure, and the
 425 results are summarized in Table 1. The neat epoxy sample exhibited a rapid progression in corrosion stages,
 426 starting at Stage II at the onset and reaching Stage IV after 500 hours of exposure. This finding indicated that
 427 the coating had imperfections, which allowed the electrodes to encounter the metallic surface, leading to
 428 corrosion in the substrate. The progression to Stage IV after 500 hours demonstrates that a thin layer of rust had
 429 formed at the coating-substrate interface, further weakening the protective properties of the coating.

430 Table 1. The corrosion stage of MXene-epoxy coated samples after exposure.

| Label | Corrosion stage determined by EEC model | | |
|------------|---|-----------|-----------|
| | Onset | 250-hr | 500-hr |
| Neat epoxy | Stage II | Stage II | Stage IV |
| 0.1%25C-E | Stage I | Stage III | Stage III |
| 0.5%25C-E | Stage I | Stage I | Stage II |
| 1.0%25C-E | Stage I | Stage I | Stage I |
| 2.0%25C-E | Stage I | Stage II | Stage III |
| 0.1%80C-E | Stage I | Stage III | Stage III |
| 0.5%80C-E | Stage I | Stage I | Stage II |
| 1.0%80C-E | Stage I | Stage I | Stage I |
| 2.0%80C-E | Stage I | Stage II | Stage III |

431
 432 Results from the corrosion stage analysis indicated that both 25C and 80C MXene have shown
 433 extraordinary improvement in corrosion resistance. The samples containing 0.1% of 25C MXene and 80C
 434 MXene, started at Stage I and transitioned to Stage III after 250 hours of exposure. This suggests that the

435 coatings initially remained intact but later developed imperfections, allowing corrosion products to diffuse
436 through the coating defects. The samples remained at Stage III even after 500 hours, indicating that the corrosion
437 process was dominated by the diffusion of corrosion products at this stage.

438 The samples with 0.5% MXene loading (0.5%25C-E and 0.5%80C-E) exhibited improved corrosion
439 resistance compared to the 0.1% loaded samples, as they remained at Stage I after 250 hours of exposure,
440 demonstrating that the coatings were still intact and providing effective protection to the metal substrate. After
441 500 hours, these samples progressed to Stage II, suggesting that some corrosion damage had developed in the
442 coating-substrate interface. The samples with 1.0% MXene loading (1.0%25C-E and 1.0%80C-E) showed the
443 most stable corrosion resistance throughout the test period, remaining at Stage I even after 500 hours of
444 exposure. This highlights the superior performance of these coatings, as they maintained their effective
445 protection properties and prevented corrosion in the metal substrate. On the other hand, the samples with 2.0%
446 MXene loading (2.0%25C-E and 2.0%80C-E) initially exhibited Stage I corrosion resistance; however, they
447 progressed to Stage II after 250 hours of exposure and reached Stage III after 500 hours. The observations from
448 2.0%25C-E and 2.0%80C-E samples indicated that while the higher MXene loading provided initial protection,
449 the voids and imperfections in coatings eventually allowed corrosive media and corrosion products to diffuse
450 through the defects.

451 The observations from the images presented in Fig. 10 are in firm agreement with the EIS and EEC results.
452 The images in Fig. 10 presented the coated samples after 250 and 500 hours of B117 salt spray exposure. In the
453 case of pure epoxy, corrosion products began accumulating from 250 hours of exposure, and after 500 hours,
454 these products penetrated the coating and reached the top surface.

455 Epoxy samples containing 25C and 80C MXene exhibited similar observations, and corrosion reactions
456 were noted in both samples containing 0.1% and 2.0% MXene. However, the corresponding mechanisms are
457 different: for the 0.1% MXene samples, the lower concentration of MXene could not provide strong
458 reinforcement against corrosion. In contrast, for the 2.0% MXene samples, the large amount of MXene created
459 agglomerates and voids, leading to accelerated corrosion damages. Consequently, the corrosion type in 0.1%
460 MXene samples was general corrosion, while in 2.0% samples, it was localized in defective areas. In the case
461 of the 2.0% MXene samples, the EIS results at 250 hours indicated that the coatings exhibited excellent barrier
462 properties. However, some coating damage was observed in areas with large MXene agglomerates; as a result,

463 these agglomerated spots developed into localized corrosion after 500 hours of salt spray exposure. The
 464 interaction between the MXene agglomerates and the epoxy matrix could play a role in this localized coating
 465 degradation. The larger MXene agglomerates might disrupt the uniformity of the epoxy matrix, creating weak
 466 spots that are more susceptible to corrosive attacks. These weak spots could lead to a higher concentration of
 467 stress, allowing the penetration of corrosive species and accelerating the corrosion process.

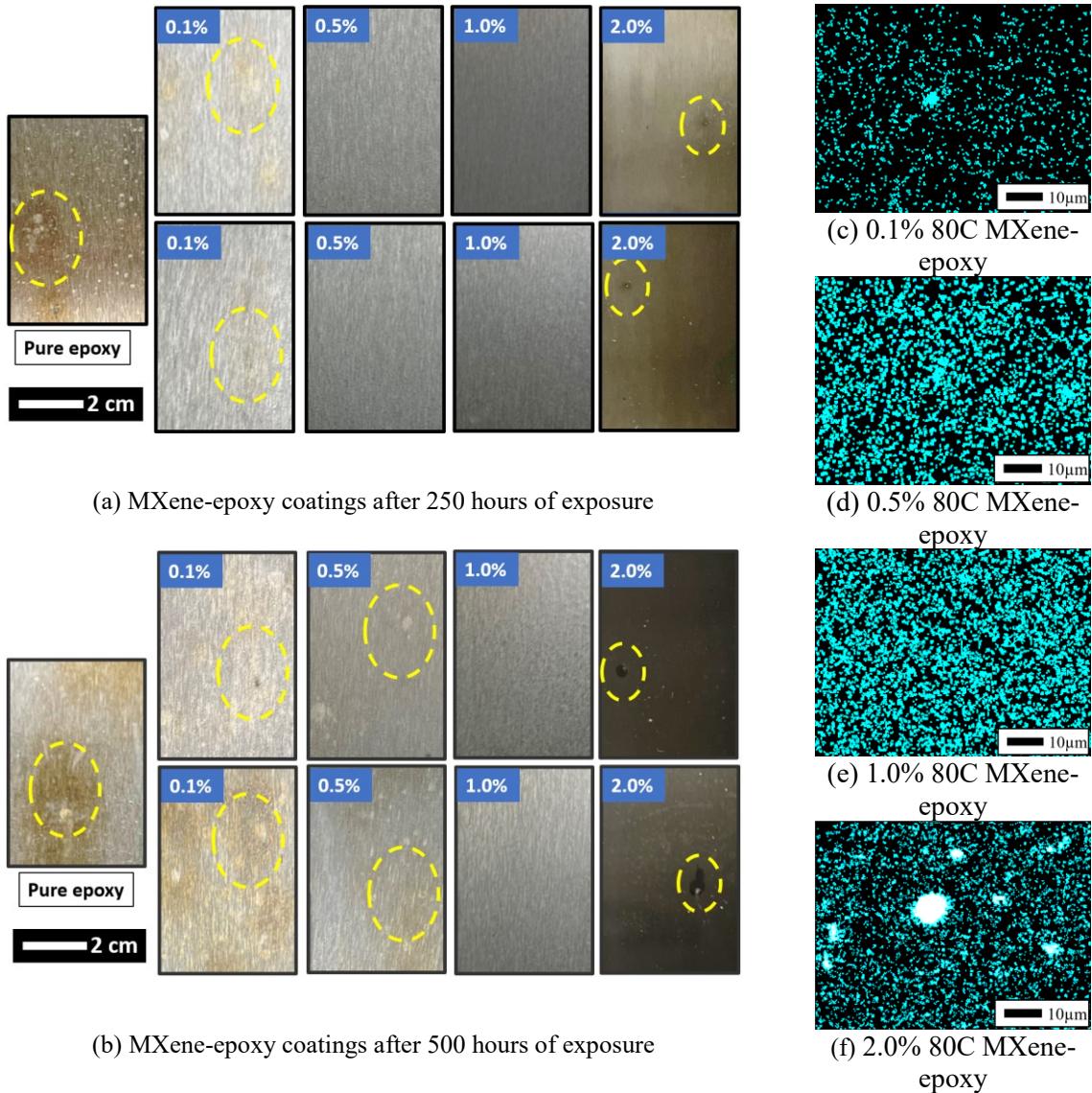


Figure 10. Images of MXene-epoxy coatings after salt spray exposure, (a) 250 hours and (b) 500 hours. EDS images (c) to (f) of cross-sectional surface for MXene-epoxy nanocomposite.

468

469 Meanwhile, the 0.5% MXene samples demonstrated improved corrosion resistance due to the increased
 470 amount of MXene; however, corrosion production and coating delamination were still observed after 500 hours.
 471 In comparison, the 1.0% MXene samples showed the most substantial improvement in corrosion resistance,

472 with no corrosion reactions detected after 500 hours. Thus, the observations from the images are in strong
473 agreement with the EIS results and the corrosion stage in Table 1.

474 To further study the mechanism of corrosion protection behavior, the sequence of Energy-dispersive X-ray
475 Spectroscopy (EDS) images (Fig. 10 (c) to (f)) were used to depicts the cross-sectional-surface of epoxy
476 composites containing 80C MXene at varying concentrations: 0.1%, 0.5%, 1.0%, and 2.0%. An analysis of
477 these images reveals critical insights into the dispersion behavior of MXene within the epoxy matrix and its
478 subsequent impact on the material's anti-corrosive properties. At the 0.1% MXene concentration, the EDS
479 micrographs reveal a sparse distribution of MXene particles within the epoxy matrix. While some degree of
480 MXene presence is evident, there exist extensive regions within the matrix that lack MXene coverage,
481 potentially resulting in insufficient reinforcement against corrosive media. As the concentration increases to
482 0.5% and 1.0%, there is a notable increase in particle density, yet the dispersion remains relatively even. This
483 homogeneity is beneficial for corrosion protection as it allows for a consistent protective layer that can prevent
484 corrosive elements from reaching the underlying material. However, in Fig. 10(f), at the highest concentration
485 of 2.0%, areas of agglomeration were found where MXene particles have clustered together. These
486 agglomerates could potentially create weak points in the composite material, which may negatively impact
487 corrosion resistance. Agglomerates act as barriers to uniform stress distribution and can be sites where corrosion
488 initiates and propagates more easily due to the less protected epoxy regions around them.

489 The increasing density of MXene particles up to 1.0% likely contributes to improved barrier properties of
490 the epoxy composite, as the particles create a tortuous path for corrosive agents, thus enhancing the corrosion
491 protection performance. However, the excessive agglomeration observed at 2.0% could compromise this
492 protective effect due to the formation of non-uniform layers that are less effective at impeding corrosion. In
493 summary, the EDS images show that while increasing the MXene content up to a point can enhance the
494 composite's protective qualities against corrosion, over-saturation leads to agglomeration, which can have the
495 opposite effect. The optimal concentration for corrosion protection likely lies at or below 1.0%, where the
496 benefits of MXene addition are maximized without the drawbacks of particle clustering.

497 ***3.6. Evaluation of coating degradation over time using coating protection index***

498 To simplify the evaluation of coating's performance degradation process in a severely corrosive
499 environment, two types of damage indices were employed; both were modified depending on those

500 implemented by previous studies [40]. As presented in Fig. 11, both type I and II coating damage indices were
501 derived by interpreting the impedance curves in Bode plots.

502 In the impedance plot of a coating that functions as an intact protective barrier, as depicted in Fig. 11, the
503 slope of the impedance curve will retain -1, and the plot will contain solely capacitive areas. Alternatively, if
504 the protective film begins to deteriorate or experience damage, the slope of the impedance curve will drop,
505 especially in low frequency ranges, as seen in Fig. 11.

506 As illustrated in the equation (1), the first damage index (DI_1) was calculated by the change of Z_{mod} values
507 measured at the lowest frequencies, in this case 0.01 Hz, after coating degradation.

$$508 \quad DI_1(\%) = \left(\frac{\log Z_{damaged}}{\log Z_{intact}} \right) * 100 \quad (1)$$

509 where, Z_{intact} and Z_{damage} represent impedance values ($Z_{0.01Hz}$) at the intact and damaged stages, accordingly.

510 The other damage index (DI_2), in equation (2), is derived from the capacitive and resistive areas of the bode
511 plot, which reflects the degradation of coating performance over the whole frequency range that tested:

$$512 \quad DI_2(\%) = \left(\frac{A_1+A_2}{A_1+A_2+A_3} \right) * 100 \quad (2)$$

513 whereby the region under the impedance slope is subdivided into capacitive (A2) and resistive (A1) areas
514 according to the threshold frequency values at -45° phase angle [41,42]. As soon as the electrolyte solution
515 penetrates the coated specimens, coating degradation and corrosion at the coating-substrate interface will
516 be initiated. In this situation, the impedance curve will respond by decreasing capacitive region and expanding
517 the resistive region [43]. Consequently, the impedance curve will include both a resistive region and a capacitive
518 region, and an additional area (A3) could be identified just above the resistive region (A1) for the damaged
519 coating, compared to an undamaged coating. For both damage indices, the greater the index value, the less the
520 coating degrades. Therefore, a sample with an index value of 100 percent, for instance, suggests that the coated
521 film serves almost as an isolating layer to protect the substrate.

522

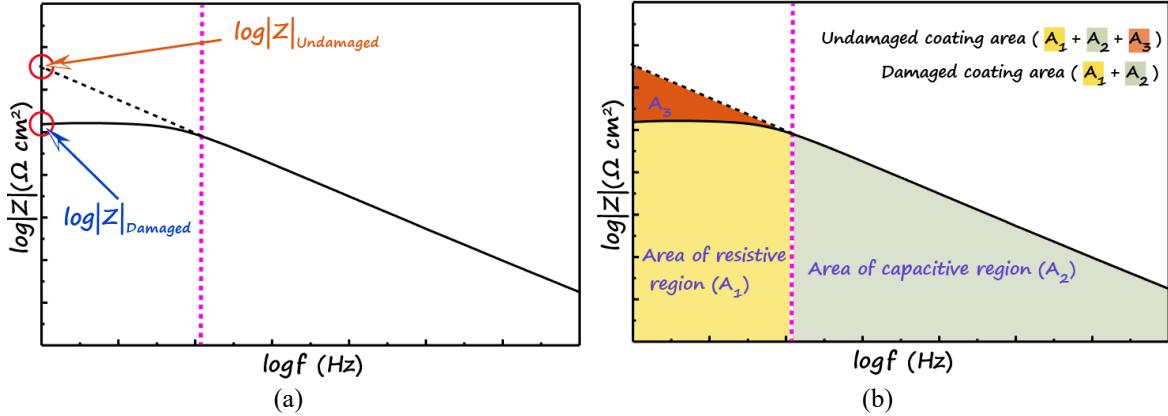
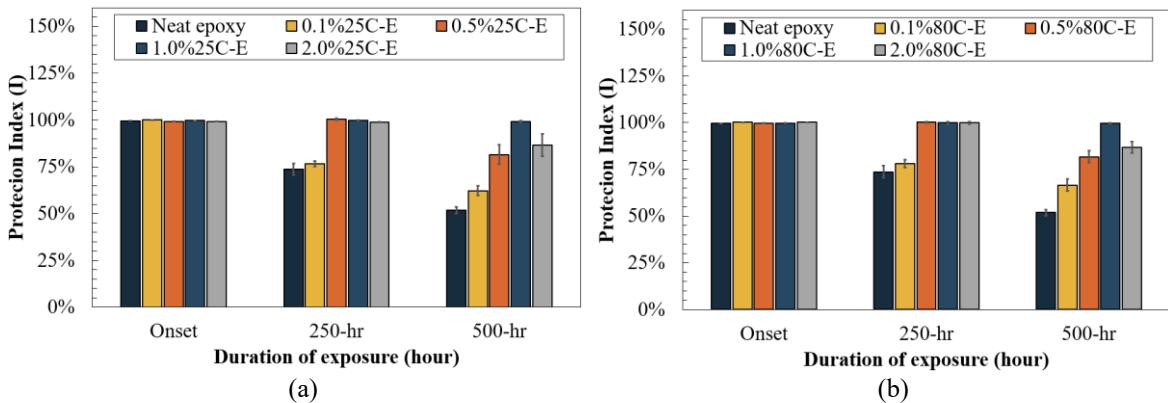


Figure 11. Damage index for the coating degradation assessment: (a)-(b), (c)-(d)

523
 524 The findings from damage index I and II offer valuable insights into the performance and durability of
 525 MXene-epoxy coatings when exposed to a corrosive environment (Fig. 12). Damage index I results demonstrate
 526 that coatings containing higher MXene concentrations (1.0% and 0.5%) exhibit improved durability and
 527 corrosion resistance, even after 500 hours of exposure. Notably, the 1.0%25C-E coating retains 99% of its initial
 528 performance after 500 hours, while the 0.5%25C-E coating maintains 82%. These findings align with the EEC
 529 model and EIS data, which suggest that these coatings possess enhanced barrier properties and heightened
 530 corrosion resistance. The 1.0%80C-E coating, in particular, retains its full initial performance after 500 hours,
 531 illustrating its exceptional resilience in a corrosive environment.



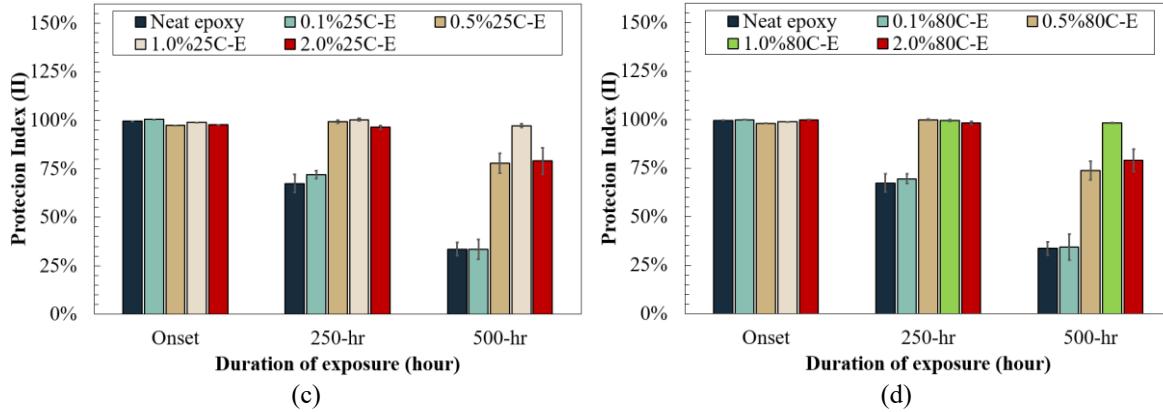


Figure 12. Protection indices of epoxy composites reinforced by (a) 25C and (b) 80C MXene.

532

533 Similar findings were observed in damage index II, and results indicate that coatings with lower MXene
 534 concentrations (0.1%25C-E and 0.1%80C-E) and neat epoxy experience substantial degradation over time.
 535 After 500 hours, the performance of these coatings drops significantly, with neat epoxy retaining only 34% of
 536 its initial performance, 0.1%25C-E at 33%, and 0.1%80C-E at 34%. These observations correspond with the
 537 EIS data, which shows that lower MXene concentrations and neat epoxy coatings exhibit increased corrosion
 538 rates, reduced coating resistance, and progression through various corrosion stages.

539 It is crucial to note that while 2.0% MXene coatings display relatively high durability, their performance
 540 declines after 500 hours due to the presence of large agglomerates within the coating. The 2.0%25C-E and
 541 2.0%80C-E coatings retain above 80% of their initial performance at both damage indices, respectively. This
 542 finding is supported by Micro-CT images and EIS data, which identify localized corrosion in areas with higher
 543 MXene agglomerates.

544 In summary, the insights from damage indices I and II, combined with the EEC model and EIS data,
 545 emphasize the effectiveness of MXene-epoxy coatings in delivering exceptional corrosion resistance and
 546 durability. Optimal MXene concentrations (1.0% and 0.5%) provide the best performance, while coatings
 547 containing lower MXene concentrations or large agglomerates experience a decline in their protective
 548 capabilities over time

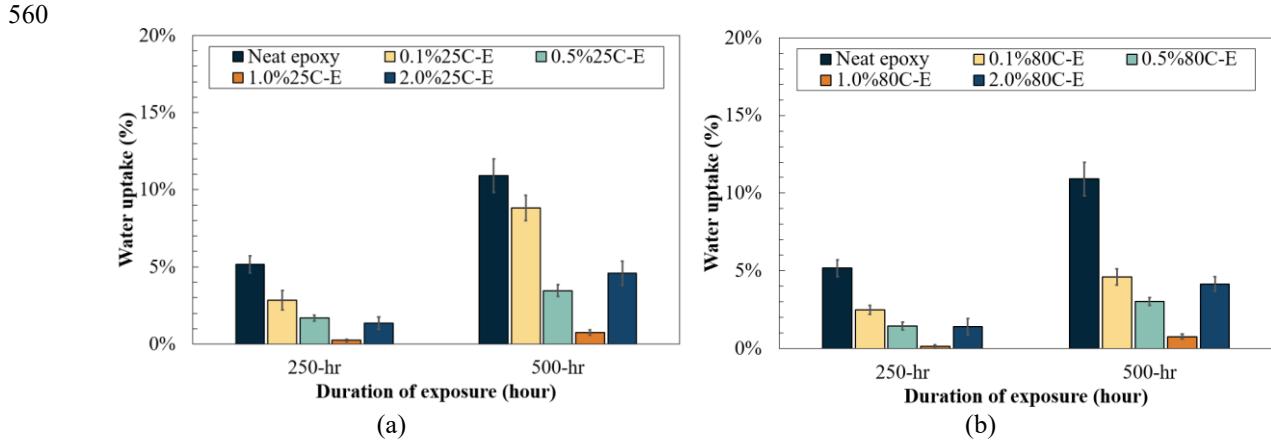
549 **3.7. Evaluation of coating performance over time using water absorption analysis**

550 The penetration of water or other active ionic species through thin polymer coatings is a common cause of
 551 coating failure. Since the capacitance and resistance of a polymeric film change as it absorbs water,
 552 electrochemical impedance spectroscopy (EIS) results provide an estimation of the amount of water uptake in

553 the polymer [44]. Due to the fact that water's dielectric constant is greater than common polymers, it is
 554 anticipated that the absorption of water will raise the capacitance of the coating. Brasher and Kingsbury utilized
 555 the model used to estimate the water absorption of a coating, as presented below:

$$556 \quad \Phi = \frac{K \log\left(\frac{C_t}{C_0}\right)}{\log(\epsilon_w)} \quad (2)$$

557 where Φ is the water content volume percentage, C_t and C_0 are the capacitance of the coating before and
 558 after the test, respectively. Additionally, ϵ_w is the dielectric constant of water at operating temperatures (=78.5
 559 at 25 degrees Celsius and 75 at 35 degrees Celsius).



560
 561 Figure 13. Water uptake of epoxy composites reinforced by (a) 25C and (b) 80C MXene.

562 The water uptake results after 250 and 500 hours of B117 Salt spray exposure provide valuable information
 563 regarding the performance of the MXene-epoxy coatings in resisting water penetration, a common cause of
 564 coating failure. The results of the water uptake are presented in Fig. 13. The data reveals that the MXene-epoxy
 565 coatings (0.1%, 0.5%, 1.0%, and 2.0%) generally exhibit lower water uptake compared to the neat epoxy. This
 566 indicates that the incorporation of MXene into the epoxy matrix improves the resistance of the coatings to water
 567 penetration, which is an essential factor in determining the overall performance and durability of protective
 568 coatings.

569 Among the MXene-epoxy coatings, the 1.0%25C-E and 1.0%80C-E samples display the lowest water
 570 uptake values, suggesting superior resistance to water penetration. This finding correlates well with the
 571 previously discussed damage indices, EEC models, and EIS results, where these coatings exhibited excellent
 572 barrier properties and increased corrosion resistance. The lower water uptake values in the MXene-epoxy
 573 coatings can be attributed to the MXene particles acting as barriers to water penetration, thus enhancing the

574 overall performance of the epoxy matrix. Additionally, the water uptake results can also be correlated with the
575 porosity of the MXene-epoxy composite coatings. The addition of MXene into the epoxy matrix has been
576 observed to reduce the porosity and void sizes within the coatings, as evidenced by the Micro-CT images.
577 However, it should be noted that the coatings with higher MXene concentrations (2.0%25C-E and 2.0%80C-E)
578 display a higher water uptake than the coatings with lower MXene concentrations. This may be due to the
579 presence of large agglomerates in the coatings, as discussed earlier, which negatively impact the protective
580 capabilities of the coatings.

581 In conclusion, the optimal MXene concentrations (1.0% and 0.5%) offer the best performance, whereas the
582 coatings with lower MXene concentrations or large agglomerates show a decline in their protective capabilities
583 over time. The synergistic effects of MXene's barrier properties and the changes in porosity due to its
584 incorporation into the epoxy matrix contribute significantly to the improved performance of MXene-epoxy
585 composite coatings. The barrier properties of MXene are primarily attributed to its high aspect ratio, which
586 forms tortuous pathways for water and other corrosive agents, effectively impeding their penetration through
587 the coating. Simultaneously, the addition of MXene to the epoxy matrix results in a more densely packed
588 composite structure, reducing porosity and void sizes.

589 ***3.8. Anti-aging properties of MXene-epoxy nanocomposites***

590 Epoxy resins usually have excellent mechanical properties, chemical resistance, and adhesion, which
591 makes them widely used in various industries. However, environmental conditions such as heat, light, and
592 oxygen can compromise their stability, promoting the generation of free radicals. These highly reactive species
593 with unpaired electrons can cause a chain reaction leading to material degradation. Over time, these radicals
594 can trigger more chemical reactions, causing further degradation of the polymer network. Therefore, it is crucial
595 to minimize free radical formation and preserve the long-term stability and performance of the epoxy resin, and
596 the quantity of free radicals can be used as a measure for the aging/degradation of the epoxy nanocomposite.

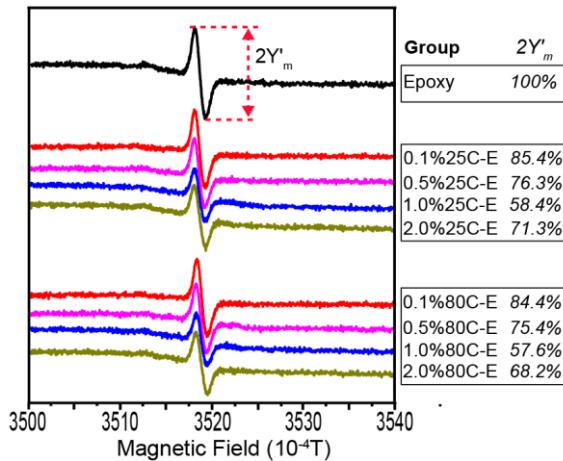
597 The anti-aging capabilities of a coating significantly enhance its corrosion protection in two primary ways;
598 firstly, anti-aging properties in materials, especially those used in coatings, are crucial in reducing the rate of
599 degradation due to environmental factors. These properties help in maintaining the structural integrity of the
600 protective barriers over extended periods [45]. By preventing the breakdown of material structures, anti-aging
601 capabilities ensure that the onset of corrosion, which often follows material weakening or failure, is significantly

602 delayed. Secondly, as coatings age, they can become more brittle [46]; anti-aging properties play a vital role in
 603 preserving the elasticity and mechanical strength of these coatings. This preservation is crucial in preventing
 604 the initiation and propagation of micro-cracking, which can occur due to the generation and diffusion of
 605 corrosion products. Moreover, maintaining the coating's integrity prevents chipping and ensures that the barrier
 606 remains seamless and intact, which is essential for effective long-term corrosion protection.

607 The anti-aging ability introduced by the addition of MXene is assessed by comparison with pure epoxy
 608 samples after exposure. The concentration of free radicals within the system can be estimated via the following
 609 equation:

$$610 \quad I \propto 2Y'_m(\Delta H_{pp})^2 \quad (1)$$

611 where $2Y'_m$ denotes the peak-to-peak amplitude, ΔH_{pp} symbolizes the width between peaks in the sample
 612 spectrum, and I stands for the signal strength [47]. Therefore, the peak-to-peak amplitude ($2Y'_m$) directly
 613 associated with the free radical concentration when the peak to peak width is consistent [47]. Therefore, Fig.
 614 14 presents the collected ESR data for all tested samples after 500 hours of exposure to a corrosive environment,
 615 and the peak in all the tested samples indicates the presence of hydroxyl (O–H) radicals, which are formed by
 616 the aging process of epoxy [48]. In order to simplify comparison, the amplitude to epoxy is assigned as 100%
 617 as a standard, with other samples computed as percentages accordingly.



618
 619 Figure 14. ESR spectra measurement of samples after exposure to corrosive environment.

620 As presented in Fig. 14, the spectrum of all the tested samples showed a simple curve with one central peak
 621 which might be the presence of free radicals that formed due to the aging process of the epoxy [48]. The
 622 collected data indicated neat epoxy demonstrated the highest intensity of these free radicals, indicating its aging
 susceptibility in a corrosive environment. However, the addition of Mxene to the epoxy matrix brought a

623 significant reduction in free radical generation. For both 25C and 80C Mxene-epoxy groups, there was a clear
624 trend of diminishing free radical intensity with increasing Mxene concentration up to 1.0%. This observation
625 aligned well with the concept that nanofillers like Mxene improved microstructural stability, as well as
626 enhanced anti-corrosion properties of the composite material. More specifically, MXene appeared to mitigate
627 the epoxy aging process under corrosive conditions by suppressing radical generation. However, at the highest
628 Mxene concentration (2.0%) an increase in free radical intensity was observed. This finding highlights the
629 previous conclusion in the discussion of Micro-CT results, in which excessive concentrations of MXene
630 nanoparticles potentially lead to defects and agglomerations within the composite. In this case, the structural
631 irregularities could facilitate oxygen diffusion into the coating, triggering a more rapid reaction of carbon
632 radicals with atmospheric oxygen, leading to accelerated degradation [47].

633 The results showed a good agreement with previous microstructural, EIS, and EDS imaging analyses,
634 offering a more holistic picture of the effects of Mxene reinforcement in epoxy-based nanocomposites.
635 Especially with the EDS data (Fig. 10 (c) to (f)), at a 0.1% concentration of MXene, the lower concentration
636 might contribute to a certain degree of aging resistance by interrupting the pathway of UV radiation and other
637 aging agents, the insufficient coverage by MXene particles likely leads to areas within the matrix that remain
638 unprotected. This could allow for oxidative and UV-induced degradation processes to occur more readily in
639 those regions, potentially compromising the composite's overall anti-aging performance. Higher concentrations
640 (0.5 and 1.0%) improve this protective effect by providing more uniform coverage. However, at 2.0% MXene,
641 particle agglomerations might compromise the composite's anti-aging performance by creating localized weak
642 points vulnerable to faster degradation. Thus, an optimal MXene concentration exists that maximizes aging
643 resistance without the negative impact of particle clumping. The comprehensive study underscores the
644 multifaceted improvements when incorporating MXene into the composite, not only in terms of microstructural
645 enhancements and anti-corrosion characteristics but also in aging resistance.

646 **4. Conclusion**

647 In summary, two fabrication methods were used to develop lamellar $Ti_3C_2T_x$ MXene nanoparticles, a 2D
648 graphene-like nanomaterial, labeled as 25C and 80C, and novel MXene-epoxy nanocomposite coatings were
649 successfully synthesized. The application of using MXene as reinforcement was introduced into epoxy resin to
650 fabricate high-performance protection coating with extraordinary robustness. The fabricated MXene and

651 MXene-based nanocomposite coatings were characterized by XRD, SEM, and TEM measurements to
652 investigate structure, morphology, and composition. Both the MXene-reinforced epoxy coating systems were
653 prepared with the varied contents of MXene nanofillers, ranging from 0.1 to 2.0 wt.%, and the major findings
654 were summarized based on experimental results:

- 655 • A comparison between the 25C and 80C variants of MXene highlights that the finer morphology and
656 better dispersion of MXene generated at 80C lead to enhanced performance of the composite material.
657 This further emphasizes the significance of optimizing the production conditions of MXene to
658 maximize its beneficial contributions to the nanocomposite.
- 659 • Both 25C and 80C MXene significantly influenced the microstructure of the epoxy composite,
660 reducing void formation and promoting homogeneity at the microscale. This improved microstructure
661 played a vital role in enhancing the overall performance of MXene-based nanocomposites.
- 662 • The results of this study demonstrated the potential of MXene nanoparticles as an effective
663 reinforcement for polymeric coatings, enhancing their corrosion resistance and durability. The superior
664 performance of the 1.0% MXene-loaded samples (1.0%25C-E and 1.0%80C-E) highlights the benefits
665 of incorporating MXene into epoxy matrices, with the coatings maintaining their protective properties
666 even after prolonged exposure to B117 salt spray.
- 667 • The findings from the damage indices show that optimal MXene concentrations (1.0% and 0.5%)
668 provide enhanced protection in MXene-epoxy coatings. In contrast, coatings with lower MXene
669 concentrations or those containing large agglomerates experience reduced protective capabilities over
670 time. Water absorption studies also reinforce these observations. MXene's intrinsic barrier properties,
671 attributed to its high aspect ratio that establishes complex pathways, combined with the reduced
672 porosity and void sizes in the epoxy matrix due to its addition, culminate in a highly resistant composite
673 against corrosive agents.
- 674 • Significantly, this study established MXene's potential in improving the anti-aging properties of epoxy.
675 By reducing the formation of free radicals and enhancing the corrosion resistance of the composite,
676 MXene has demonstrated its potential as a novel solution for epoxy degradation issues.

677 In conclusion, this study investigates the integration of $Ti_3C_2T_x$ MXene into epoxy coatings and its potential
678 to bolster anti-corrosion properties through multiple mechanisms. $Ti_3C_2T_x$ MXene significantly enhances the

679 barrier properties of epoxy coatings due to its high aspect ratio and layered structure. This configuration
680 establishes complex, tortuous pathways for corrosive agents, effectively impeding their penetration and
681 delaying the onset of corrosion processes on the substrate. Micro-CT analysis demonstrates that MXene
682 uniformly disperses within the epoxy matrix, effectively filling voids and reducing defects that could initiate
683 corrosion, thus preserving the integrity of the coating under various conditions. Furthermore, the observed
684 decrease in free radical formation indicates that MXene not only improves the physical and chemical properties
685 but also significantly boosts the oxidative stability of the coatings. This enhancement prolongs the lifespan of
686 materials, as degradation in epoxy typically leads to diminished performance and facilitates the initiation and
687 propagation of microcracks. These attributes collectively contribute to the superior performance of MXene-
688 epoxy composites in corrosive environments, underscoring their potential in extending the application range of
689 these coatings to sectors such as marine, automotive, and aerospace where durability and reliability are critical.
690 Therefore, the discovery in this manuscript provides guidance for designing and exploiting a high-performance
691 MXene reinforced polymeric composites. The incorporation of MXene nanoparticles into epoxy coatings results
692 in a significant improvement in their protective properties, paving the way for the development of advanced,
693 long-lasting, and multifunctional coatings for various applications.

694 **CRediT authorship contribution statement**

695 **Xingyu Wang:** Conceptualization, Data curation, Investigation, Formal analysis, Methodology,
696 Validation, Writing - original draft, Writing - review & editing. **Sampada Koirala:** Formal analysis,
697 Methodology, Validation, Writing - original draft. **Luyang Xu:** Formal analysis, Methodology, Validation.
698 **Qiaobin Li:** Formal analysis, Methodology, Validation. **Zhibin Lin:** Supervision, Conceptualization,
699 Investigation, Methodology, Validation, Visualization, Writing - review & editing. **Xiaoning Qi:** Formal
700 analysis, Methodology, Validation. **Zhongyu Yang:** Formal analysis, Methodology, Validation. **Ying Huang:**
701 Supervision, Investigation, Writing - review & editing., **Danling Wang:** Supervision, Conceptualization,
702 Investigation, Methodology, Validation, Visualization, Writing - review & editing.

703 **Declaration of competing interest**

704 The authors declare that they have no known competing financial interests or personal relationships that
705 could have appeared to influence the work reported in this paper.

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831

832 **Figure Captions**

833 Figure 1. The scientific publications with the keywords of “MXene nanocomposite” and “MXene coating”,
834 results were obtained from the Web of Science.

835 Figure 2. Schematic of the fabrication process of MXene-polymer nanocomposite.

836 Figure 3. (a) XRD results of pure MXene samples, (b) MAX residuals found in 25C MXene, (c) exfoliated
837 MXene sheets in 25C sample, and (d) exfoliated MXene sheets in 80C sample.

838 Figure 4. (a) Particle size distribution of 25C and 80C MXene. SEM images of (b) 25C MXene particles and
839 (b) 80C particles. TEM images of (d) 25C and (e) 80C MXene nanosheets after ultrasonication. EDS
840 images of fracture surface of epoxy that reinforced by (f) 25C and (g) 80C MXene.

841 Figure 5. Micro-CT images of neat epoxy (a) transparent 3-D image, (b) (c) cross-sectional images.

842 Figure 6. Cross-sectional images of MXene epoxy obtained by Micro-CT, (a) to (d) samples with 25C MXene
843 at 0.1, 0.5, 1.0 and 2.0 wt.%. (e) to (h) samples with 80C MXene at 0.1, 0.5, 1.0 and 2.0 wt.%.

844 Figure 7. Bode plots of the MXene-epoxy coatings at the initial stage, (a) impedance plot, and (b) phase angle
845 plot.

846 Figure 8. Bode plots of the MXene-epoxy coatings after exposure, (a) impedance plot after 250 hours, (b)
847 phase angle plot after 250 hours, (c) impedance plot after 500 hours, (d) phase angle plot after 500
848 hours.

849 Figure 9. EEC model for each corrosion stage.

850 Figure 10. Images of MXene-epoxy coatings after salt spray exposure, (a) 250 hours and (b) 500 hours. EDS
851 images (c) to (f) of cross-sectional surface for MXene-epoxy nanocomposite.

852 Figure 11. Damage index for the coating degradation assessment: (a)-(b), (c)-(d)

853 Figure 12. Protection indices of epoxy composites reinforced by (a) 25C and (b) 80C MXene.

854 Figure 13. Water uptake of epoxy composites reinforced by (a) 25C and (b) 80C MXene.

855 Figure 14. ESR spectra measurement of samples after exposure to corrosive environment.

856 **Table Captions**

857 Table 1. The corrosion stage of MXene-epoxy coated samples after exposure.
858