

# The Glycine N-Methyltransferase Case Study: Another Challenge for QM-Cluster Models?

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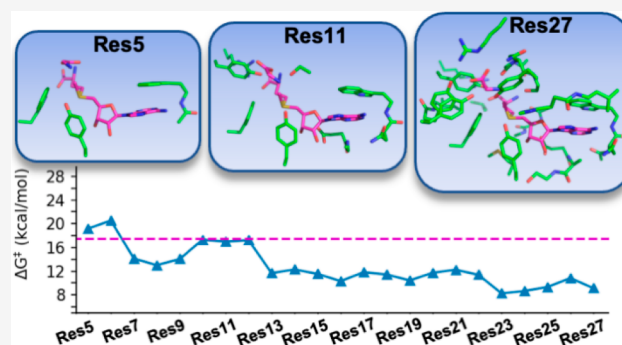


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**ABSTRACT:** The methyl transfer reaction between SAM and glycine catalyzed by glycine N-methyltransferase (GNMT) was examined using QM-cluster models generated by Residue Interaction Network Residue Selector (*RINRUS*). *RINRUS* is a Python-based tool that can build QM-cluster models with rules-based processing of the active site residue interaction network. This way of enzyme model-building allows quantitative analysis of residue and fragment contributions to kinetic and thermodynamic properties of the enzyme. Many residue fragments are important for the GNMT catalytic reaction, such as Gly137, Asn138, and Arg175, which interact with the glycine substrate, and Trp30, Asp85, and Tyr242, which interact with the SAM cofactor. Our study shows that active site fragments that interact with the glycine substrate and the SAM cofactor must both be included in the QM-cluster models. Even though the proposed mechanism is a simple one-step reaction, GNMT may be a rather challenging case study for QM-cluster models because convergence in energetics requires models with >350 atoms. “Maximal” QM-cluster models built with either qualitative contact count ranking or quantitative interaction energies from functional group symmetry adapted perturbation theory provide acceptable results. Hence, important residue fragments that contribute to the energetics of the methyl-transfer reaction in GNMT are correctly identified in the RIN. Observations from this work suggest new directions to better establish an effective approach for constructing atomic-level enzyme models.



## INTRODUCTION

In computational enzymology, scientists encounter many complex problems in attempting to understand enzymatic reaction mechanisms on an atomic scale. Ontologies and automated workflows are lacking that can fragment the macromolecule into a computationally achievable size with either multiscale quantum mechanics/molecular mechanics (QM/MM) or QM-only “cluster models”. Without a standard set of building principles available in the community, automation, reproducibility, and calibration are challenging. There have been some recent successful attempts to calibrate quantum chemical levels of theory (one-electron basis sets, wave function or density functional choice, and implicit solvation scheme) for modeling atomic-level enzyme kinetics.<sup>1,2</sup> However, an overall lack of automation and model design standards raise the learning curve for new practitioners, prevent a deeper understanding of differences between QM/MM and QM-cluster enzymology, and inhibit quantitative comparison to experimental results.

*RINRUS* (Residue Interaction Network Residue Selector), is a stand-alone Python-based toolkit developed by our group that

integrates mathematical and chemical analytics tools for systematic and quantitative study of protein structure and reactivity within the framework of QM-cluster models.<sup>3</sup> The software designs front-end atomic-level models based on several schemes ranking the importance of active site residues and fragments, primarily employing (1) simple structural metrics such as the radial distance of fragments from a defined active site “center”, (2) qualitative descriptors derived from the residue interaction network,<sup>4–6</sup> or (3) quantitative descriptors derived from functional group/intramolecular symmetry adapted perturbation theory (F/I-SAPT).<sup>7–11</sup> *RINRUS* contains an algorithmic method for trimming residues, capping atomic valencies, and freezing the C<sub>α</sub>/C<sub>β</sub> atoms of residues. Thus, reproducibility is embedded into the *RINRUS*-designed QM-

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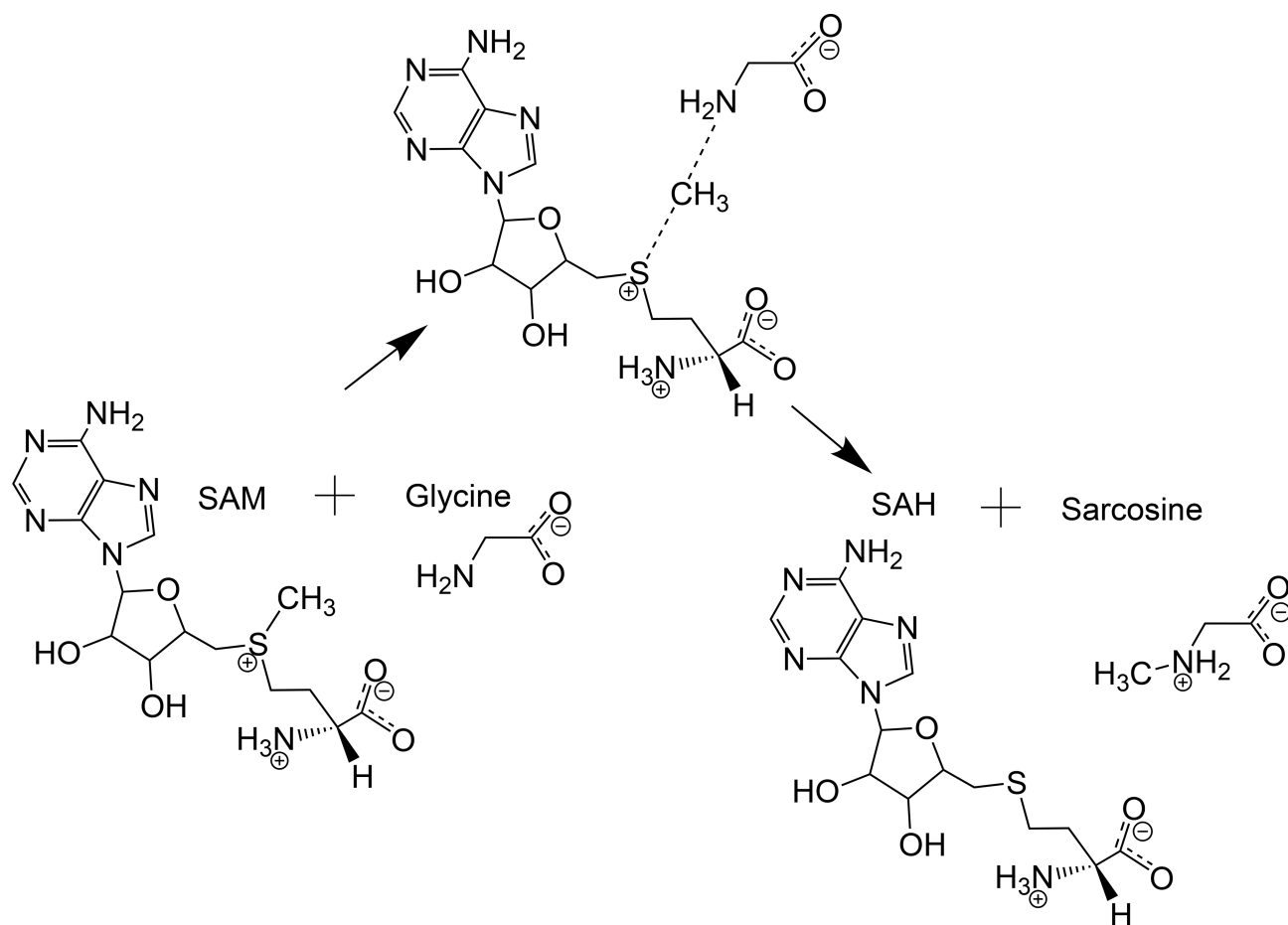
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**Scheme 1. Reaction Mechanism of the Methyl Transfer by Glycine *N*-Methyltransferase (GNMT), Which Catalyzes the Transfer of a Methyl Group from *S*-Adenosylmethionine (SAM) to Glycine, Resulting in the Formation of *S*-Adenosylhomocysteine (SAH) and Sarcosine (*N*-methylglycine)<sup>a</sup>**



<sup>a</sup>In this one-step reaction, the methyl group is located between the S atom of SAM and N atom of glycine.

cluster model input files, which are generated and can be subsequently run using several popular open-source and commercial quantum chemistry packages.

This work uses *RINRUS* to study the proposed mechanism for glycine *N*-methyltransferase (GNMT). GNMT belongs to the enzyme class of methyltransferases, which catalyze the transfer of a methyl group from an *S*-adenosyl methionine (SAM, AdoMet) cofactor to small-molecule substrates. SAM is a well-known methyl donor and is converted to *S*-adenosyl homocysteine (SAH, AdoHcy) in the process. The methyltransferase class of enzymes is druggable targets implicated in several metabolic diseases,<sup>12–15</sup> and thus atomic-level exploration and calibration is valuable.

One well-studied example of a methyltransferase enzyme is human catechol-*O*-methyltransferase (COMT). An X-ray crystal structure of COMT (PDB: 3BWM) complexed with SAM and a 3,5-dinitrocatechol inhibitor at 1.98 Å resolution<sup>16</sup> has been a focus of several QM/MM and QM-cluster model benchmark efforts.<sup>17–24</sup> The nonspherical nature of the COMT active site provides a useful case study to demonstrate slow convergence of energies and properties with respect to model size when expanding radially from the Mg<sup>2+</sup> ion at the center of the active site. Recently, a QM-cluster model study based on the X-ray crystal structure of human COMT was published by our group to explore the convergence of the energetics of the methyl

transfer reaction with respect to growing QM-cluster model size using five different model-building schemes.<sup>19</sup> Activation free energies and reaction free energies were computed by using density functional theory (DFT) for 550 unique QM-cluster models built with the *RINRUS* toolkit. *K*-means clustering was used to identify a set of residues that contribute systematically for convergence of  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  with a modest model size of 200–300 atoms. The findings also indicate that building models stepwise using residues ranked by the number of interactions with a substrate/cofactor led to accelerated convergence within 2 kcal/mol of the largest ~500-atom QM-cluster models.<sup>19</sup>

Another type of methyltransferases is glycine *N*-methyltransferase (GNMT), which is found in endocrine system organs of several mammals.<sup>25</sup> GNMT transfers the methyl group from SAM to glycine, forming sarcosine (Scheme 1). Unlike many methyltransferases, GNMT is known for regulating the SAM/SAH ratio,<sup>14,26–28</sup> which is important for SAM-dependent methyltransferase in cells. GNMT and COMT both belong to the AdoMet methyltransferase class I family, sharing a structural core; yet there are two fundamental differences—the GNMT active site is metal-free, and the methyl group of SAM is transferred to a nitrogen atom on a glycine substrate rather than an oxygen on a catechol substrate.<sup>29</sup> The first X-ray crystal structure of GNMT in complex with SAM was determined at 3.0 Å resolution and has a unique “closed” structure.<sup>30</sup> It has been

observed that in the active site of GNMT, the binding of SAM and glycine follows a sequential order, with SAM binding before glycine,<sup>31</sup> and the enzyme shows a sigmoidal rate behavior with respect to SAM concentration.<sup>26,32</sup> Compared to COMT, there are fewer computational studies on GNMT. Our research on COMT made a case that ranking residue importance based on RIN descriptors can be used to build reliable QM-cluster models with a size of 200–300 atoms. This work will test the same hypothesis regarding the proposed GNMT reaction mechanism.

The apparent kinetic rate constant ( $k_{\text{cat}}$ ) of the wild-type GNMT from rat liver cytosol was measured to be  $27.0 \text{ min}^{-1}$  at  $30^\circ\text{C}$  and  $\text{pH } 7.2$ .<sup>33</sup> The free energy of activation converted from the rate constant of the wild-type protein is  $18.2 \text{ kcal/mol}$ . A R175 K mutation reduced the binding affinity of glycine but not of SAM, indicating that Arg175 is important for glycine ligand binding. The rate constant of the R175 K mutant was measured to be  $9.94 \text{ min}^{-1}$  with a significantly reduced  $k_{\text{cat}}/K_{\text{M}}^{\text{SAM}}$  and  $k_{\text{cat}}/K_{\text{M}}^{\text{Gly}}$  compared to the wild-type, which indicated that this mutation reduced catalytic efficiency.<sup>30</sup> In 2003, X-ray crystal structures of the wild-type GNMT in complex with SAM and acetate (a known potent competitive inhibitor of glycine)<sup>33,34</sup> and the R175 K mutated enzyme in complex with SAM (PDBs: 1NBH and 1NBI)<sup>33</sup> were determined with similar resolution ( $3.0$  and  $2.8 \text{ \AA}$  resolution, respectively). More recently, the activity of rat GNMT was measured radiometrically using steady-state kinetic measurements by Zhang and Klinman.<sup>29</sup> The rate constant of GNMT was determined to be  $174.5 \pm 23.2 \text{ min}^{-1}$  for the wild type protein at  $37^\circ\text{C}$  and neutral pH under saturated substrate conditions. The free energy of activation can be converted from the rate constant of the wild-type protein to  $17.4$ – $17.6 \text{ kcal/mol}$ .

The first theoretical study of the GNMT active site kinetics was a QM-cluster model study using different sized (ranging from 22 to 98 atoms) models, in 2005 by Velichkova and Himo.<sup>35</sup> The models were based on the acetate-inhibited X-ray crystal structure (PDB: 1NBH) and employed DFT with the B3LYP<sup>36–40</sup> functional and 6-31G(d,p) basis sets for geometry optimization. At the optimized geometry, a single-point energy calculation was performed with a larger basis set [B3LYP/6-311+G(2d,2p)]. Implicit solvation effects were calculated as a single-point energy based on the optimized geometry at the B3LYP/6-31G(d,p) level using the conductor-like COSMO<sup>41–44</sup> solvation model with the polarizable dielectric continuum (dielectric constant  $\epsilon = 4$ ). However, implicitly solvated single point energies on top of gas-phase optimized geometries can sometimes produce unreliable kinetics for QM-cluster models.<sup>45</sup> The smallest model used by Velichkova and Himo included only the SAM, glycine, and Arg175 fragments. With this smallest model, the activation energy and reaction energy were predicted to be  $11.2$  and  $-20.1 \text{ kcal/mol}$ , respectively. From the largest model in their study (including SAM, glycine, Arg175, Tyr21, Gly137, Asn138, and Tyr194) the free energies of activation and reaction were computed to be  $15.0$  and  $-14.1 \text{ kcal/mol}$ . By adding one residue at a time to the smallest model, the change in activation free energies ranged from  $-1.3$  (SAM + glycine + Arg175 + Gly137) to  $+6.4 \text{ kcal/mol}$  (SAM + glycine + Arg175 + Tyr194), providing a semiquantitative estimation of the influence of active site residues on kinetics/thermodynamics. A later computational study by Moliner et al.<sup>46</sup> used QM/MM MD simulations with the rat-liver X-ray crystal structure and found the free energy of activation to be  $15.2 \text{ kcal/mol}$ , which is close to the value of the largest model of Velichkova and Himo as well as experimental

data. In that study, the QM region included only the SAM and glycine and was computed at the AM1 or M06-2X/6-31+G(d,p) level of theory.

In this work, active site fragment ranking schemes based on residue interaction network information are employed for building different sized QM-cluster models to study the GNMT N-demethylation reaction mechanism. Systematic QM-cluster model building provides insight into the function and importance of various residues and how to build QM-cluster models in a more general way that is not only limited to chemical intuition and consideration of experimental results. Furthermore, an increasing number of case studies and benchmarks performed in our group will provide guidance on employing residue interaction networks to build the “right model” for the “right reason”, which will lead to an improved comparison with experimental biokinetics and drug discovery. RINRUS-based QM-cluster models applied to the larger class of methyltransferases may elucidate previously unknown trends or commonalities in structure and function.

## COMPUTATIONAL METHODS

The X-ray crystal structure of GNMT in complex with S-adenosyl-L-methionine (SAM, positively charged) and acetate (ACT, negatively charged; PDB: 1NBH) was used to construct the models for quantum mechanical computations. The acetate inhibitor was changed to glycine (Gly, negatively charged form) in all models. In this study, only the wild type protein is used to construct models. RINRUS then utilizes the *reduce* program<sup>47</sup> to add H atoms to the PDB file.

In the early computational study by Velichkova and Himo, eight small QM models were studied, focusing on the effects of six important residues. The largest model that Velichkova and Himo<sup>35</sup> investigated contained SAM, glycine, Tyr21, Gly137, Asn138, Arg175, and Tyr194. We also created a set of QM-cluster models composed of these residue fragments, labeled **Himo-A**, **Himo-B**, **Himo-C**, **Himo-D**, **Himo-E**, **Himo-F**, **Himo-G**, **Himo-L1**, **Himo-L2**, and **Himo-L3**, and **Himo-L+H**, where **Himo-A** through **Himo-G** include SAM, Gly, and Arg175 and one or two additional active site residues (see Table S1); **Himo-L1**, **Himo-L2**, and **Himo-L3** are three conformations of the largest (“L”) model constructed by Velichkova and Himo that contains seven fragments and 98 atoms with seven atoms kept frozen. **Himo-L1** and **Himo-L3** are constructed from the X-ray crystal structure with the OH group on Tyr21 pointing toward or away from SAM respectively in the initial TS guess, while **Himo-L2** is constructed starting from the optimized **Himo-L+H** TS structure but with His142 removed. Two additional models were tested, including the **Himo-Seed** model, which has only SAM and a Gly substrate, and the **Himo-L+H** model, which has all the fragments of **Himo-L1** plus His142. His142 clearly has a large impact on the kinetics and thermodynamics in the study by Velichkova and Himo. However, it was not included in their largest QM-cluster model. In all Himo-based QM-cluster models, the SAM cofactor was truncated the same way as in the Velichkova and Himo study,<sup>35</sup> to an ethylmethylthioethane cation. All other QM-cluster models reported in this study used the full SAM molecule. Selected frozen atoms for the Himo-based models were equivalent to those frozen in the Velichkova and Himo study.

For constructing new QM-cluster models with RINRUS, a residue interaction network (RIN) of the entire protein must first be realized. In this study, the program *probe*<sup>48</sup> was used to

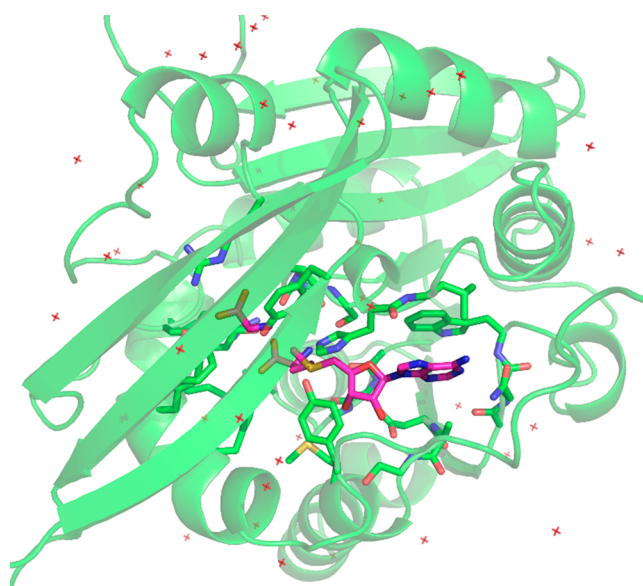


generate residue–residue atom contact information. Then, a “seed” needs to be defined based on the fragments fundamental to the catalytic reaction. Ranking schemes for fragment selection then prioritize fragments that have the highest number of interatomic contacts, the highest number of interaction types, or the largest magnitude interaction energies with the seed. In the first set of *RINRUS* QM-cluster models, SAM and Gly were defined as the seed, where contact counts between atoms of active site fragments with atoms of both the glycine substrate and SAM cofactor from the *probe*-based RIN were summed. A total of 25 residues (Table 1) were identified having one or more interatomic contacts<sup>48</sup> with Gly or SAM (the active site of the

**Table 1. Residue and Trimming Information for Various Sized QM-Cluster Models Built Based on Residue Fragments–Seed (SAM+Gly) Interaction Counts of *probe* RIN<sup>a</sup>**

		contact counts	charge	# of atoms	frozen atoms	$\Delta G^\ddagger$	$\Delta G_{\text{rxn}}$
1	SAM293		+1		X		
2	Gly294		0		X		
3	Trp117	627	0		C <sub>α</sub>		
	<i>Asn116</i> (mc)		0		C <sub>α</sub>		
4	Trp30	385	0		C <sub>α</sub> C <sub>β</sub>		
5	Tyr21	300	0	129	C <sub>α</sub> C <sub>β</sub>	19.2	−20.2
6	<i>Asn116</i> (sc)	269	0	136		20.5	−17.9
7	Ser139	244	0	145	C <sub>α</sub>	14.1	−20.9
8	Ala86	190	0	160	C <sub>α</sub>	13.0	−21.4
	<i>Asp85</i> (mc)		0		C <sub>α</sub>		
9	Arg40	189	+1	182	C <sub>α</sub> C <sub>β</sub>	14.0	−24.3
10	<i>Asp85</i> (sc)	183	0	187	C <sub>β</sub>	17.3	−17.9
11	Tyr194	179	0	206	C <sub>α</sub> C <sub>β</sub>	17.0	−22.5
12	Tyr242	172	0	225	C <sub>α</sub> C <sub>β</sub>	17.2	−24.3
13	Asn138	165	0	239	C <sub>α</sub>	11.7	−20.8
14	Arg175	164	+1	261	C <sub>α</sub> C <sub>β</sub>	12.3	−17.6
15	Tyr33	128	+1	280	C <sub>α</sub> C <sub>β</sub>	11.5	−22.3
16	His142	118	+1	295	C <sub>α</sub> C <sub>β</sub>	10.3	−22.6
17	Tyr220	111	+1	314	C <sub>α</sub> C <sub>β</sub>	11.8	−23.8
18	Ser87	84	+1	325	C <sub>α</sub>	11.4	−23.7
19	Ala115	57	+1	332	C <sub>α</sub>	10.5	−24.5
20	Ile34	55	+1	351	C <sub>α</sub> C <sub>β</sub>	11.7	−23.6
21	Gly66	54	+1	363	C <sub>α</sub>	12.3	−18.7
	<i>Cys65</i> (mc)				C <sub>α</sub>		
22	Ala64	48	+1	370	C <sub>α</sub>	11.4	−19.2
23	Gly137	39	+1	377	C <sub>α</sub>	8.3	−29.7
24	Val69	28	+1	391	C <sub>α</sub> C <sub>β</sub>	8.7	−28.4
25	Leu136	24	+1	398	C <sub>α</sub>	9.3	−28.9
26	Leu143	18	+1	417	C <sub>α</sub>	10.9	−26.8
27	Met90	15	+1	432	C <sub>α</sub>	9.1	−29.7
	Himo-L1		+1	98		13.0	−18.0

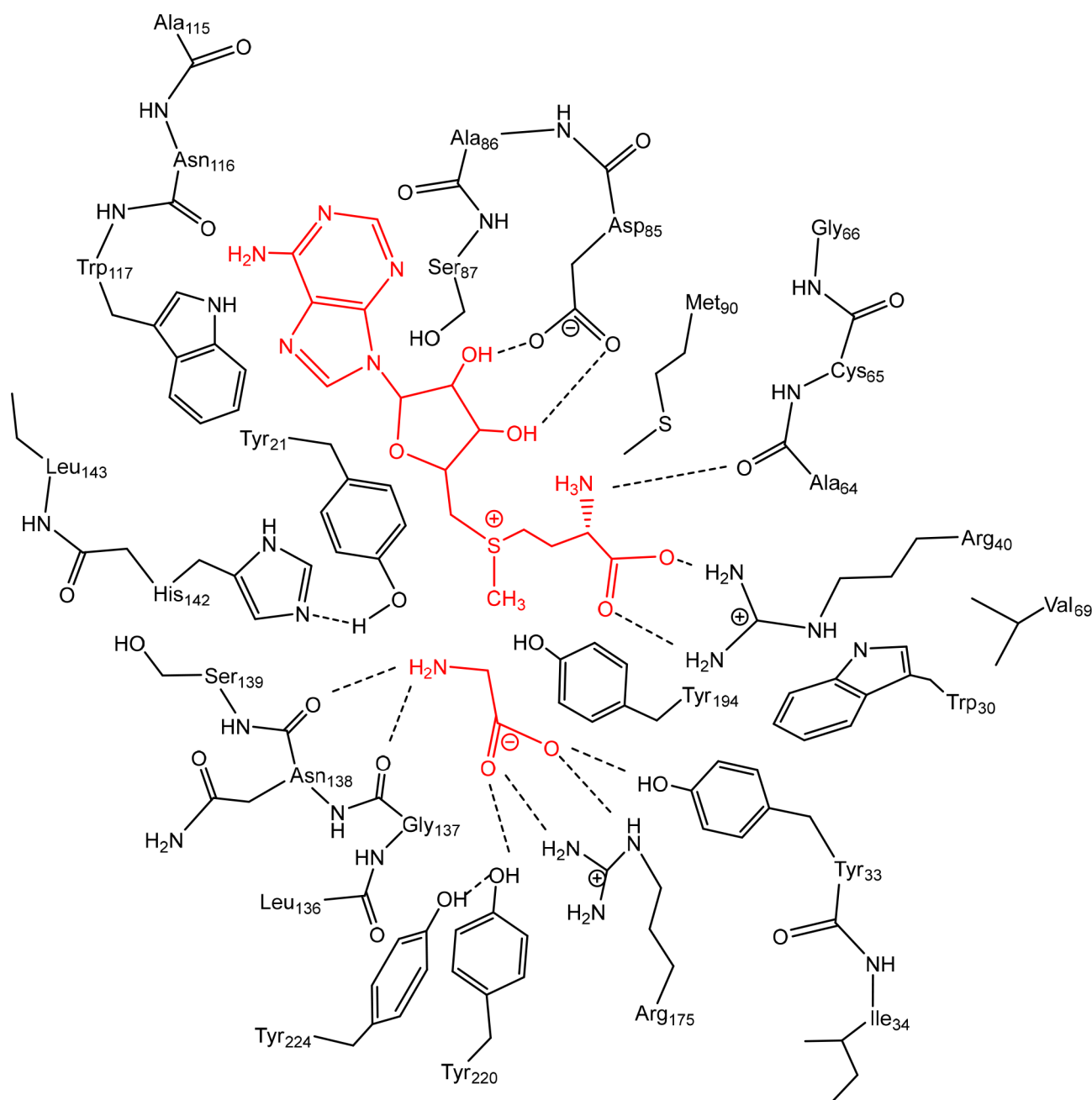
<sup>a</sup>C<sub>α</sub> indicates frozen alpha-carbon atoms for that residue. C<sub>α</sub> C<sub>β</sub> indicates frozen alpha- and beta-carbons for that residue. X indicates no atom was frozen in the specified residue/fragment in the model. Residue ID numbers written in italics were not identified in the smaller RIN as having contact with the seed but were added as bridging residues connecting two discrete residues, such as the Res8 model. Asp85 was not ranked among the top 8 but was added as a bridging residue. Gibbs free energies of activation and reaction are shown (in kcal/mol).



**Figure 1.** Active site of the X-ray crystal structure 1NBH chain A and the 3D structure shown in stick representation of the maximal model (Res27). The seed that includes the SAM and Gly is highlighted in magenta, and all other residues in the RIN are shown in green.

enzyme is shown in Figure 1 with the 25 residues shown as green sticks and Gly+SAM shown as magenta sticks; in addition, the 2D structure for this RIN is shown in Figure 2), and these residues were ranked based on the decreasing number of contacts with the seed (shown in Table 1 with the charge of each model and number of contacts between each residue fragment and the seed). His142 was singly protonated on the  $\delta$  position as in the studies by Velichkova and Himo<sup>35</sup> and Moliner et al.<sup>46</sup> Among the 25 residues that interact with the seed fragments, Gly137 and Tyr194 interact with both SAM and the glycine substrate; Tyr33, Asn138, Arg175, Tyr220, and Tyr242 interact with the glycine substrate only; and the remaining 18 residue fragments interact with SAM only. The details of *probe* interaction types between the residue fragments and seed fragments are shown in Table S2. Based on the generated RIN, either main chain, side chain, or entire residues were trimmed and capped by *RINRUS* to fulfill the atom valency in each model. The smallest *RINRUS*-designed QM-cluster model includes the SAM; glycine substrate; and Trp117, Trp30, and Tyr21 residues. These three residues all have more than 300 interaction counts with the seed. By increasing one fragment at a time based on the *probe* ranking, a total of 23 models were constructed with the largest “maximal” model containing 25 residues/fragments and the SAM+glycine seed (27 fragments total, 432 atoms). For the *probe* RIN with a Gly seed, 10 residue fragments were detected to interact with Gly. The smallest model has five residue fragments (SAM + Gly + Tyr21 + Trp117 + Arg175, 159 atoms). Tyr242 and Asn138 were ranked sixth and seventh in the ranking scheme (Table 2). Gly137 was found to have the least number of interaction counts with the substrate Gly, and with Gly137 included, the largest QM-cluster model with a Gly seed has 253 atoms.

The largest QM-cluster model with the SAM+Gly seed that was constructed from the *probe*-based RIN was used to compute seed–residue interaction energies with functional-group symmetry-adapted perturbation theory (F-SAPT),<sup>8</sup> which is implemented in the PSI4 v1.5<sup>49</sup> program. F-SAPT can analyze and partition the noncovalent interaction energy between two



**Figure 2.** 2D structure of the maximal Res27 QM-cluster model, which includes all residues that interact with SAM and glycine in PDB 1NBH.

bodies into electrostatic, exchange-repulsion, induction, and dispersion energies as in conventional SAPT. However, F-SAPT can further decompose one of the parent bodies into smaller subfragments such as user defined chemically meaningful functional groups. Here, one body in the F-SAPT computation is defined as the seed (Gly or Gly+SAM), and the second body is defined as the main chain and/or side chain of each amino acid residue present in the maximal QM-cluster model. The individual interaction energies between the seed and each main chain and/or side chain fragments are partitioned, and these F-SAPT interaction energies are incorporated into ranking the importance of residue fragments in the catalytic reaction step.<sup>10,19</sup>

As in our recently published work,<sup>10</sup> the zeroth-order formulation of F-SAPT, F-SAPT0, is used for computing interaction energies with the jun-cc-pVDZ basis set on all atoms

and frozen core electrons.<sup>8,11</sup> Residues in the network are ranked according to the decreasing absolute value of the total interaction energy. When the substrate Gly alone was used as the seed, 22 residue fragments interact with Gly with an absolute interaction energy higher than 0.5 kcal/mol (Table 3), among which 16 have interaction energies higher than 1.0 kcal/mol. When SAM and Gly fragments are combined as the seed, 24 residue fragments interacting with SAM or Gly have an absolute interaction energy higher than 0.5 kcal/mol (Table 3), and 22 of the residue fragments have interaction energies higher than 1.0 kcal/mol. Two sets of F-SAPT-based QM-cluster models were incrementally built, with the size of the Gly-only seed and SAM + Gly seed models ranging from 81 to 384 and 81 to 385 atoms, respectively.

All subsequent QM-cluster model computations were performed using the Gaussian 16 program.<sup>50</sup> Density functional

**Table 2. Residue Information for the Various Sized QM-Cluster Models Built Based on Residue Fragments-Seed (Gly) Interaction Counts of *probe-RIN*<sup>a</sup>**

		contact counts	res charge	# of atoms	$\Delta G^\ddagger$	$\Delta G_{\text{rxn}}$
1	Gly294		−1			
2	SAM293		0			
3	Trp117		0			
4	Trp21		0			
5	Arg175	164	+1	129	17.3	−18.5
6	Tyr242	172	+1	148	16.7	−22.1
7	Asn138	165	+1	167	16.8	−37.8
8	Tyr33	128	+1	186	15.3	−25.9
9	Tyr194	116	+1	205	15.6	−13.5
10	Tyr220	111	+1	224	14.5	−19.9
11	Trp30	37	+1	246	13.8	−16.1
12	Gly137	17	+1	253	5.7	−32.4

<sup>a</sup>Gibbs free energies of activation and reaction are shown (in kcal/mol).

theory (DFT) with the hybrid B3LYP exchange-correlation functional<sup>36–40</sup> was employed with the 6-31G(d') basis set for N, O, and S atoms<sup>51–53</sup> and the 6-31G basis sets for C and H atoms.<sup>54</sup> QM-cluster models incorporated the Grimme D3 (Becke-Johnson) dispersion correction (GD3BJ)<sup>55,56</sup> and a conductor-like polarizable continuum model (CPCM)<sup>57,58</sup> with

a cavity built using the universal force field (UFF) atomic radii, a nondefault electrostatic scaling factor of 1.2, and the default parameters for water with an attenuated dielectric constant of  $\epsilon = 4$ . This dielectric constant value<sup>35</sup> has been previously determined as appropriate for simulating the less-polarized environment within an enzyme active site.<sup>59,60</sup> Implicit solvation was incorporated into all geometry optimizations and harmonic frequency calculations. The computational methodology has been successfully employed in several enzyme studies which have reliable comparisons to experimental results.<sup>19,61–64</sup> Unscaled harmonic vibrational frequency calculations were used to identify all stationary points as either minima (no imaginary frequency) or transition states (only one imaginary frequency). Reactants and products corresponding to the methyl transfer TS were located by following the intrinsic reaction coordinate (IRC).<sup>65,66</sup> The same “freeze code” scheme of Gaussian 16 used in other enzyme studies by our group is used in this study. It needs to be noted that all atoms in the SAM and glycine fragments are unfrozen in all *RINRUS* models. Zero-point energies (ZPE) and thermal enthalpy and free energy corrections were computed at 1 atm and 310 K.

The structures of every reactant, transition state, and product of every QM-cluster model reported here are represented in a PDB (Protein Data Bank) formatted file included in SI. In the PDB files, the Cartesian coordinates are listed along with the corresponding atom freeze code information. The atom freeze

**Table 3. Residue Main-Chain or Side-Chain (mc/sc) Fragment Ranking Based on SAPT Absolute Total Energy Using Gly As the Seed or SAM Plus Gly As the Seed<sup>a</sup>**

	Gly only		$E_{\text{int}}$	charge	# of atoms	$\Delta G^\ddagger$	$\Delta G_{\text{rxn}}$	Gly plus SAM		$E_{\text{int}}$	charge	# of atoms	$\Delta G^\ddagger$	$\Delta G_{\text{rxn}}$
1	Gly294							Gly294						
2	SAM293							SAM293						
3	Arg175	sc	−110.0	1	81	30.7	−9.2	Arg175	sc	−77.7	1	81	30.7	−9.2
4	Arg40	sc	−33.3	2	103	29.6	−18.7	Asp85	sc	−66.0	0	91	15.1	−33.2
5	Asp85	sc	24.7	1	113	15.1	−22.3	Arg40	sc	−25.8	1	113	15.1	−22.3
6	Asn138	sc	−20.6	1	125	14.0	−19.8	Ala64	mc	−24.5	1	125	15.1	−22.3
7	Tyr220	sc	−15.1	1	144	13.8	−19.2	<b>Leu136</b>	mc	−21.2	1	137	16.7	−22.4
8	Tyr33	sc	−12.8	1	163	17.2	−16.9	Tyr220	sc	−16.8	1	156	17.3	−19.3
9	Asn138	mc	7.0	1	170	14.8	−27.4	Asn138	sc	−13.7	1	170	14.3	−25.6
10	Gly137	mc	6.5	1	184	10.7	−25.4	<b>Ser139</b>	sc	−13.6	1	181	11.7	−23.4
11	His142	sc	5.6	1	199	12.8	−36.9	His142	sc	−12.1	1	196	16.1	−19.8
12	Ala64	mc	2.2	1	214	13.4	−36.8	<b>Trp30</b>	sc	−10.8	1	218	10.4	−32.5
13	Tyr194	sc	−1.6	1	233	12.3	−35.6	Tyr33	sc	−10.4	1	237	9.1	−32.7
14	Tyr21	sc	1.4	1	252	11.3	−25.1	Trp117	sc	−8.9	1	259	9.0	−30.9
15	Ala86	mc	−1.2	1	269	13.9	−24.5	Tyr194	sc	−5.4	1	278	9.0	−30.9
16	Met90	sc	1.1	1	284	14.2	−24.5	Gly137	mc	−4.9	1	278	9.0	−30.9
17	Tyr242	sc	−1.0	1	303	8.2	−31.2	Tyr21	sc	4.6	1	297	11.9	−21.0
18	Ala115	mc	0.9	1	315	10.4	−24.8	Ala86	mc	4.4	1	304	10.9	−18.6
19	Asn116	mc	−0.9	1	315	10.4	−24.8	<b>Ala115</b>	mc	−4.1	1	318	13.5	−16.8
20	Trp30	sc	−0.6	1	337	9.2	−23.3	Met90	sc	−3.5	1	333	13.7	−16.4
21	Trp117	sc	0.6	1	361	7.7	−23.7	<b>Asn116</b>	mc	−3.1	1	333	13.7	−16.4
22	Ser87	sc	−0.5	1	365	9.5	−28.4	Tyr242	sc	−2.3	1	352	11.2	−23.8
23	Leu143	sc	−0.5	1	384	9.5	−28.5	Asn138	mc	−1.8	1	352	11.2	−23.8
24	Leu136	mc	−0.3					Ser87	sc	1.3	1	363	9.4	−27.2
25	Ile34	sc	−0.3					Leu143	sc	−0.8	1	382	10.1	−27.3
26	Asn116	sc	0.3					Ala86	sc	−0.7	1	385	9.7	−27.8
27	Ala86	sc	−0.2					Val69	sc	−0.1				
28	Ser139	sc	0.1					Asn116	sc	0.0				
29	Val69	sc	0.0					Ile34	sc	0.0				

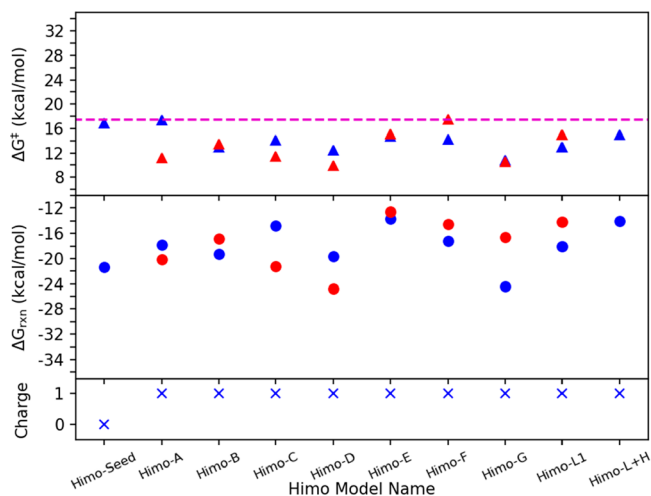
<sup>a</sup>The F-SAPT total interaction energy (in kcal/mol) of each residue, charges and number of atoms of each model are listed along with the free energy of activation and reaction (in kcal/mol). The residues highlighted in bold are not included in *ResX-fsapt1* models.

code is specified in the last column of each line in the PDB file and is denoted by either “0” or “-1”. Specifically, a freeze code of “0” indicates that the atom is unfrozen and can move freely during the computation, while a code of “-1” indicates that the atom is frozen and remains fixed in its position throughout the computation.

The following labeling scheme **ResX-Y** is used for the QM-cluster model illustrating the reaction mechanism, where X = number of residues/fragments in the model (5 to 27); Y = R (reactant), TS (transition state), or P (product). For example, **Res9-R** refers to the optimized nine-fragment reactant structure. The Gly-only seed models follow the labeling scheme **ResX-Y-Gly**. For the F-SAPT-based models, when Gly is the seed fragment, the labels **ResX-Y-fsapt1** are used, and when both Gly and SAM are included in the seed, the label **ResX-Y-fsapt2** is used.

## RESULTS AND DISCUSSION

**Himo-Based Models.** Velichkova and Himo reported activation and reaction energies ( $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$ ) of 15.0 and  $-14.1$  kcal/mol for their largest model. The minimal model (SAM+Gly) as well as models of Velichkova and Himo were retested, and the 3D cartoon of the optimized TS structures are shown in Figure S1.  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  values of all Himo-based models computed in this study are shown in Figure 3 and Table

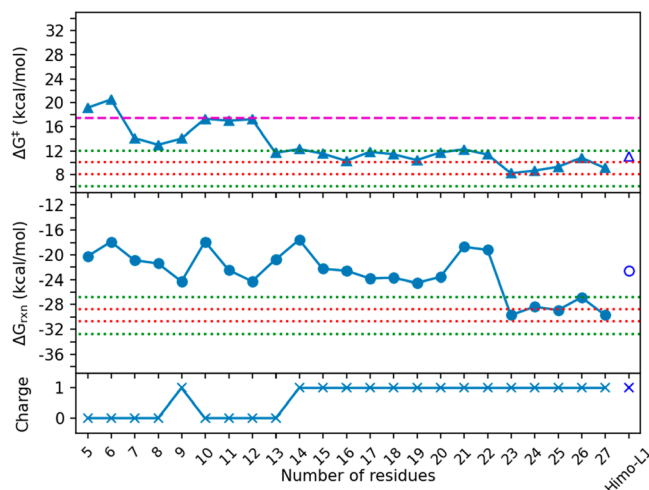


**Figure 3.** Free energy diagram of different sized **Himo**-based models (in blue, computed using B3LYP/6-31G(d')/6-31G, built the same way as Himo) along with the **Himo-Seed** model (SAM plus Gly) and Himo-largest model plus His142 (**Himo-L+H**) built based on the models in ref 35. The free energies of the original models reported in ref 35 are shown in red. Symbols on top (triangles), middle (circles), and bottom (crosses) show the free energies of activation, reaction, and charges for each model, respectively. The dashed line in magenta represents the experimental rate constant value converted to a free energy of activation of 17.5 kcal/mol.

S1. Comparatively, the initial QM-cluster models described in this work are constructed in the same fashion as Velichkova and Himo, but we utilize different basis sets, solvation schemes, and dispersion corrections. Hence, deviations from the original values of Velichkova and Himo are expected. The  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of our **Himo-seed** model (SAM+Gly) are 17.0 and  $-21.5$  kcal/mol. The **Himo-A** model (SAM+Gly+Arg175) gives  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  values of 17.3 and  $-17.9$  kcal/mol versus 11.2 and  $-20.1$  kcal/mol of model A in Velichkova and Himo, shown in

Figure 3. Our **Himo-L1** model has  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  values of 13.0 and  $-18.0$  kcal/mol at 310 K, which are lower than the values of the largest model of Velichkova and Himo (15.0 and  $-14.1$  kcal/mol) and the experimental  $\Delta G^\ddagger$  value (17.4–17.6 kcal/mol), respectively. In the study of Velichkova and Himo, adding Tyr21, Asn138, Tyr194, or Tyr21+Gly137+Asn138+Tyr194 led to a positive  $\Delta \Delta G^\ddagger$  value compared to model A. However, adding any of the core residue fragments to our **Himo-A** model leads to a negative  $\Delta \Delta G^\ddagger$  value, except for **Himo-L2** and **Himo-L3**. Our **Himo-L+H** model adds His142 to the largest model of Himo and has a  $\Delta G^\ddagger$  of 15.0 kcal/mol, which is also 2.3 kcal/mol lower than that of **Himo-A** and a  $\Delta G_{\text{rxn}}$  of  $-14.0$  kcal/mol, which is 3.9 kcal/mol higher than that of **Himo-A**. For the largest Himo model, we found three conformations, **Himo-L1**, **Himo-L2**, and **Himo-L3** (Figure S1). Large energy differences are seen in the three models (Table S1), and  $\Delta G^\ddagger/\Delta G_{\text{rxn}}$  are 13.0/ $-18.0$ , 17.4/ $-15.8$ , and 17.6/ $-16.0$  kcal/mol for **Himo-L1**, **Himo-L2**, and **Himo-L3** models, respectively. All the differences between our results and the ones from previous studies indicate that QM-cluster models including only seven residue fragments may not be an adequate enough representation of the GNMT active site to reliably depict the reaction environment, even though the largest model tested by Velichkova and Himo has a free energy of activation close to the experimental value. Hence, to gain more insight about GNMT and its catalysis, a more systematic approach is needed for finding a comprehensive QM-cluster model that eliminates size and composition deficiencies as a source of error.

**RINRUS-Based Models: probe Residue Fragment Ranking with a SAM+Gly Seed.** The  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  values of the *probe*-based RINRUS QM-cluster models and the model charges are shown in Table 1 and Figure 4. Residue effects on the free energies are also examined using  $\Delta \Delta G^\ddagger$  and  $\Delta \Delta G_{\text{rxn}}$



**Figure 4.** Free energy diagram of the 23 different sized RINRUS models along with the largest model in ref 35 built and computed in this study (B3LYP/6-31G(d')/6-31G). Symbols on top, middle, and bottom show the free energies of activation for *N*-demethylation, reaction free energies, and charges for each model, respectively. The energies of the **Himo-L1** model (98 atoms) are shown in blue empty symbols. The dashed line in magenta represents the experimental value converted free energy of activation of 17.5 kcal/mol. The red dotted lines show the tight energy convergence range (within  $\pm 1$  kcal/mol of the  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of the maximal model **Res27**); the green dotted lines show the loose energy convergence range (within  $\pm 3$  kcal/mol of the  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of the maximal model **Res27**).



[changes in  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  when the model size increases by one residue (Figure S2)]. Among various sized models, the  $\Delta G^\ddagger$  values of RINRUS models range from 8.3 to 20.5 kcal/mol,  $\Delta G_{\text{rxn}}$  values range from  $-29.7$  to  $-17.6$  kcal/mol at 310 K. The  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  values of the maximal model Res27 are 9.1 and  $-29.7$  kcal/mol (Table 1, Figure 4). In this work, we propose two definitions for the “convergence” of a model building scheme: if both  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of a model and all subsequent larger models are within either  $\pm 1$  or  $\pm 3$  kcal/mol of the  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of the maximal model in the scheme.

The two smallest models, Res5 and Res6, have  $\Delta G^\ddagger$  values higher than 19 kcal/mol (Table 1, Figure 4). Starting from Res7, the  $\Delta G^\ddagger$  values of all models are below 17.5 kcal/mol, and unsigned  $\Delta\Delta G^\ddagger$  values stabilize after Ser139 is added. Still, some significant  $\Delta\Delta G^\ddagger$  and  $\Delta\Delta G_{\text{rxn}}$  values are seen when the charged residues Arg40 (Res9,  $\Delta\Delta G_{\text{rxn}} = -2.9$  kcal/mol, Figure S2) and Asp85 (Res10,  $\Delta\Delta G^\ddagger$  and  $\Delta\Delta G_{\text{rxn}}$  of +3.3 and +6.4 kcal/mol, respectively in Figure S2) are sequentially added. From Res10 to Res12,  $\Delta\Delta G^\ddagger$  is small;  $\Delta G_{\text{rxn}}$  values become more negative. Adding Tyr194 (Res11) has a large effect on the reaction energy with  $\Delta\Delta G_{\text{rxn}} = -4.6$  kcal/mol. However, from Res12 to Res14, large  $\Delta\Delta G^\ddagger$  and  $\Delta\Delta G_{\text{rxn}}$  values are observed by adding Asn138 (Res13,  $-5.5$  and  $+3.5$  kcal/mol, respectively) and Arg175 (Res14,  $\Delta\Delta G_{\text{rxn}} = +3.2$  kcal/mol). Among fragments of Ser139, Arg40, Asp85, Tyr194, Asn138, and Arg175 that have a large impact on free energies, only Tyr194, Asn138, and Arg175 residues were included in QM-cluster models constructed by Velichkova and Himo.<sup>35</sup> The positively charged Arg175 was in every model of Velichkova and Himo as it hydrogen bonds to the two oxygen atoms of the Gly carboxyl group. To better characterize the effect of Arg175 in smaller models, Arg175 is added to Res5 to Res12, which are built from the X-ray structure, and all models are optimized.  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  values range from 12.6 to 27.2 kcal/mol and  $-23.3$  to  $-2.6$  kcal/mol, respectively (Table S3, Figure S3), and most models with Arg175 added are 4 to 7 kcal/mol higher in  $\Delta G^\ddagger$  and about 6 kcal/mol higher in  $\Delta G_{\text{rxn}}$  compared to the original models. However, no similar trends in  $\Delta\Delta G^\ddagger$  or  $\Delta\Delta G_{\text{rxn}}$  are found in the two sets of models (original Res5 to Res12 models without Arg175 and those with Arg175, Figure S3). Large changes in free energies and conformations, such as a side chain reorientation of Tyr21 and large translation of the Gly substrate, are observed in these small models (Figure S4). Energy/structure discrepancies between the sets of models with and without Arg175 suggest that smaller QM-cluster models are not reliable for studying this reaction as conformational freedom of each residue fragment is large. The Arg175 side chain hydrogen, which hydrogen bonds to the carboxyl group of Gly, may not be the only key residue for pinning down the Gly substrate in the active site or achieving energy consistency.

After Arg175 is included in the Res14 probe model, unsigned  $\Delta\Delta G^\ddagger$  values are less than 1.5 kcal/mol (Figure 4 and Figure S2), and  $\Delta G^\ddagger$  values are within 3.0 kcal/mol energy compared to that of Res27. Unsigned  $\Delta\Delta G_{\text{rxn}}$  values are less than 1.3 kcal/mol. However, convergence is disrupted by the addition of Gly66 (Res21) and Ala64 (Res22), as the  $\Delta\Delta G_{\text{rxn}}$  values are +4.9 and  $-0.5$  kcal/mol, respectively. The value of  $\Delta G_{\text{rxn}}$  for the Res22 model is 10.5 kcal/mol less negative than that of the maximal model.

Gly137 is included in Res23, which has an outsized structural effect on stabilizing the substrate, as its main-chain oxygen is a hydrogen bond acceptor to an amine hydrogen atom on the Gly substrate. The addition of Gly137 also causes the main-chain

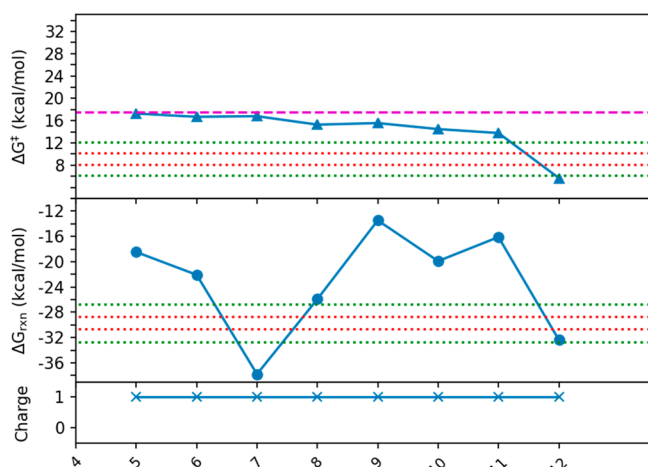
carbonyl group of Asn138 to shift and form two hydrogen bonds with the  $\text{NH}_2$  group of Gly while only one hydrogen bond is present in Res22. These newly established hydrogen bonds in Res23 make the Gly substrate a stronger nucleophile which causes a decrease in kinetic and thermodynamic values ( $\Delta\Delta G^\ddagger = -3.1$  and  $\Delta\Delta G_{\text{rxn}} = -10.5$  kcal/mol compared to Res22, Figure S2). Gly137 also causes a decrease in  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  ( $\Delta\Delta G^\ddagger = -1.3$  and  $\Delta\Delta G_{\text{rxn}} = -4.7$  kcal/mol), which was also reported by Velichkova and Himo.<sup>35</sup>  $\Delta G^\ddagger$  values of the five largest probe-based models (Res23 to Res27) range from 8.3 to 10.9 kcal/mol, and  $\Delta G_{\text{rxn}}$  values range from  $-29.7$  to  $-26.8$  kcal/mol (Figure 4). The tight convergence within  $\pm 1$  kcal/mol free energy between a model and Res27 is disrupted by the addition of Val69 (Res24,  $\Delta\Delta G_{\text{rxn}} = +1.3$  kcal/mol compared to that of Res27) and Leu143 (Res26, changes in  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  are +1.7 and +2.9 kcal/mol compared to Res27). Our looser definition of QM-cluster model convergence is achieved with the 377-atom Res23 model. The energetics of GNMT catalysis are sensitive to the addition of residue fragments even when the model sizes are relatively large. Res23 is the smallest RINRUS energy converged model, which contains all residues in the Himo-L model (98 atoms) and additional residue fragments that stabilize SAM and Gly in the active site of the enzyme.

**RINRUS-Based Models: probe Residue Fragment Ranking with a Smaller Seed (Gly Only).** It needs to be noted again that in all RINRUS-built QM-cluster models the SAM was not trimmed. SAM is a much larger molecule than free glycine, and several residues near the SAM cofactor have higher interaction counts than those that interact with Gly. For example, Trp117 (interacting only with SAM, 627 contacts) and Trp30 (348 interaction counts with SAM and only 37 with Gly) are ranked third and fourth in the RIN, while Arg175 (only interacts with Gly, 164 contacts) is ranked 14th. Hence, how to effectively weigh these interactions in order to balance the cofactor/substrate size difference may be another factor to consider when building QM-cluster models.

We hypothesized that if we include all residues that interact only with the Gly, RINRUS would capture enough residues that interact also with SAM to obtain quicker convergence of kinetics and thermodynamics with much smaller QM-cluster models compared to those converged in the original probe-ranking scheme with Gly+SAM seed. Therefore, a set of small QM-cluster models was built using probe-RIN contacts including Gly substrate as the only seed fragment.  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  for these models are shown in Table 2 and Figure 5. Consistently high  $\Delta G^\ddagger$  values are seen in the first seven models (Res5-Gly to Res11-Gly  $\Delta G^\ddagger$  ranges from 13.8 to 17.3 kcal/mol, which are 4.7 to 8.2 kcal/mol higher than the maximal SAM+Gly model Res27), then large changes are observed in  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  after adding Gly137 in Res12-Gly ( $\Delta\Delta G^\ddagger$  and  $\Delta\Delta G_{\text{rxn}}$  are  $-8.1$  and  $-16.3$  kcal/mol, respectively). Specifically, the Gly137 main chain blocks a shift of the SAM methionine group toward the Tyr194 and Gly substrate, which is observed in other models. Hence, residues that interact solely with SAM are also needed to reach energy convergence and proper conformation, which indicates that QM-cluster models built based on a RIN with only Gly as the seed fragment is not adequate for describing the methyl transfer reaction in GNMT.

**F-SAPT Interaction Energy-Based Models: Gly Only and SAM+Gly As Seeds.** Considering the limitations of using probe-SAM+Gly or Gly-only residue interaction counts to build QM-cluster models for achieving energy convergence in smaller models, a F-SAPT residue interaction energy-based ranking

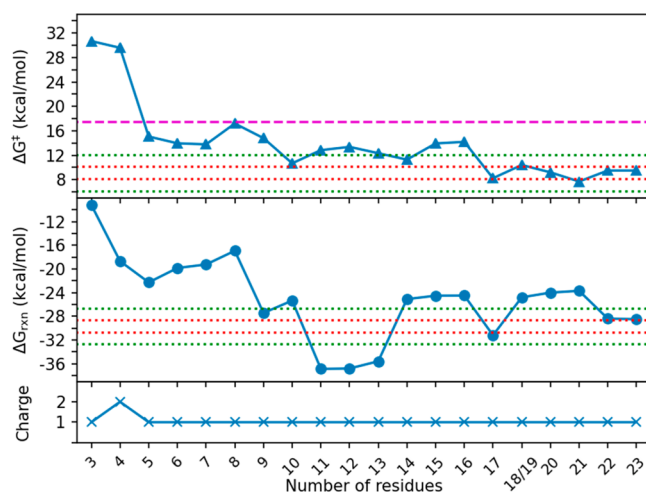




**Figure 5.** Free energies of activation and reaction of the QM-cluster model built based on *probe* ranking of residue fragment interactions with Gly substrate only as the seed. The dashed line in magenta represents the experimental value converted free energy of activation of 17.5 kcal/mol. The red dotted lines show the tight energy convergence range (within  $\pm 1$  kcal/mol of the  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of the maximal model Res27); the green dotted lines show the loose energy convergence range (within  $\pm 3$  kcal/mol of the  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of the maximal model Res27).

scheme was explored. F-SAPT ranks the charged residue fragments Arg175, Arg40, and Asp85 much higher than *probe*-interaction counts with Gly+SAM seed. Arg40 and Asp85 residues interact only with SAM and are not included in the *probe* Gly-only RINRUS models (Table 2). As charged residues, they have relatively large F-SAPT interaction energies. The three charged residues are the first fragments that are added to the F-SAPT-based RINRUS models independent of including the SAM cofactor in the seed. Interestingly, the Gly-only seed has its interaction energy ranked in the order Arg175 > Arg40 > Asp85, while the Gly+SAM seed ranks Arg175 > Asp85 > Arg40 (Table 3). The magnitude of interaction energies of the third-ranked fragment (Asp85; 24.7 kcal/mol in the Gly-only seed and Arg40; −25.8 kcal/mol in the Gly+SAM seed) is comparable to the magnitude of the various uncharged polar residues in the active site such as Asn138 and Ser139. Arg40 positions close to the carboxyl group of methionine on SAM and Asp85 hydrogen bonds to the two hydroxyl groups of the SAM adenosyl group. Neither residue is the closest to the methyl transfer reaction center compared to other residues of the active site RIN.

Based on the Gly interaction energies with active site fragments, 20 models, labeled Res3-fsapt1 to Res23-fsapt1, were investigated (Table 3 and Figure 6). Note that the Res18-fsapt1 and Res19-fsapt1 models are equivalent because the Asn116 main chain ranked 19th is already added to cap the Ala115 main chain in Res18-fsapt1. Immense changes in the kinetic and thermodynamic values manifest in the three smallest-sized models, which is expected.  $\Delta\Delta G^\ddagger$  and  $\Delta\Delta G_{\text{rxn}}$  values are diminished after adding more residues to the model where Gly and the adenosyl group of SAM are stabilized by polar uncharged residues and residues with hydrophobic side chains (Figure 6). However, unsigned  $\Delta\Delta G_{\text{rxn}}$  values are greater than 4 kcal/mol when adding Asn138 in Res9-fsapt1, His142 in Res11-fsapt1, Tyr21 in Res14-fsapt1, Tyr242 in Res17-fsapt1, Ala115 in Res18-fsapt1, and Ser87 in Res22-fsapt1. Among the larger models, only Res17-fsapt1 (Tyr242) has a large  $\Delta\Delta G^\ddagger$  value of −6.0 kcal/mol. All other models have an unsigned

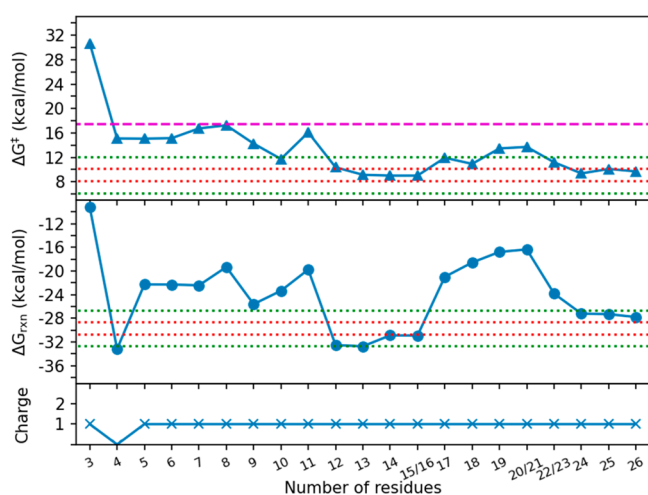


**Figure 6.** Free energy diagram of the 21 different sized models built based on residue ranking using F-SAPT interaction energy with seed Gly only. Symbols on top, middle, and bottom show the free energies of activation for *N*-demethylation, reaction free energies, and charges for each model, respectively. The dashed line in magenta represents the experimental value converted free energy of activation of 17.5 kcal/mol. The red dotted lines show the tight energy convergence range (within  $\pm 1$  kcal/mol of the  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of the maximal model Res27); the green dotted lines show the loose energy convergence range (within  $\pm 3$  kcal/mol of the  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of the maximal model Res27).

$\Delta\Delta G^\ddagger$  of less than 3.0 kcal/mol. In the *probe* models, Tyr21 is first included in the smallest model (Res5, ranked fifth), while Tyr242 and His142 are ranked 12th and 16th. The  $\Delta\Delta G^\ddagger$  and  $\Delta\Delta G_{\text{rxn}}$  values of adding Tyr242 (+0.2 and −1.8 kcal/mol) and His142 (−0.8 and −0.3 kcal/mol) in the *probe* models are much smaller in magnitude than those of the F-SAPT models. Starting from Res18-fsapt1 where Ala115 is included, the adenosyl group of SAM shifts compared to smaller-sized models but aligns well with larger-sized models.  $\Delta\Delta G^\ddagger$  and  $\Delta\Delta G_{\text{rxn}}$  values of Res23-fsapt1 (384 atoms) are less than 0.1 kcal/mol, and  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  values of both models are less than 1.5 kcal/mol compared to those of the maximal *probe* SAM+Gly model Res27 (model we used to calculate F-SAPT interaction energies), which indicates that they are converged.

When SAM+Gly is used as the seed, significant changes are seen again when adding fragments to the smallest models. Unsigned changes in  $\Delta G_{\text{rxn}}$  greater than 3.0 kcal/mol were observed when adding Asn138 in Res9-fsapt2, Trp30 in Res12-fsapt2, Tyr21 in Res17-fsapt2, Tyr242 in Res22-fsapt2, and Ser87 in Res24-fsapt2.  $\Delta\Delta G^\ddagger$  values of Res9-fsapt2 (Asn138), Res11-fsapt2 (His142), Res12-fsapt2 (Trp30), and Res22-fsapt2 (Tyr242) are −3.0, +4.4, −5.7, and +2.9 kcal/mol, respectively (Figure 7). With the F-SAPT fragment ranking scheme, starting from Res24-fsapt2 (363 atoms), the changes in  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of the largest three models are minuscule ( $\Delta G^\ddagger$  ranges from 9.4 to 10.1 kcal/mol and  $\Delta G_{\text{rxn}}$  ranges from −27.8 to −27.2 kcal/mol). The absolute values of  $\Delta\Delta G^\ddagger$  and  $\Delta\Delta G_{\text{rxn}}$  are less than 1 kcal/mol compared to those of Res26-fsapt2 and less than 3 kcal/mol compared to those of Res27, indicating that these three models are converged.

There are six residue fragments that have an absolute interaction energy less than 0.5 kcal/mol with Gly-only seed. Among the six, Leu136 and Val69 are ranked in the bottom four, while Ile34, Asn116, Ala86, and Ser139 are ranked 20th, sixth, eighth, and seventh according to *probe* interaction counts. All of



**Figure 7.** Free energy diagram of the 24 different sized models built based on residue ranking using F-SAPT interaction energy with seed Gly plus SAM. Symbols on top, middle, and bottom show the free energies of activation for *N*-demethylation, reaction free energies, and charges for each model, respectively. The dashed line in magenta represents the experimental value converted free energy of activation of 17.5 kcal/mol. The red dotted lines show the tight energy convergence range (within  $\pm 1$  kcal/mol of the  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of the maximal model **Res27**); the green dotted lines show the loose energy convergence range (within  $\pm 3$  kcal/mol of the  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  of the maximal model **Res27**).

these residues have small effects on the energetics, except Ser139, as we see that when Ser139 is included in **Res7** in the *probe* SAM+Gly seed models,  $\Delta\Delta G^\ddagger = -6.4$  kcal/mol. However, based on the F-SAPT interaction energies with SAM+Gly, Ser139 is ranked 10th and its effects on energetics are small (unsigned  $\Delta\Delta G$  values are less than 3 kcal/mol), which indicates that when Ser139 is added to an already large model, its effect on free energies may be smaller. In the residue ranking of F-SAPT interaction energy with Gly+SAM, only Val69, Asn116, and Ile34 residue fragments have lower interaction energies than 0.5 kcal/mol. Hence, energy convergence is assumed to be achieved in both Gly-only and SAM+Gly F-SAPT based models with a model size of about 360 atoms.

**Larger Basis Set and Density Functional Effects.** The average  $\Delta G^\ddagger$  value of our converged *probe*-based SAM+Gly *RINRUS* models and F-SAPT models (**Res23** to **Res27**, **Res22-fsapt1**, and **Res23-fsapt1** and **Res24-fsapt2** to **Res26-fsapt2**) is 9.5 kcal/mol, while the maximal QM-cluster model from *probe*-based interaction counts using Gly as the only fragment in the seed was an outlier and was discarded. The mean  $\Delta G^\ddagger$  value of our largest/converged models is in poor agreement with experimental measured kinetic data for the methyl transfer reaction in GNMT. However, the main purpose of this investigation is to minimize or eliminate the QM-cluster model size and composition as a source of error. To test the accuracy of GNMT QM-cluster models, a brief set of tests was performed by first employing a larger basis set (6-31+G\*\*) on a set of small *RINRUS*-constructed models (**Res8** to **Res16**; Figure S5 and Table S4). Notable changes in model  $\Delta G^\ddagger$  values are observed ( $\Delta\Delta G^\ddagger$  ranges from  $-0.3$  to  $+2.7$  kcal/mol). However, the computational cost increases immensely (1009 versus 1924 basis functions for **Res8** and 1798 versus 3467 basis functions for **Res16**). Besides testing an increased basis set size, two other density functionals, PBE1PBE<sup>67,68</sup> with GD3BJ and M06-2X<sup>69</sup> without GD3BJ, were also tested using the *probe*-

based **Res27** model (Figure S5) and the default 6-31G(d')/6-31G basis set combination. The  $\Delta G^\ddagger$  values also change notably ( $\Delta\Delta G^\ddagger = +3.1$  and  $+1.5$  kcal/mol, respectively). Considering the massive computational cost and weaker dependence on DFT kinetics and thermodynamics, it is difficult to recommend adopting larger basis sets while testing fragment ranking and selection schemes in computational enzymology. However, determining an appropriate and short list of density functionals for QM-cluster modeling of enzymes is currently under investigation in our laboratories and others.<sup>1,2,70–73</sup>

**Conclusions.** The methyl transfer reaction between SAM and Gly catalyzed by GNMT was examined by using QM-cluster models generated by *RINRUS*. Several residue fragment ranking schemes have been explored, including *probe* interaction counts and F-SAPT interaction energies between residue fragments with Gly only or SAM+Gly. The maximal model **Res27** in which all of the residue fragments that interact with SAM+Gly are included provides  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  values of 9.1 and  $-29.7$  kcal/mol. The  $\Delta G^\ddagger$  and  $\Delta G_{\text{rxn}}$  values of **Res23** to **Res26** differ from that of **Res27** by less than 3 kcal/mol. Hence, convergence in energies starts from the model with 377 atoms (**Res23**) by using the *probe*-based contact count ranking. Small Gly-only *probe*-based models are unreliable for calculating the reaction energetics because the residue fragments that interact with SAM are missing from those models, and the free energies of the models are not converged. F-SAPT ranks charged residues much higher, but to achieve energy convergence, QM-cluster models with more than 360 atoms are still required, whether Gly or Gly +SAM are used as the seed fragments. Although F-SAPT-based QM-cluster models seem to converge kinetic and thermodynamic values with slightly smaller models, an F-SAPT ranking scheme requires expensive front-end calculations. Hence, significantly more calibration of enzyme case studies as well as an exploration of additional residue ranking schemes is essential.

Our larger QM-cluster models show a  $\Delta G^\ddagger$  value much lower than that of the experiment, but we consciously use a somewhat outmoded level of theory to focus on kinetic and thermodynamic convergence with respect to model size. However, studies in our group have shown systematic underprediction of activation energies using the B3LYP-GD3BJ/6-31G(d') level of theory. We have observed increases in  $\Delta G^\ddagger$  values (up to  $+2.7$  kcal/mol) of several small models using the larger basis set 6-31+G\*\* and increases in  $\Delta G^\ddagger$  values (up to  $+3.1$  kcal/mol) of the maximal model with newer generation functionals. Another source of error in our computations may arise from using only QM-cluster models based on the X-ray crystal structure instead of sampling conformations and nonequilibrium structures from molecular dynamics simulations.

Interestingly, the convergence of kinetic and thermodynamic values with respect to QM-cluster model size behaves much differently for the GNMT case study than for COMT. While there is growing precedent that the largest *RINRUS*-constructed model (dubbed the “maximal” model) will reliably emulate the active site environment, even removing one or two fragments from the maximal GNMT QM-cluster models can qualitatively affect activation free energies and/or reaction energies. For very different reasons than COMT, the GNMT case study provides a challenge for benchmarking computational enzymology. We propose that the primary factor is the disparate molecular size of the small glycine substrate and the large SAM cofactor. Residues that interact with the Gly substrate were the focus of the first QM-cluster model study by Velichkova and Himo, but in their models, the SAM cofactor was trimmed to a small fragment.

When incorporating the complete SAM fragment, our *RINRUS*-constructed QM-cluster models based on a *probe*-ranking scheme and using Gly as the seed fragment gave inconsistent results compared to those of our other *RINRUS* model-building schemes. Problems with the Gly-only seed models indicate that both the methionine and adenosyl moiety of the SAM cofactor contribute a great deal to the kinetics and thermodynamics of the GNMT catalysis. Overall, small QM-cluster models cannot provide reliable kinetic or thermodynamic results, further indicating that a systematic approach to approximating the chemical environment of the active site is important for building QM-cluster models of enzymes.

While activation energies of the largest *RINRUS*-built QM-cluster models of GNMT are qualitatively lower than experimental results, the level of theory can be greatly improved when seeking a more effective approach for studying computational enzymology with QM-cluster and QM/MM methods. Currently, several studies are being carried out in our laboratories to calibrate an appropriate electronic structure theory for atomic-level enzyme models.

## ■ ASSOCIATED CONTENT

### SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.jpcb.3c04138>.

Arg175-add-models, basis-sets, fsapt-seed1-models, fsapt-seed2-models, himo-based-models, probe-seed1-models, probe-seed2-models (ZIP)

Text: Large basis set effects. Figure S1: The optimized TS structures of all Himo-based models computed in this study. Figure S2: The changes in free energies of activation and reaction for each model when a new residue/fragment added and forming a larger model. Figure S3: Free energy diagram of the original *probe*-based *RINRUS* Res5 to Res12 models and those with Arg175 added. Figure S4: The optimized reactants, transition states, and products of Res12 and Res12 plus Arg175 models aligned. Figure S5: Free energy diagram of the *RINRUS* Res8 to Res16 models computed with larger basis set and Res27 model computed with different functionals. Table S1. Free energies of activation and reaction for various sized Himo-based models built in this study. Table S2. Residue fragment and seed (Gly plus SAM) interaction map/fingerprint. Table S3. Gibbs free energies of activation and reaction of Arg175 added Res5 to Res12 models. Table S4. Free energies of activation and reaction for *RINRUS* models from Res8 to Res16 run with larger basis set 6-31+G\*\* and Res27 at PBE1PBE with GD3BJ and M06-2X without GD3BJ. Table S5. Thermochemical data of all Himo-based model reported reactants, transition states, and products of the methyl transfer reaction catalyzed by glycine *N*-methyltransferase. Table S6. Thermochemical data of all probe SAM +Gly seed *RINRUS* model reported reactants, transition states, and products of the methyl transfer reaction catalyzed by glycine *N*-methyltransferase. Table S7. Thermochemical data of all Arg175 added *probe* SAM +Gly seed *RINRUS* model (Res5 to Res12 models) reported reactants, transition states, and products of the methyl transfer reaction catalyzed by glycine *N*-methyltransferase. Table S8. Thermochemical data of all *probe* Gly seed *RINRUS* model reported reactants,

transition states, and products of the methyl transfer reaction catalyzed by glycine *N*-methyltransferase. Table S9. Thermochemical data of all Gly seed F-SAPT model reported reactants, transition states, and products of the methyl transfer reaction catalyzed by glycine *N*-methyltransferase. Table S10. Thermochemical data of all SAM+Gly seed F-SAPT model reported reactants, transition states, and products of the methyl transfer reaction catalyzed by glycine *N*-methyltransferase. Table S11. Thermochemical data of all large basis set models and newer functionals Res27 model reported reactants, transition states, and products of the methyl transfer reaction catalyzed by glycine *N*-methyltransferase (PDF)

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### Notes

The authors declare no competing financial interest.

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## ■ REFERENCES

- (1) Sirirak, J.; Lawan, N.; Van der Kamp, M. W.; Harvey, J. N.; Mulholland, A. J. Benchmarking Quantum Mechanical Methods for Calculating Reaction Energies of Reactions Catalyzed by Enzymes. *PeerJ. Phys. Chem.* **2020**, 2, No. e8.
- (2) Wappett, D. A.; Goerigk, L. Toward a Quantum-Chemical Benchmark Set for Enzymatically Catalyzed Reactions: Important Steps and Insights. *J. Phys. Chem. A* **2019**, 123 (32), 7057–7074.
- (3) Cheng, Q.; DeYonker, N. J.; Summers, T. J.; Agbaglo, D. A.; Suhagia, T.; Palma, M. A. GitHub - Natedey/RINRUS: Residue Interaction Network Residue Selector (RINRUS) Public Release. <https://github.com/natedey/RINRUS> (accessed September 14, 2022).
- (4) Amitai, G.; Shemesh, A.; Sitbon, E.; Shklar, M.; Netanel, D.; Venger, I.; Pietrokovski, S. Network Analysis of Protein Structures Identifies Functional Residues. *J. Mol. Biol.* **2004**, 344 (4), 1135–1146.
- (5) Doncheva, N. T.; Klein, K.; Domingues, F. S.; Albrecht, M. Analyzing and Visualizing Residue Networks of Protein Structures. *Trends Biochem. Sci.* **2011**, 36 (4), 179–182.
- (6) Piovesan, D.; Minervini, G.; Tosatto, S. C. E. The RING 2.0 Web Server for High Quality Residue Interaction Networks. *Nucleic Acids Res.* **2016**, 44 (W1), W367–W374.



- (7) Parrish, R. M.; Sherrill, C. D. Spatial Assignment of Symmetry Adapted Perturbation Theory Interaction Energy Components: The Atomic SAPT Partition. *J. Chem. Phys.* **2014**, *141* (4), No. 044115.
- (8) Parrish, R. M.; Parker, T. M.; Sherrill, C. D. Chemical Assignment of Symmetry-Adapted Perturbation Theory Interaction Energy Components: The Functional-Group SAPT Partition. *J. Chem. Theory Comput.* **2014**, *10* (10), 4417–4431.
- (9) Parrish, R. M.; Gonthier, J. F.; Corminboeuf, C.; Sherrill, C. D. Communication: Practical Intramolecular Symmetry Adapted Perturbation Theory via Hartree-Fock Embedding. *J. Chem. Phys.* **2015**, *143* (5), No. 051103.
- (10) Summers, T. J.; Hemmati, R.; Miller, J. E.; Agbaglo, D. A.; Cheng, Q.; DeYonker, N. J. Evaluating the Active Site-Substrate Interplay between x-Ray Crystal Structure and Molecular Dynamics in Chorismate Mutase. *J. Chem. Phys.* **2023**, *158* (6), No. 065101.
- (11) Parker, T. M.; Burns, L. A.; Parrish, R. M.; Ryno, A. G.; Sherrill, C. D. Levels of Symmetry Adapted Perturbation Theory (SAPT). I. Efficiency and Performance for Interaction Energies. *J. Chem. Phys.* **2014**, *140* (9), No. 094106.
- (12) Yeo, E.-J.; Briggs, W. T.; Wagner, C. Inhibition of Glycine N-Methyltransferase by 5-Methyltetrahydrofolate Pentaglutamate. *J. Biol. Chem.* **1999**, *274* (53), 37559–37564.
- (13) Yen, C.-H.; Lin, Y.-T.; Chen, H.-L.; Chen, S.-Y.; Chen, Y.-M. A. The Multi-Functional Roles of GNMT in Toxicology and Cancer. *Toxicol. Appl. Pharmacol.* **2013**, *266* (1), 67–75.
- (14) Wagner, C.; Briggs, W. T.; Cook, R. J. Inhibition of Glycine N-Methyltransferase Activity by Folate Derivatives: Implications for Regulation of Methyl Group Metabolism. *Biochem. Biophys. Res. Commun.* **1985**, *127* (3), 746–752.
- (15) Luka, Z.; Mudd, S. H.; Wagner, C. Glycine N-Methyltransferase and Regulation of S-Adenosylmethionine Levels. *J. Biol. Chem.* **2009**, *284* (34), 22507–22511.
- (16) Rutherford, K.; Le Trong, I.; Stenkamp, R. E.; Parson, W. W. Crystal Structures of Human 108V and 108M Catechol O-Methyltransferase. *J. Mol. Biol.* **2008**, *380* (1), 120–130.
- (17) Kulik, H. J. Large-Scale QM/MM Free Energy Simulations of Enzyme Catalysis Reveal the Influence of Charge Transfer. *Phys. Chem. Chem. Phys.* **2018**, *20* (31), 20650–20660.
- (18) Kulik, H. J.; Zhang, J.; Klinman, J. P.; Martinez, T. J. How Large Should the QM Region Be in QM/MM Calculations? The Case of Catechol O-Methyltransferase. *J. Phys. Chem. B* **2016**, *120* (44), 11381–11394.
- (19) Summers, T. J.; Cheng, Q.; Palma, M. A.; Pham, D.-T.; Kelso, D. K.; Webster, C. E.; DeYonker, N. J. Cheminformatic Quantum Mechanical Enzyme Model Design: A Catechol-O-Methyltransferase Case Study. *Biophys. J.* **2021**, *120* (17), 3577–3587.
- (20) Kuhn, B.; Kollman, P. A. QM-FE and Molecular Dynamics Calculations on Catechol O-Methyltransferase: Free Energy of Activation in the Enzyme and in Aqueous Solution and Regioselectivity of the Enzyme-Catalyzed Reaction. *J. Am. Chem. Soc.* **2000**, *122* (11), 2586–2596.
- (21) Ruggiero, G. D.; Williams, I. H.; Roca, M.; Moliner, V.; Tuñón, I. QM/MM Determination of Kinetic Isotope Effects for COMT-Catalyzed Methyl Transfer Does Not Support Compression Hypothesis. *J. Am. Chem. Soc.* **2004**, *126* (28), 8634–8635.
- (22) Lameira, J.; Bora, R. P.; Chu, Z. T.; Warshel, A. Methyltransferases Do Not Work by Compression, Cratic, or Desolvation Effects, but by Electrostatic Preorganization. *Proteins Struct. Funct. Bioinforma.* **2015**, *83* (2), 318–330.
- (23) Rod, T. H.; Ryde, U. Quantum Mechanical Free Energy Barrier for an Enzymatic Reaction. *Phys. Rev. Lett.* **2005**, *94* (13), No. 138302.
- (24) Rod, T. H.; Ryde, U. Accurate QM/MM Free Energy Calculations of Enzyme Reactions: Methylation by Catechol O-Methyltransferase. *J. Chem. Theory Comput.* **2005**, *1* (6), 1240–1251.
- (25) Heady, J. E.; Kerr, S. J. Purification and Characterization of Glycine N-Methyltransferase. *J. Biol. Chem.* **1973**, *248* (1), 69–72.
- (26) Ogawa, H.; Fujioka, M. Induction of Rat Liver Glycine Methyltransferase by High Methionine Diet. *Biochem. Biophys. Res. Commun.* **1982**, *108* (1), 227–232.
- (27) Cook, R. J.; Horne, D. W.; Wagner, C. Effect of Dietary Methyl Group Deficiency on One-Carbon Metabolism in Rats. *J. Nutr.* **1989**, *119* (4), 612–617.
- (28) Balaghi, M.; Horne, D. W.; Wagner, C. Hepatic One-Carbon Metabolism in Early Folate Deficiency in Rats. *Biochem. J.* **1993**, *291* (1), 145–149.
- (29) Zhang, J.; Klinman, J. P. Convergent Mechanistic Features between the Structurally Diverse N- and O-Methyltransferases: Glycine N-Methyltransferase and Catechol O-Methyltransferase. *J. Am. Chem. Soc.* **2016**, *138* (29), 9158–9165.
- (30) Huang, Y.; Komoto, J.; Konishi, K.; Takata, Y.; Ogawa, H.; Gomi, T.; Fujioka, M.; Takusagawa, F. Mechanisms for Auto-Inhibition and Forced Product Release in Glycine N-Methyltransferase: Crystal Structures of Wild-Type, Mutant R175K and S-Adenosylhomocysteine-Bound R175K Enzymes. *J. Mol. Biol.* **2000**, *298* (1), 149–162.
- (31) Konishi, K.; Fujioka, M. Chemical Modification of a Functional Arginine Residue of Rat Liver Glycine Methyltransferase. *Biochemistry* **1987**, *26* (25), 8496–8502.
- (32) Ogawa, H.; Gomi, T.; Fujioka, M. Mammalian Glycine N-Methyltransferases. Comparative Kinetic and Structural Properties of the Enzymes from Human, Rat, Rabbit and Pig Livers. *Comp. Biochem. Physiol. Part B Comp. Biochem.* **1993**, *106* (3), 601–611.
- (33) Takata, Y.; Huang, Y.; Komoto, J.; Yamada, T.; Konishi, K.; Ogawa, H.; Gomi, T.; Fujioka, M.; Takusagawa, F. Catalytic Mechanism of Glycine N-Methyltransferase. *Biochemistry* **2003**, *42* (28), 8394–8402.
- (34) Fu, Z.; Hu, Y.; Konishi, K.; Takata, Y.; Ogawa, H.; Gomi, T.; Fujioka, M.; Takusagawa, F. Crystal Structure of Glycine N-Methyltransferase from Rat Liver. *Biochemistry* **1996**, *35* (37), 11985–11993.
- (35) Velichkova, P.; Himov, F. Methyl Transfer in Glycine N-Methyltransferase. A Theoretical Study. *J. Phys. Chem. B* **2005**, *109* (16), 8216–8219.
- (36) Lee, C.; Yang, W.; Parr, R. G. Development of the Colle-Salvetti Correlation-Energy Formula into a Functional of the Electron Density. *Phys. Rev. B* **1988**, *37* (2), 785–789.
- (37) Becke, A. D. Density-Functional Exchange-Energy Approximation with Correct Asymptotic Behavior. *Phys. Rev. A* **1988**, *38* (6), 3098–3100.
- (38) Becke, A. D. Density-functional Thermochemistry. I. The Effect of the Exchange-only Gradient Correction. *J. Chem. Phys.* **1992**, *96* (3), 2155–2160.
- (39) Becke, A. D. Density-functional Thermochemistry. II. The Effect of the Perdew–Wang Generalized-gradient Correlation Correction. *J. Chem. Phys.* **1992**, *97* (12), 9173–9177.
- (40) Becke, A. D. Density-functional Thermochemistry. III. The Role of Exact Exchange. *J. Chem. Phys.* **1993**, *98* (7), 5648–5652.
- (41) Cammi, R.; Mennucci, B.; Tomasi, J. Second-Order Møller–Plesset Analytical Derivatives for the Polarizable Continuum Model Using the Relaxed Density Approach. *J. Phys. Chem. A* **1999**, *103* (45), 9100–9108.
- (42) Cammi, R.; Mennucci, B.; Tomasi, J. Fast Evaluation of Geometries and Properties of Excited Molecules in Solution: A Tamm–Dancoff Model with Application to 4-Dimethylaminobenzonitrile. *J. Phys. Chem. A* **2000**, *104* (23), 5631–5637.
- (43) Cossi, M.; Rega, N.; Scalmani, G.; Barone, V. Polarizable Dielectric Model of Solvation with Inclusion of Charge Penetration Effects. *J. Chem. Phys.* **2001**, *114* (13), 5691–5701.
- (44) Cossi, M.; Scalmani, G.; Rega, N.; Barone, V. New Developments in the Polarizable Continuum Model for Quantum Mechanical and Classical Calculations on Molecules in Solution. *J. Chem. Phys.* **2002**, *117* (1), 43–54.
- (45) Hwang, J. K.; King, G.; Creighton, S.; Warshel, A. Simulation of Free Energy Relationships and Dynamics of SN2 Reactions in Aqueous Solution. *J. Am. Chem. Soc.* **1988**, *110* (16), 5297–5311.
- (46) Świderek, K.; Tuñón, I.; Williams, I. H.; Moliner, V. Insights on the Origin of Catalysis on Glycine N-Methyltransferase from Computational Modeling. *J. Am. Chem. Soc.* **2018**, *140* (12), 4327–4334.

- (47) Word, J. M.; Lovell, S. C.; Richardson, J. S.; Richardson, D. C. Asparagine and Glutamine: Using Hydrogen Atom Contacts in the Choice of Side-Chain Amide Orientation 1 Edited by J. Thornton. *J. Mol. Biol.* **1999**, *285* (4), 1735–1747.
- (48) Word, J. M.; Lovell, S. C.; LaBean, T. H.; Taylor, H. C.; Zalis, M. E.; Presley, B. K.; Richardson, J. S.; Richardson, D. C. Visualizing and Quantifying Molecular Goodness-of-Fit: Small-Probe Contact Dots with Explicit Hydrogen Atoms 1 Edited by J. Thornton. *J. Mol. Biol.* **1999**, *285* (4), 1711–1733.
- (49) Smith, D. G. A.; Burns, L. A.; Simmonett, A. C.; Parrish, R. M.; Schieber, M. C.; Galvelis, R.; Kraus, P.; Kruse, H.; Di Remigio, R.; Alenaizan, A.; et al. PSI4 1.4: Open-Source Software for High-Throughput Quantum Chemistry. *J. Chem. Phys.* **2020**, *152* (18), No. 184108.
- (50) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Petersson, G. A.; Nakatsuji, H.; et al. *Gaussian 16*, revision B.01; Gaussian Inc., 2016.
- (51) Hariharan, P. C.; Pople, J. A. The Influence of Polarization Functions on Molecular Orbital Hydrogenation Energies. *Theor. Chim. Acta* **1973**, *28* (3), 213–222.
- (52) Petersson, G. A.; Al-Laham, M. A. A Complete Basis Set Model Chemistry. II. Open-Shell Systems and the Total Energies of the First-Row Atoms. *J. Chem. Phys.* **1991**, *94* (9), 6081–6090.
- (53) Foresman, J. B.; Frisch, A. *Exploring Chemistry with Electronic Structure Methods*, 2nd ed.; Gaussian Inc.: Pittsburgh, PA, 1996; pp 266, 278–283.
- (54) Hehre, W. J.; Ditchfield, R.; Pople, J. A. Self-Consistent Molecular Orbital Methods. XII. Further Extensions of Gaussian-Type Basis Sets for Use in Molecular Orbital Studies of Organic Molecules. *J. Chem. Phys.* **1972**, *56* (5), 2257–2261.
- (55) Grimme, S.; Ehrlich, S.; Goerigk, L. Effect of the Damping Function in Dispersion Corrected Density Functional Theory. *J. Comput. Chem.* **2011**, *32* (7), 1456–1465.
- (56) Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H. A Consistent and Accurate Ab Initio Parametrization of Density Functional Dispersion Correction (DFT-D) for the 94 Elements H–Pu. *J. Chem. Phys.* **2010**, *132* (15), No. 154104.
- (57) Barone, V.; Cossi, M. Quantum Calculation of Molecular Energies and Energy Gradients in Solution by a Conductor Solvent Model. *J. Phys. Chem. A* **1998**, *102* (11), 1995–2001.
- (58) Cossi, M.; Rega, N.; Scalmani, G.; Barone, V. Energies, Structures, and Electronic Properties of Molecules in Solution with the C-PCM Solvation Model. *J. Comput. Chem.* **2003**, *24* (6), 669–681.
- (59) Siegbahn, P. E. M.; Blomberg, M. R. A. Transition-Metal Systems in Biochemistry Studied by High-Accuracy Quantum Chemical Methods. *Chem. Rev.* **2000**, *100* (2), 421–438.
- (60) Blomberg, M. R. A.; Borowski, T.; Himio, F.; Liao, R.-Z.; Siegbahn, P. E. M. Quantum Chemical Studies of Mechanisms for Metalloenzymes. *Chem. Rev.* **2014**, *114* (7), 3601–3658.
- (61) DeYonker, N. J.; Webster, C. E. Phosphoryl Transfers of the Phospholipase D Superfamily: A Quantum Mechanical Theoretical Study. *J. Am. Chem. Soc.* **2013**, *135* (37), 13764–13774.
- (62) Summers, T. J.; Cheng, Q.; DeYonker, N. J. A Transition State “Trapped”? QM-Cluster Models of Engineered Threonyl-TRNA Synthetase. *Org. Biomol. Chem.* **2018**, *16* (22), 4090–4100.
- (63) DeYonker, N. J.; Webster, C. E. A Theoretical Study of Phosphoryl Transfers of Tyrosyl-DNA Phosphodiesterase I (Tdp1) and the Possibility of a “Dead-End” Phosphohistidine Intermediate. *Biochemistry* **2015**, *54* (27), 4236–4247.
- (64) Cheng, Q.; DeYonker, N. J. Acylation and Deacylation Mechanism and Kinetics of Penicillin G Reaction with Streptomyces R61 DD-peptidase. *J. Comput. Chem.* **2020**, *41* (18), 1685–1697.
- (65) Fukui, K. Formulation of the Reaction Coordinate. *J. Phys. Chem.* **1970**, *74* (23), 4161–4163.
- (66) Fukui, K. The Path of Chemical Reactions - the IRC Approach. *Acc. Chem. Res.* **1981**, *14* (12), 363–368.
- (67) Adamo, C.; Barone, V. Toward Reliable Density Functional Methods without Adjustable Parameters: The PBE0 Model. *J. Chem. Phys.* **1999**, *110* (13), 6158–6170.
- (68) Ernzerhof, M.; Scuseria, G. E. Assessment of the Perdew–Burke–Ernzerhof Exchange–Correlation Functional. *J. Chem. Phys.* **1999**, *110* (11), 5029–5036.
- (69) Zhao, Y.; Truhlar, D. G. The M06 Suite of Density Functionals for Main Group Thermochemistry, Thermochemical Kinetics, Non-covalent Interactions, Excited States, and Transition Elements: Two New Functionals and Systematic Testing of Four M06-Class Functionals and 12 Other Function. *Theor. Chem. Acc.* **2008**, *120* (1–3), 215–241.
- (70) Wappett, D. A.; Goerigk, L. A Guide to Benchmarking Enzymatically Catalysed Reactions: The Importance of Accurate Reference Energies and the Chemical Environment. *Theor. Chem. Acc.* **2021**, *140* (6), 68.
- (71) Ahmadi, S.; Barrios Herrera, L.; Chehelamirani, M.; Hostaš, J.; Jalife, S.; Salahub, D. R. Multiscale Modeling of Enzymes: QM-Cluster, QM/MM, and QM/MM/MD: A Tutorial Review. *Int. J. Quantum Chem.* **2018**, *118* (9), No. e25558.
- (72) Wappett, D. A.; Goerigk, L. Benchmarking Density Functional Theory Methods for Metalloenzyme Reactions: The Introduction of the MME55 Set. *ChemRxiv*, DOI: 10.26434/chemrxiv-2023-w341f.
- (73) Glanowski, M.; Kachhap, S.; Borowski, T.; Szaleniec, M. *Model Setup and Procedures for Prediction of Enzyme Reaction Kinetics with QM-Only and QM:MM Approaches* **2022**, 2385, 175–236.