



# Extensive H<sub>2</sub>O degassing in deeply erupted submarine glasses inferred from Samoan melt inclusions: The EM2 mantle source is damp, not dry

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## ABSTRACT

Submarine glasses erupted at intraplate volcanic hotspot settings sampling enriched mantle (EM)—characterized by high <sup>87</sup>Sr/<sup>86</sup>Sr—exhibit lower H<sub>2</sub>O/Ce than glasses representing less enriched mantle domains, leading to the interpretation that the EM mantle is H<sub>2</sub>O-poor (“dry”). We test whether low H<sub>2</sub>O/Ce observed in pillow glasses of EM lavas resulted from degassing of higher H<sub>2</sub>O/Ce primary melts by measuring H<sub>2</sub>O/Ce and <sup>87</sup>Sr/<sup>86</sup>Sr in olivine-hosted melt inclusions from two deeply-erupted (3950 and 2190 m below sea level (mbsl)) Samoan submarine lavas from Vailulu'u and Malumalu seamounts. Vailulu'u (H<sub>2</sub>O/Ce = 161–275) and Malumalu (149–232) melt inclusions have an average H<sub>2</sub>O/Ce (197 ± 58 2SD, N = 15) that is nearly twice as high as H<sub>2</sub>O/Ce in pillow glasses from these two seamounts (average H<sub>2</sub>O/Ce = 106 ± 51, N = 65), and comparable to pillow glasses from non-EM hotspots. We show that lower H<sub>2</sub>O/Ce in submarine Samoan glasses compared to melt inclusions results from greater closed-system degassing, and concomitant loss of H<sub>2</sub>O, because EM melts have higher initial concentrations of CO<sub>2</sub>. We show that the lower H<sub>2</sub>O/Ce in global EM pillow glasses compared to non-EM pillow glasses can be modeled to be the result of more extensive degassing of H<sub>2</sub>O in EM melts, which owes to higher CO<sub>2</sub> in primary melts (20,000–90,000 ppm) of EM sources compared to non-EM melts (300–50,000 ppm CO<sub>2</sub>). Instead of originating from a dry mantle, we conclude that EM lavas derive from a damp mantle, but EM melts lose more H<sub>2</sub>O by degassing than non-EM melts.

## 1. Introduction

Although volatiles are a minor component in silicate melts, they play an outsize role on the properties and behavior of magmatic systems. Hydrogen and carbon affect the depth and extent of melting, the composition of melts, how melts evolve, and also the rheology of the mantle (e.g., Asimow and Langmuir, 2003; Gaetani and Grove, 1998; Hirschmann, 2006; Hirth and Kohlstedt, 1996; Hirth and Kohlstedt, 2003; Portnyagin et al., 2007). Therefore, connecting volatile element compositions to mantle domains characterized by particular isotopic compositions, such as enriched mantle (EM), can provide additional insights into mantle heterogeneities. A variety of geochemical “flavors” have been identified in the mantle via geochemical taxonomy of ocean island basalts (OIB) (e.g., Hart et al., 1986; Hart, 1988; Hofmann and White, 1982; Zindler and Hart, 1986), and the various mantle species are broadly encompassed by several canonical mantle endmembers that

include HIMU (high  $\mu = ^{238}\text{U}/^{204}\text{Pb}$ ), EM1 (enriched mantle 1), EM2 (enriched mantle 2), and DM (depleted mantle). Samoan lavas have unique geochemical signatures that extend to Enriched Mantle 2 (EM2) isotopic compositions (e.g., high <sup>87</sup>Sr/<sup>86</sup>Sr, intermediate <sup>206</sup>Pb/<sup>204</sup>Pb, and low <sup>143</sup>Nd/<sup>144</sup>Nd). The EM2 mantle source has been suggested to be derived from recycled continental materials (e.g., White and Hofmann, 1982; White and Duncan, 1996; Jackson and Macdonald, 2022). Therefore, characterization of Samoan EM2 lavas can provide insights into the role that deep continental crust subduction plays in controlling the volatile budgets of the mantle.

The ratios H<sub>2</sub>O/La or H<sub>2</sub>O/Ce measured in submarine glasses are commonly used to evaluate H<sub>2</sub>O enrichment or depletion in the mantle because H<sub>2</sub>O, La, and Ce have similar degrees of incompatibility (e.g., Michael, 1995). Thus, their ratios are not significantly affected by partial melting and crystal fractionation. The EM2 mantle source has been interpreted to be ‘dry’ because the H<sub>2</sub>O/Ce (or H<sub>2</sub>O/La) is lower at

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higher  $^{87}\text{Sr}/^{86}\text{Sr}$  for Samoan glasses (Workman et al., 2006). Workman et al. (2006) found that  $\text{H}_2\text{O}/\text{La}$  anticorrelates with  $^{87}\text{Sr}/^{86}\text{Sr}$ : the more geochemically enriched (higher  $^{87}\text{Sr}/^{86}\text{Sr}$ ) Samoan glasses have lower  $\text{H}_2\text{O}/\text{La}$  than more geochemically depleted Samoan glasses; the high  $\text{H}_2\text{O}/\text{La}$  MORB mantle endmember anchors the low  $^{87}\text{Sr}/^{86}\text{Sr}$  portion of the array. Workman et al. (2006) explained this relationship between  $^{87}\text{Sr}/^{86}\text{Sr}$  and  $\text{H}_2\text{O}/\text{La}$  as the result of diffusive  $\text{H}_2\text{O}$  loss from EM2 reservoirs during long-term storage in a dry and depleted mantle, resulting in a "dry" EM2 mantle domain that has low  $\text{H}_2\text{O}/\text{La}$  and low  $\text{H}_2\text{O}$ . Similarly, Dixon et al. (2002) found that high  $^{87}\text{Sr}/^{86}\text{Sr}$  basalts erupted in MOR settings had low  $\text{H}_2\text{O}/\text{Ce}$ . However, unlike Workman et al. (2006), Dixon et al. (2002) suggested that the dry nature of the EM reservoir was due to dehydration of sediments during subduction, resulting in low  $\text{H}_2\text{O}/\text{Ce}$  signatures in subducted, dehydrated sediments, which ultimately contribute to enriched mantle domains. Buzimis and Peslier (2015) offered further discussion regarding the origin of apparently dry EM2 domains and argued that the low  $\text{H}_2\text{O}/\text{Ce}$  in EM lavas is the result of recycling of pyroxenite-bearing oceanic lithosphere. The pyroxenites they examined have low  $\text{H}_2\text{O}/\text{Ce}$  and high  $\text{H}_2\text{O}$ , a result of preferential partitioning of Ce into clinopyroxene compared to  $\text{H}_2\text{O}$ .

Additional work highlights the low  $\text{H}_2\text{O}/\text{Ce}$  in enriched mantle domains (Kendrick et al., 2017). In particular, classical EM1 and EM2 oceanic hotspots—Pitcairn and Societies, respectively—exhibit negative correlations between  $^{87}\text{Sr}/^{86}\text{Sr}$  and  $\text{H}_2\text{O}/\text{Ce}$  in submarine glasses (Kendrick et al., 2014). Thus, low  $\text{H}_2\text{O}/\text{Ce}$  appears to be a defining feature of EM lavas, and constraining the origin of this geochemical signature is critical for understanding the origin of the EM mantle and/or the petrogenesis of EM lavas.

Existing models for the origin of low  $\text{H}_2\text{O}/\text{La}$  and  $\text{H}_2\text{O}/\text{Ce}$  in EM-flavored OIB pillow glasses quenched in deep submarine environments are predicated on the assumption that the low  $\text{H}_2\text{O}/\text{Ce}$  is a source feature and the EM primary melts have low  $\text{H}_2\text{O}/\text{Ce}$  inherited from the mantle source (Dixon et al., 2002; Dixon and Clague, 2001; Wallace, 2002; Workman et al., 2006; Kendrick et al., 2014). This interpretation assumes that magmatic degassing follows open system behavior, a mechanism that results in significant loss of  $\text{CO}_2$ , but not  $\text{H}_2\text{O}$ , from the melt by degassing: the  $\text{H}_2\text{O}/\text{Ce}$  in deeply erupted glasses (i.e.,  $\geq 0.1$  kbar; Workman et al., 2006) resulting from this process are similar to the mantle source despite having degassed most of their magmatic  $\text{CO}_2$ .

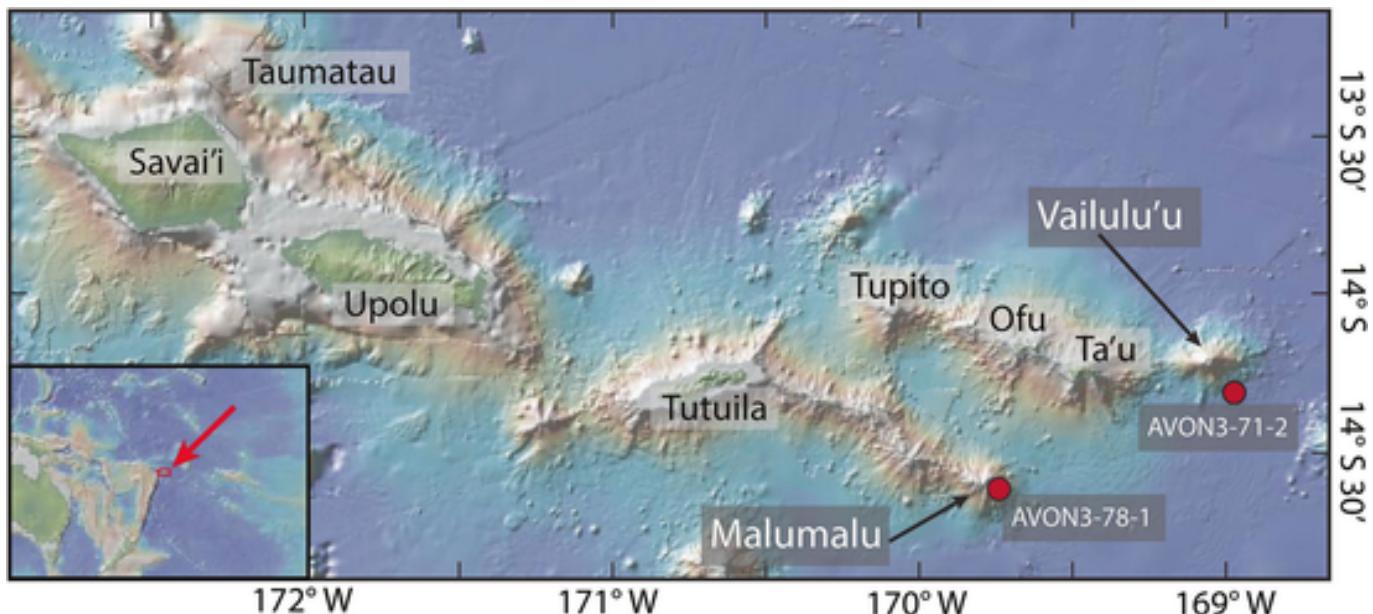
In this study, we test this assumption using combined  $\text{H}_2\text{O}/\text{Ce}$  and  $^{87}\text{Sr}/^{86}\text{Sr}$  measurements in individual olivine-hosted melt inclusions isolated from two well-characterized submarine Samoan basalts that were characterized by Workman et al. (2006). We present volatile ( $\text{H}_2\text{O}$ ,  $\text{CO}_2$ , Cl, F, S), major, and trace element concentrations and  $^{87}\text{Sr}/^{86}\text{Sr}$  on Samoan melt inclusions from both samples—AVON3-78-1 and AVON3-71-2—which were obtained from Malumalu and Vailulu'u seamounts, respectively. Malumalu lavas reach  $^{87}\text{Sr}/^{86}\text{Sr}$  of up to 0.708901, which is the most extreme EM2 composition that we explore in this study. In contrast, Vailulu'u lavas have  $^{87}\text{Sr}/^{86}\text{Sr}$  that range from 0.705352 to 0.706720, and thus represent a less enriched composition at the Samoan hotspot. Previous  $^{87}\text{Sr}/^{86}\text{Sr}$  analyses of Samoan melt inclusions demonstrated extreme heterogeneity (0.70434 to 0.70926) that has been verified using both *in situ* LA-ICP-MS approach (laser-ablation inductively coupled plasma mass spectrometry; Jackson and Hart, 2006) and an approach involving wet chemistry followed by TIMS (thermal ionization mass spectrometry; Reinhard et al., 2018). Critically for this study, the melt inclusions from the two lavas examined here span nearly the same range of  $^{87}\text{Sr}/^{86}\text{Sr}$  (0.704858–0.709225) as identified in prior melt inclusion studies (Jackson and Hart, 2006; Reinhard et al., 2018) and have similar trace element characteristics as the Samoan pillow glasses examined for  $\text{H}_2\text{O}/\text{La}$  and  $^{87}\text{Sr}/^{86}\text{Sr}$  by Workman et al. (2006) (0.704521–0.708901). We interpret this to indicate that the melt inclusions thus sample the same mantle sources as the pillow glasses studied by Workman et al. (2006).

However, Samoan melt inclusions in this study exhibit two key differences with the Samoan submarine glasses reported in Workman et al. (2006): 1) the melt inclusions have higher  $\text{H}_2\text{O}/\text{Ce}$  than Samoan submarine glasses and 2) the inclusions exhibit no relationship between  $\text{H}_2\text{O}/\text{Ce}$  and  $^{87}\text{Sr}/^{86}\text{Sr}$ , which contrasts with the negative correlation between  $^{87}\text{Sr}/^{86}\text{Sr}$  and  $\text{H}_2\text{O}/\text{Ce}$  observed in Samoan pillow glasses. We model the difference in  $\text{H}_2\text{O}/\text{Ce}$  between melt inclusions and pillow glasses as being a result of closed-system degassing, a mechanism that results in greatly diminished  $\text{CO}_2$  and  $\text{H}_2\text{O}$  concentrations in the melt (in contrast to open-system degassing, which results in greatly diminished  $\text{CO}_2$ , but not  $\text{H}_2\text{O}$ , melt concentrations). During closed-system degassing, also called equilibrium or batch equilibrium degassing (e.g., Gerlach and Taylor, 1990), the  $\text{CO}_2$  is exsolved as a gas phase, but remains in isotopic equilibrium with the melt. In open-system degassing, also called fractional equilibrium degassing or Rayleigh distillation, an infinitesimally small quantity of the equilibrium gas phase is exsolved and escapes from the melt, thus changing the composition of the system. This results in the isotopic composition of the total sum of the gas phase being out of equilibrium with the melt, and the  $\text{CO}_2$  is stripped off from the melt more efficiently than in closed-system degassing. In our closed-system degassing model, the melt inclusions experience less closed-system degassing compared to submarine glasses because of deep entrapment of the melt inclusions in growing olivine crystals within magma chambers. Melts that were not trapped continued to degas both  $\text{CO}_2$  and  $\text{H}_2\text{O}$  up until eruption and quenching on the seafloor, resulting in lower  $\text{H}_2\text{O}/\text{Ce}$  in the submarine glasses compared to the deeply entrapped melt inclusions.

We also present a model suggesting that, across global oceanic hotspots, primary melts with higher  $^{87}\text{Sr}/^{86}\text{Sr}$  (e.g., Samoan and Societies EM glasses, Pitcairn EM1 glasses) have higher primary melt  $\text{CO}_2$  than lower  $^{87}\text{Sr}/^{86}\text{Sr}$  melts (e.g., Foundation, Hawai'i (Lō'ihi), and Easter hotspots). Higher  $\text{CO}_2$  in high  $^{87}\text{Sr}/^{86}\text{Sr}$  (EM) melts results in greater degassing of both  $\text{CO}_2$  and  $\text{H}_2\text{O}$  (and a greater reduction in the  $\text{H}_2\text{O}/\text{Ce}$  ratio) compared to low  $^{87}\text{Sr}/^{86}\text{Sr}$  melts, which explains the negative global correlation between  $\text{H}_2\text{O}/\text{Ce}$  and  $^{87}\text{Sr}/^{86}\text{Sr}$  in OIB glasses. This new result suggests that submarine glasses sourcing EM domains do not provide reliable records for evaluating the  $\text{H}_2\text{O}$  and  $\text{H}_2\text{O}/\text{Ce}$  in the mantle source. In melt inclusions, degassing is arrested at great depth, making it possible to identify  $\text{H}_2\text{O}/\text{Ce}$  ratios that are closer to values in the primary melt.

## 2. Methods

Submarine pillow basalt samples in this study are from two Samoan seamounts (Fig. 1) that are young ( $< 8$  ka; Sims et al., 2008) and associated with visually fresh lavas: Malumalu (sample AVON3-78-1; 2260–2190 mbsl) and Vailulu'u (sample AVON3-71-2; 4420–3950 mbsl) are olivine-rich lavas with glassy chill margins on the pillow rims. Olivine-hosted melt inclusions from Vailulu'u and Malumalu seamounts in this study were rehomogenized on a heating stage (see supplement for methods), and these rehomogenized inclusions were then characterized for  $^{87}\text{Sr}/^{86}\text{Sr}$  (15 melt inclusions total),  $^{143}\text{Nd}/^{144}\text{Nd}$  (only three melt inclusions), stable hydrogen isotope ratios (δD), and major, trace and volatile element concentrations (Table S1). This supplement also includes methodological details for the glass major element (electron probe microanalyzer (EPMA)), volatile ( $\text{H}_2\text{O}$ ,  $\text{CO}_2$ , F, Cl, S) (secondary ion mass spectrometry (SIMS)), hydrogen isotope (SIMS), and trace element (LA-ICP-MS) analyses and the host olivine major element (EPMA) analyses. Strontium separation chemistry and  $^{87}\text{Sr}/^{86}\text{Sr}$  analyses by TIMS used in this study follow the same procedure as described in Anderson et al. (2021). The supplement includes details of Nd separation chemistry for melt inclusions, and  $^{143}\text{Nd}/^{144}\text{Nd}$  analyses by TIMS at the University of California, Santa Barbara Isotope Geochemistry Facility. Only three melt inclusions (all from the Vailulu'u Seamount sample) in this study had sufficient Nd for the isotopic analysis.



**Fig. 1.** Map of the Samoan Islands. Red dots indicate the location of AVON3 cruise dredge sites 71 and 78, from the flanks of Vailulu'u and Malumalu, respectively. Map created using GeoMapApp ([www.geomapapp.org](http://www.geomapapp.org); Ryan et al., 2009). (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

The results of in situ analyses (of  $\delta D$ , and major, trace and volatile element concentrations) of geologic reference materials (ALV519-4-1, ALV1833-1, GL07 D52-5, and BCR-2) and values from the literature for these geologic reference materials can be found in Table S2, and  $^{87}\text{Sr}/^{86}\text{Sr}$  and  $^{143}\text{Nd}/^{144}\text{Nd}$  analyses of BCR-2 can be found in Tables S6 and S8. Our measurement of  $\text{H}_2\text{O}$  concentrations in ALV519-4-1 and ALV1833-1 are 1.6% and 1.2%, respectively, different from our preferred values for  $\text{H}_2\text{O}$  in ALV519-4-1 (Bryan and Moore, 1977) and ALV1833-1 (Kumamoto et al., 2017). Our measurement of Ce in ALV519-4-1 differs from Gale et al. (2013) by  $\sim 3.4\%$ .

Melt inclusion major, trace, and volatile element concentrations were corrected for post-entrapment crystallization (PEC) by addition or subtraction of equilibrium olivine in 0.1% increments until the major element composition of the inclusion is in equilibrium with the host olivine forsterite (Fo) content (e.g., Danyushevsky et al., 2000). In the calculation, two assumptions are made: (i) the olivine-melt  $\text{Fe}/\text{Mg}$   $K_d = 0.30$  (Roeder and Emslie, 1970; Ford et al., 1983;  $K_d = (\text{Fe}^{2+}/\text{Mg})_{\text{olivine}}/(\text{Fe}^{2+}/\text{Mg})_{\text{melt}}$ ) and (ii)  $\text{Fe}^{3+}$  comprises 10% of the total moles of iron in the melt (following Hauri, 1996). Unless stated otherwise, all major, trace, and volatile element concentrations reported below have been corrected for equilibrium olivine addition/subtraction so that the inclusion is in equilibrium with the host olivine.

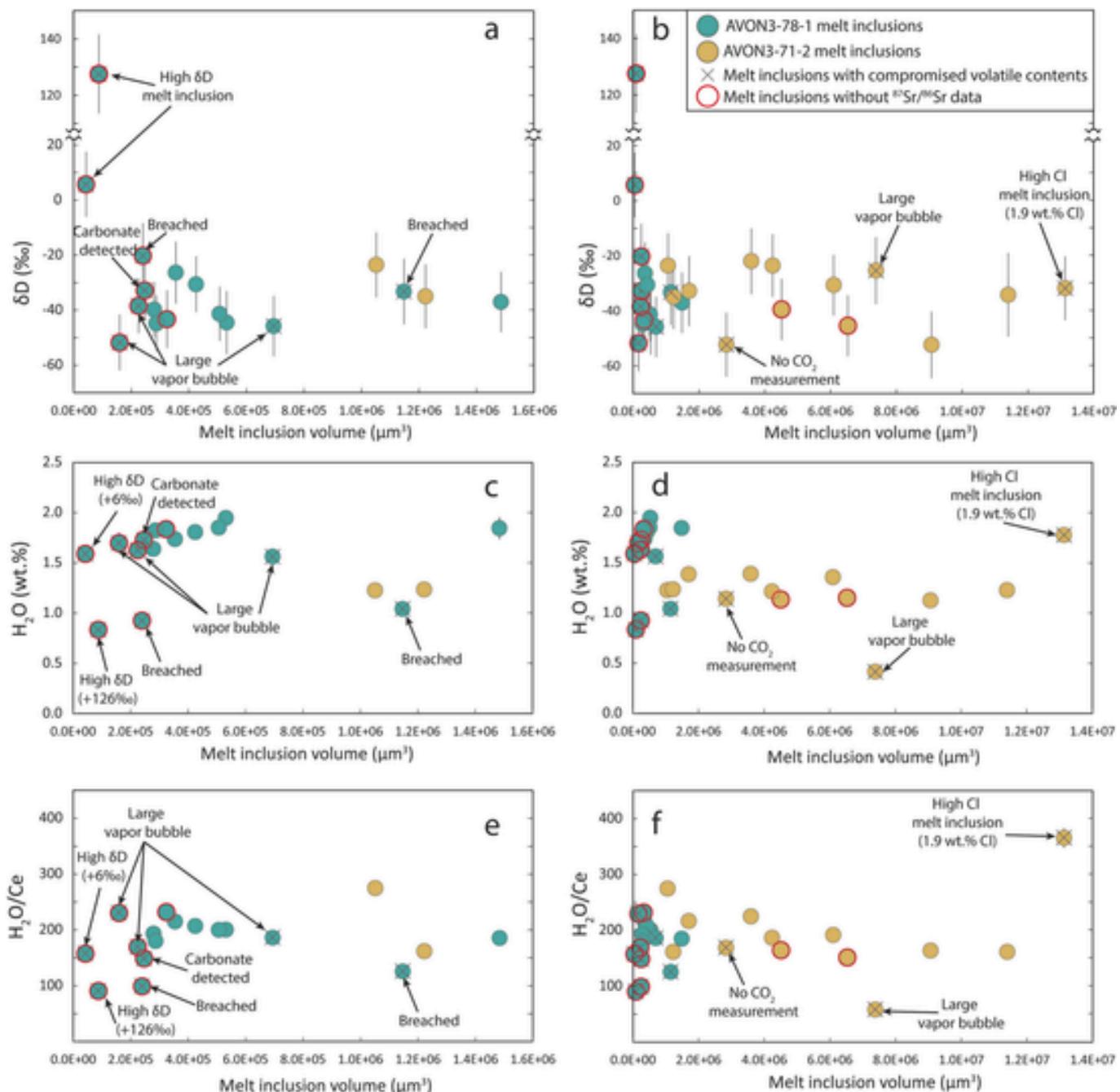
The supplement also details the method of accounting for all  $\text{CO}_2$  within a melt inclusion, including both the  $\text{CO}_2$  in the glass analyzed by SIMS, the  $\text{CO}_2$  contained within the vapor bubble obtained by Raman spectroscopy, and the volumes of bubble and glass for each melt inclusion obtained by X-ray computed microtomography (see Table S3 for the method of reconstructing the total  $\text{CO}_2$  content of each melt inclusion and supplementary section S1.3 for analytical methods).

### 3. Methods for data treatment

We use a series of criteria to identify inclusions where the volatile contents have been compromised. This way the results and discussion can focus on primary magmatic signatures. Fig. 2 shows all melt inclusions from which we collected volatile data. Inclusions whose volatile contents were likely compromised are designated with a black "x" in Fig. 2. We do not consider these inclusions further in our data analysis and interpretation so as to focus on primary magmatic signatures.

Inclusions are excluded if they show evidence of being affected by one or more of the following:

- Diffusive proton loss.** Hydrogen isotopes can indicate whether  $\text{H}_2\text{O}$  loss from the melt inclusion has occurred by proton diffusion through the host olivine. Melt inclusions that have been affected by proton diffusion will have low  $\text{H}_2\text{O}$  contents and high  $\delta D$  values due to faster diffusion of hydrogen over deuterium (Hauri, 2002; Gaetani et al., 2012; Bucholz et al., 2013; Hartley et al., 2015). Therefore, we measured  $\delta D$  on all melt inclusions reported in this study. Melt inclusions in the Malumalu sample (AVON3-78-1) have a  $\delta D$  range of  $-51.8\text{‰}$  to  $+127.6\text{‰}$ , and melt inclusions in the Vailulu'u sample (AVON3-71-2) have a  $\delta D$  range of  $-52.4\text{‰}$  to  $-22.0\text{‰}$  (Fig. 2). Diffusive  $\text{H}_2\text{O}$  loss from melt inclusions should produce a systematic relationship between melt inclusion  $\delta D$  and volume. Melt inclusions from both lavas with volumes  $>10^5 \mu\text{m}^3$  exhibit no systematic relationship between  $\delta D$  and melt inclusion volume, and they exhibit a relatively narrow range in  $\delta D$  of  $-35.9 \pm 17.4\text{‰}$  (2SD,  $N = 18$ ). However, the two melt inclusions with positive  $\delta D$  values, both from Malumalu (samples AVON3-78-1#13 and AVON3-78-1#32), have the smallest volumes ( $\sim 8.8 \times 10^4 \mu\text{m}^3$  and  $4.3 \times 10^4 \mu\text{m}^3$ ). Prior work has shown that smaller volume melt inclusions are more susceptible to diffusive proton loss that results in elevated  $\delta D$  (Hauri, 2002; Gaetani et al., 2012; Bucholz et al., 2013; Hartley et al., 2015). Therefore, the two small-volume inclusions are excluded.
- Breaching.** Two melt inclusions have likely suffered breaching because  $\text{CO}_2$  is  $<200 \text{ ppm}$  and  $\text{H}_2\text{O}$  is  $<1 \text{ wt\%}$  (see Fig. S4; AVON3-78-1#25, AVON3-78-1#37). Both melt inclusions exhibit melt-filled cracks that intersect the melt inclusions, which is strong physical evidence that supports the breaching interpretation.
- Extremely large vapor bubbles.** Three melt inclusions (AVON3-78-1#10, AVON3-78-1#28, AVON3-71-2#20) have vapor bubbles that make up  $>10 \text{ vol\%}$  of the melt inclusion and are excluded. Prior work suggested that large vapor bubbles result from simultaneous entrapment of a mix of melt and  $\text{CO}_2$  fluid (e.g.



**Fig. 2.** (a–b)  $\delta\text{D}$  (‰), (c–d)  $\text{H}_2\text{O}$  (wt.-%), and (e–f)  $\text{H}_2\text{O}/\text{Ce}$  versus melt inclusion volume ( $\mu\text{m}^3$ ). Left-hand side panels focus (zoom in) on the Malumalu melt inclusions and right-hand side panels (zoom out) to show the Vailulu'u melt inclusions. In panel (a)  $\delta\text{D}$  errors represent the  $2\sigma$  in run measurement precision of each analysis, which is generally slightly lower than the reproducibility of the secondary standard ALV1833-1 (i.e.,  $\pm 16\%$ , 2SD; see Table S2). Melt inclusion  $\text{H}_2\text{O}$  compositions are corrected for olivine addition/subtraction to be in equilibrium with the host olivine (olivine correction is as in Section 2 of the main text). Measurement errors for  $\text{H}_2\text{O}$ , and reproducibility of  $\text{H}_2\text{O}$  analyses on ALV1833-1, are smaller than the data symbol (see Table S2 and Supplementary section S1.6). As discussed in Section 3 of the text, we infer that the volatile contents of eleven melt inclusions shown in this figure have been compromised in one of several ways—hydrogen diffusion loss (and resultant elevated  $\delta\text{D}$ ), breaching, large vapor bubbles, extremely high Cl concentrations, presence of carbonate, and/or missing Raman-based  $\text{CO}_2$  data—and are marked with an “x” symbol. The compromised melt inclusions are not shown in subsequent figures.

Anderson, 1974; Frezzotti, 2001; Hanyu et al., 2020; Moore et al., 2015), which is not representative of melt-only compositions.

iv) No  $\text{CO}_2$  detected by Raman. No Fermi doublets were observed by Raman spectroscopy for two melt inclusions (AVON3-71-2#14, AVON3-71-2#18), thus the  $\text{CO}_2$  densities may have been too low for detection. To err on the side of caution, we are excluding these two melt inclusions from the interpretations of this study. The sample AVON3-71-2#18 that is being excluded for this sole

reason (i.e., undetectable  $\text{CO}_2$  in the vapor bubble) would not affect the interpretations of this study if it were included. For other excluded melt inclusions,  $\text{CO}_2$  densities in the vapor bubbles were highly uncertain (AVON3-71-2#20;  $\pm 0.092$ ) or so low that negative values resulted from the calibration (AVON3-78-1#25, AVON3-78-1#37, AVON3-71-2#14), and the volatile data for these four melt inclusions are excluded from the discussion.

v) *Carbonate detected by Raman.* If carbon exists as carbonate in a melt inclusion, the  $\text{CO}_2$  budget of the melt inclusion may be underestimated. In this study, carbonates were detected by Raman in two Malumalu melt inclusions (AVON3-78-1#15 and AVON3-78-1#32), and volatile data for these inclusions are excluded.

## 4. Results

### 4.1. Hydrogen and strontium isotopes

After filtering the data for the compromised melt inclusions, the Samoan melt inclusions from Malumalu ( $-38.5 \pm 13.5\text{\textperthousand}$ ,  $N = 8$ ) and Vailulu'u ( $-33.9 \pm 19.7\text{\textperthousand}$ ,  $N = 10$ ) are relatively homogenous in  $\delta\text{D}$  with an average  $\delta\text{D}$  value of  $-35.9\text{\textperthousand}$  ( $\pm 17.4\text{\textperthousand}$ , 2SD,  $N = 18$ ). The Samoan inclusions have heavier  $\delta\text{D}$  than most MORB and Lō'ihi (now known as Kama'ehuakanaloa) values, consistent with heavy  $\delta\text{D}$  values (ca.  $-40\text{\textperthousand}$ ) found in OIBs with recycled components (Loewen et al., 2019).

Critically,  $^{87}\text{Sr}/^{86}\text{Sr}$  measurements in melt inclusions provide a test for whether the Samoan melt inclusions of this study represent the Samoan mantle sources sampled by Vailulu'u and Malumalu glasses. If they do represent the same sources, then the volatile and trace element concentrations of the melt inclusions can be directly compared to concentration measurements in Samoan pillow lava glasses. The  $^{87}\text{Sr}/^{86}\text{Sr}$  range for all of the Malumalu (sample AVON3-78-1) melt inclusions reported here is large, from 0.706594 to 0.709225 (Figs. 3, 4). The inclusions largely fall within the range reported for Malumalu pillow glasses and whole rocks (0.706374–0.708901; Workman et al., 2006). The  $^{87}\text{Sr}/^{86}\text{Sr}$  range for all of the Vailulu'u (sample AVON3-71-2) melt inclusions reported here is smaller compared to AVON3-78-1 melt inclusions, from 0.704858 to 0.705718. The range of Vailulu'u melt inclusion  $^{87}\text{Sr}/^{86}\text{Sr}$  overlaps with the range reported for Vailulu'u pillow glasses (0.705352–0.706720; Workman et al., 2006) but, relative to Vailulu'u pillow glasses and whole rocks, the Vailulu'u melt inclusions are shifted to less radiogenic (lower)  $^{87}\text{Sr}/^{86}\text{Sr}$  values. Taken together, the  $^{87}\text{Sr}/^{86}\text{Sr}$  for the Malumalu and Vailulu'u melt inclusions in this study bracket the  $^{87}\text{Sr}/^{86}\text{Sr}$  obtained pillow glasses from each volcano: the lowest  $^{87}\text{Sr}/^{86}\text{Sr}$  in the melt inclusions (0.704858) is somewhat lower than the lowest value measured in whole rocks and pillow glasses (0.705352) from these two seamounts, and the highest  $^{87}\text{Sr}/^{86}\text{Sr}$  in the melt inclusions (0.709225) is slightly higher than the highest value measured in pillow glasses (0.708901). This statement is also true for the  $(\text{La}/\text{Sm})_N$  and  $\text{K}_2\text{O}/\text{TiO}_2$  for the melt inclusions compared to the whole rocks and pillow glasses (Fig. 4). Therefore, the Malumalu and Vailulu'u melt inclusions from these two hand samples represent the full compositional range previously identified in these volcanoes and, in fact, span nearly the entire range encountered in  $^{87}\text{Sr}/^{86}\text{Sr}$  (i.e., 0.7045 to 0.7089) for the eastern Samoan volcanic province—including Ta'u, Vailulu'u, and Malumalu volcanoes—examined by Workman et al. (2006). Thus, the melt inclusions can be used to interpret mantle source geochemical characteristics, including volatile budgets, of the volcanoes examined in Workman et al. (2006).

### 4.2. Volatiles

After omitting melt inclusions that have been contaminated, breached, affected by hydrogen diffusion, and have unreliable  $\text{CO}_2$  analyses by Raman, the magmatic volatile content of the Samoan melt inclusions can be explored. Fig. 5 provides ratios ( $\text{H}_2\text{O}/\text{Ce}$ ,  $\text{CO}_2/\text{Nb}$ ,  $\text{Cl}/\text{Nb}$ ,  $\text{S}/\text{Gd}$ , and  $\text{F}/\text{Nd}$ ) of volatile ( $\text{H}_2\text{O}$ ,  $\text{CO}_2$ ,  $\text{Cl}$ ,  $\text{S}$ , and  $\text{F}$ ) to nonvolatile ( $\text{Ce}$ ,  $\text{Nb}$ ,  $\text{Gd}$ , and  $\text{Nd}$ ) incompatible trace elements (ITEs) that have similar mineral-melt partition coefficients during mantle melting and magmatic differentiation processes. Below we examine the volatile concentrations and their ratios to nonvolatile ITEs.

#### 4.2.1. $\text{CO}_2$ in Samoan melt inclusions

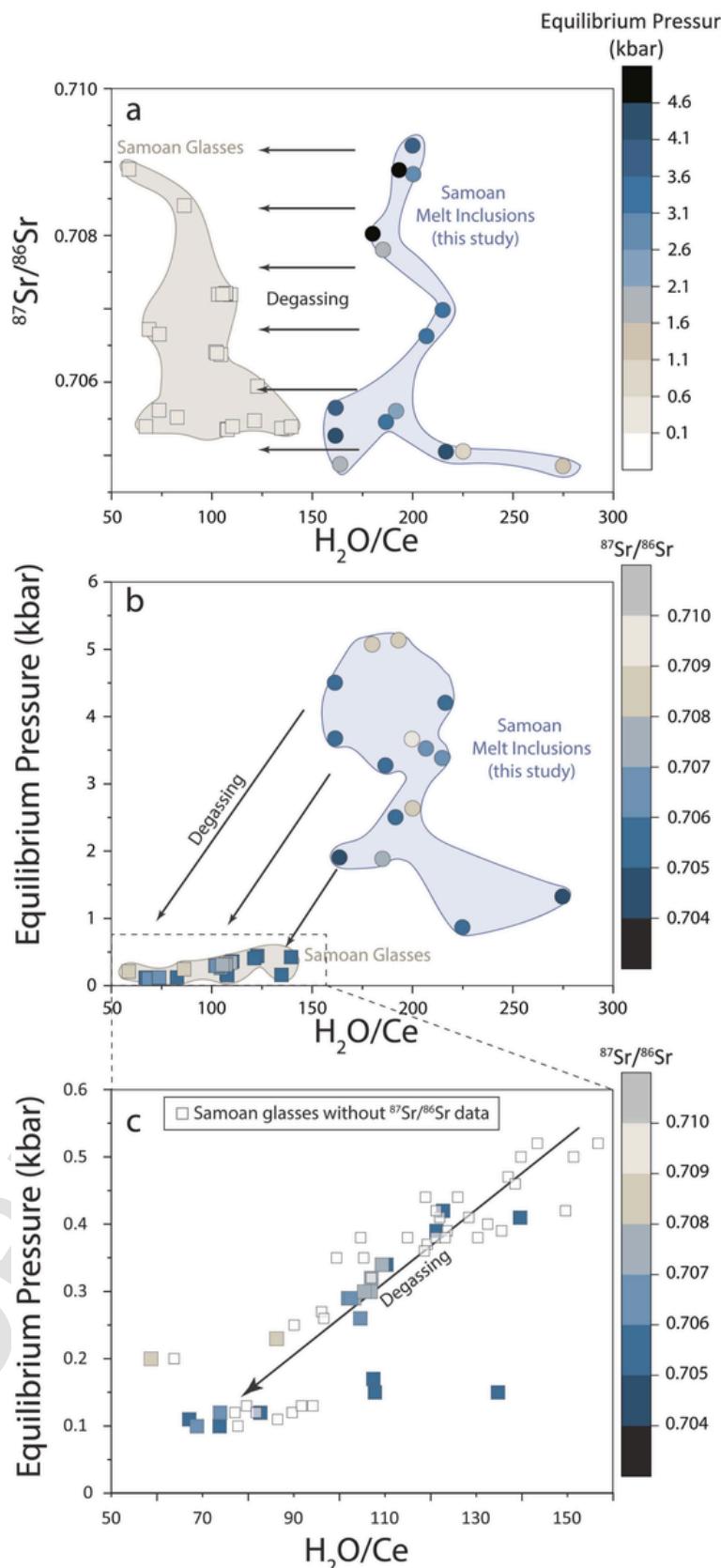
We use the  $\text{CO}_2$  concentration in the melt inclusion glass analyzed by SIMS and the  $\text{CO}_2$  density in the vapor bubble from Raman spectroscopy to reconstruct the total  $\text{CO}_2$  concentration of the melt inclusion. When comparing melt inclusion and vapor bubble  $\text{CO}_2$ , 3 to 94 mass percent (average  $\sim 59$  mass percent) of the total  $\text{CO}_2$  (i.e., melt inclusion  $\text{CO}_2$  + vapor bubble  $\text{CO}_2$ ) resides in the vapor bubble, consistent with prior findings of large  $\text{CO}_2$  fractions residing in melt inclusion vapor bubbles (e.g., DeVitre et al., 2021; Aster et al., 2016; Moore et al., 2015). Malumalu and Vailulu'u melt inclusions have relatively similar total  $\text{CO}_2$  contents of 1506–5746 ppm and 658–4771 ppm, respectively. The associated host pillow glasses for the Malumalu and Vailulu'u samples have total  $\text{CO}_2$  contents of 70 ppm and 179 ppm, respectively (Workman et al., 2006), which is far lower than the melt inclusions from these two volcanoes. These observations are consistent with greater  $\text{CO}_2\text{-H}_2\text{O}$  saturation pressures for the melt inclusions than the pillow glasses.

During melting and crystallization, the incompatibility (i.e., the bulk partition coefficient) of  $\text{CO}_2$  is similar to Nb and Ba, thus the ratios of  $\text{CO}_2/\text{Nb}$  or  $\text{CO}_2/\text{Ba}$  have been used to estimate upper mantle carbon content (Hirschmann, 2018; Michael and Graham, 2015; Saal et al., 2002; Cartigny et al., 2008; Le Voyer et al., 2017). However, only undegassed,  $\text{CO}_2$ -undersaturated melts will have  $\text{CO}_2/\text{Nb}$  or  $\text{CO}_2/\text{Ba}$  that are representative of the mantle source, but such melts are rare (Hauri et al., 2018; Graham and Michael, 2021; Le Voyer et al., 2017; Michael and Graham, 2015; Saal et al., 2002). These  $\text{CO}_2$ -undersaturated melts tend to be geochemically depleted (Hauri et al., 2018; Shimizu et al., 2023) and have higher  $\text{CO}_2/\text{Nb}$  than most submarine volcanic glasses, which are  $\text{CO}_2$ -saturated and thus degassed (e.g., the Samoan pillow glasses and melt inclusions examined here). With partially degassed melts, the best estimate for primary magmatic  $\text{CO}_2$  comes from the highest  $\text{CO}_2/\text{Nb}$  (or  $\text{CO}_2/\text{Ba}$ ) ratio of a melt inclusion suite or submarine glass suite (Matthews et al., 2017; Rosenthal et al., 2015). The Samoan melt inclusions in this study have a measured  $\text{CO}_2/\text{Nb}$  range of 19 to 125 for Vailulu'u melt inclusions and 27 to 133 for Malumalu melt inclusions, and such ratios are lower than inferred for mantle sources of oceanic hotspot lavas (e.g.,  $> 1000$  for Iceland; Matthews et al., 2021) and MORB ( $\text{CO}_2/\text{Nb} > 283$ ; Michael and Graham, 2015). Thus, we suggest the Samoan samples are  $\text{CO}_2$ -saturated and degassed, which is consistent with highly variable total  $\text{CO}_2$  contents (658 to 5746 ppm) relative to the narrow range of Nb concentrations (26.3–55.8 ppm) in our Samoan melt inclusions (Fig. S6b,d). The Samoan pillow glasses from these two volcanoes have even lower (more degassed)  $\text{CO}_2$  concentrations while exhibiting similar Nb concentrations to the melt inclusions (Fig. S6b,d).

#### 4.2.2. Cl in Samoan melt inclusions versus Samoa pillow glasses

Malumalu melt inclusions have 651–1198 ppm Cl, and Vailulu'u melt inclusions have 412–765 ppm Cl (excluding the high Cl inclusion AVON3-71-2#14, which has a Cl concentration of 1.9 wt% Cl). The associated pillow glasses from the two samples hosting the melt inclusions—AVON3-78-1 and AVON3-71-2—have 1004 ppm Cl and 1490 ppm Cl, respectively. The suite of pillow glasses from all Malumalu and Vailulu'u samples have Cl contents that vary from 886 to 1725 ppm and 547 to 1818 ppm, respectively.

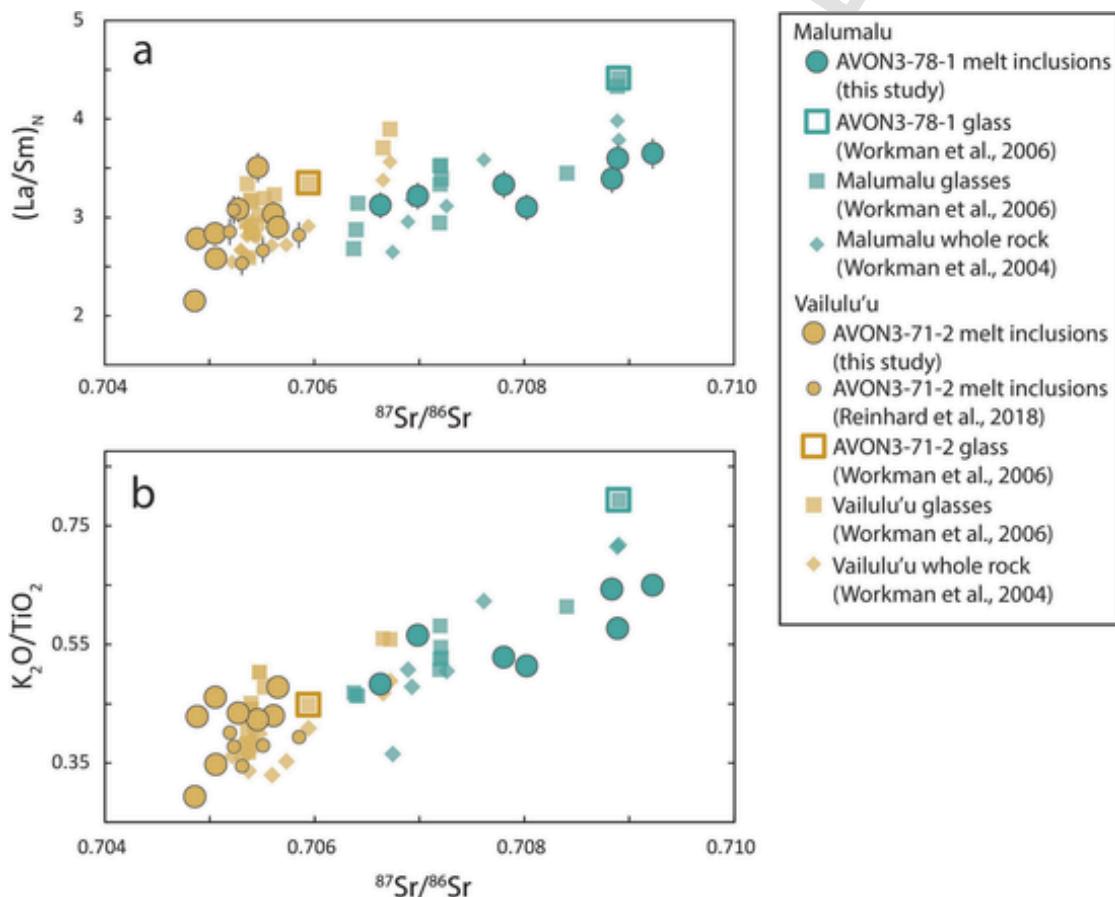
To determine if any melt inclusions are influenced by the assimilation of seawater-derived materials, we use  $\text{Cl}/\text{Nb}$  as an indicator of assimilation of seawater-derived materials (e.g., Kendrick et al., 2013, 2015; Kent et al., 1999a, 1999b, 2002). This is a useful ratio because Cl and Nb behave similarly during crystal fractionation and melting due to their similar incompatibility in basalt melts (Rowe and Lassiter, 2009). Malumalu melt inclusions have a  $\text{Cl}/\text{Nb}$  range of 14–25 and Vailulu'u melt inclusions have a  $\text{Cl}/\text{Nb}$  range of 14–22 (excluding melt inclusion sample AVON3-71-2#14, which has  $\text{Cl}/\text{Nb} = 840$ ) (Fig. 5). The Samoan melt inclusions have  $\text{Cl}/\text{Nb}$  that overlap with mid-ocean ridge



**Fig. 3.**  $^{87}\text{Sr}/^{86}\text{Sr}$ ,  $\text{H}_2\text{O}/\text{Ce}$ , and equilibrium pressure for submarine pillow glasses from Malumalu and Vailulu'u seamounts are compared with melt inclusions from Malumalu sample AVON3-78-1 and Vailulu'u sample AVON3-71-2. (a)  $^{87}\text{Sr}/^{86}\text{Sr}$  versus  $\text{H}_2\text{O}/\text{Ce}$  for Samoan melt inclusions and glasses with the color bar representing equilibrium pressure (kbar). The pillow glass data show separation in  $\text{H}_2\text{O}/\text{Ce}$  from the melt inclusion data, even though both datasets span a similar range in

**Fig. 3.**—continued

$^{87}\text{Sr}/^{86}\text{Sr}$ . Circles represent melt inclusions and squares represent Samoan pillow glasses from [Workman et al. \(2006\)](#). Melt inclusions in this figure do not include any volatile-compromised melt inclusions (as shown in [Fig. 2](#)). VESIcal was used to calculate the equilibrium pressure of pillow glasses and melt inclusions using MagmaSat ([Ghiorso and Gualda, 2015](#); [Iacovino et al., 2021](#); [Wieser et al., 2022](#)) assuming temperature (1200 °C). Samoan pillow glass data are from [Workman et al. \(2006\)](#), except major elements used to calculate equilibrium pressure for AVON3-71-2 and AVON3-78-1 (which are from [Kendrick et al., 2015](#)). (b) Equilibrium pressure versus  $\text{H}_2\text{O}/\text{Ce}$  for Samoan melt inclusions and pillow glasses with the color bar representing  $^{87}\text{Sr}/^{86}\text{Sr}$ . (c) Expanded view of panel b that shows equilibrium pressure versus  $\text{H}_2\text{O}/\text{Ce}$  for Samoan pillow glasses only. Smaller square data points with black outlines are pillow glasses lacking  $^{87}\text{Sr}/^{86}\text{Sr}$  data. Only samples with  $\text{CO}_2\text{-H}_2\text{O}$  saturation pressures  $\geq 0.1$  kbar are shown (just one Samoan glass with  $^{87}\text{Sr}/^{86}\text{Sr}$  data has a saturation pressure  $< 0.1$  kbar, sample AVON3-68-03 Rpt, so its removal does not significantly impact the dataset).



**Fig. 4.** Incompatible trace element ratios versus  $^{87}\text{Sr}/^{86}\text{Sr}$  are shown for Samoan pillow glasses from Malumalu and Vailulu'u seamounts and new melt inclusion data from Malumalu sample AVON3-78-1 and Vailulu'u sample AVON3-71-2. The melt inclusions have incompatible trace element ratios and  $^{87}\text{Sr}/^{86}\text{Sr}$  compositions that fall within the range identified in Samoan pillow glasses, indicating that the Samoan melt inclusions sample melts represented by the pillow glass dataset. (a)  $(\text{La}/\text{Sm})_N$  versus  $^{87}\text{Sr}/^{86}\text{Sr}$ , where N signifies normalization to primitive mantle ([McDonough and Sun, 1995](#)). (b)  $\text{K}_2\text{O}/\text{TiO}_2$  versus  $^{87}\text{Sr}/^{86}\text{Sr}$ . Data for Samoan pillow glasses are from [Workman et al. \(2006\)](#) and [Kendrick et al. \(2015\)](#). Error bars for  $\text{La}/\text{Sm}$  (4.2%) are shown for the melt inclusions and are calculated as  $((\text{La error})^2 + (\text{Sm error})^2)^{0.5}$ , where the 2RS of the secondary standard BCR-2 is used for La error (1.9%) and Sm error (3.8%) (see Table S2). Error bars on  $\text{K}_2\text{O}/\text{TiO}_2$  are smaller than the data symbols. Melt inclusions that are compromised with respect to their volatile contents are not shown.

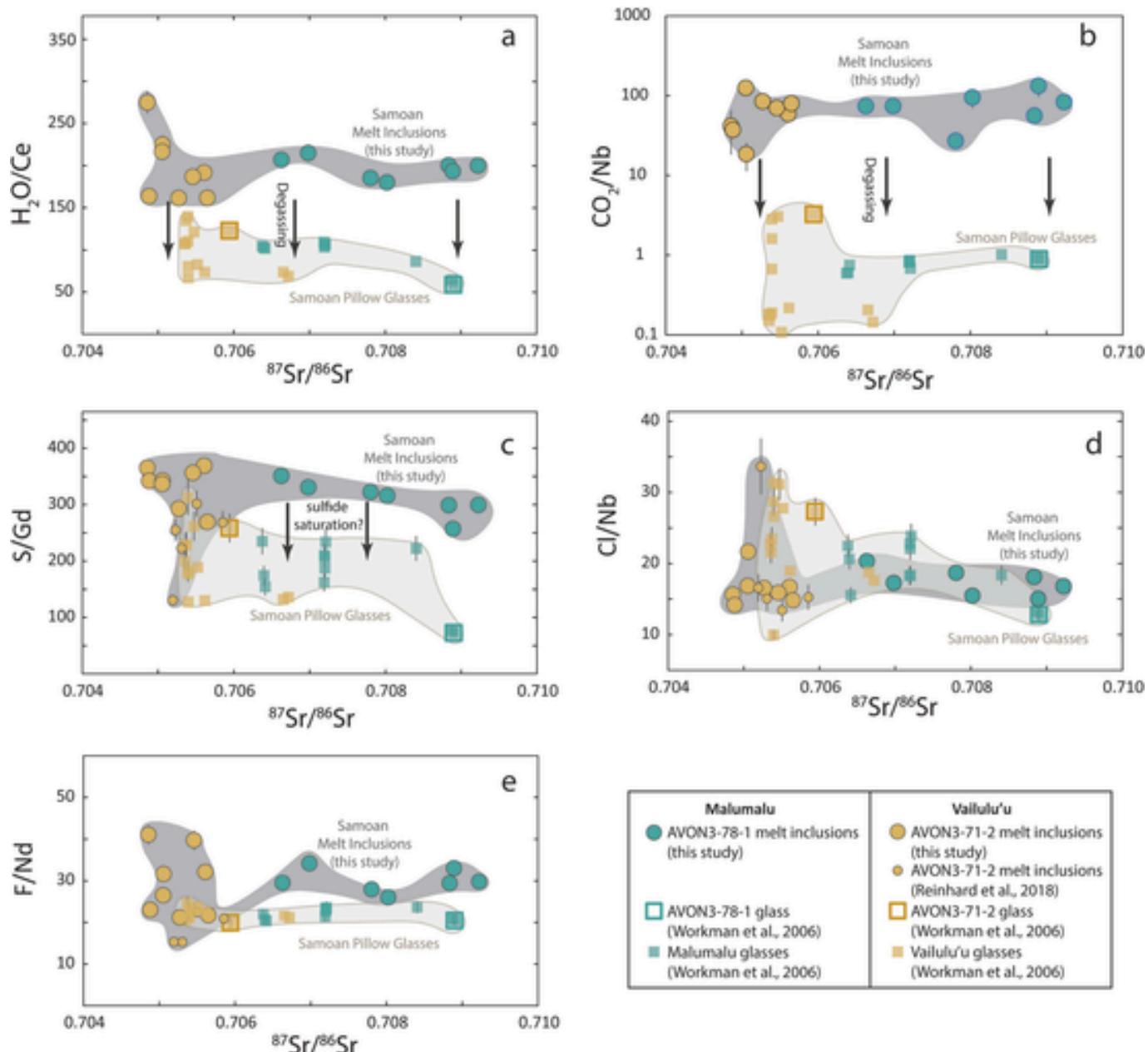
basalts (MORB), where  $\text{Cl}/\text{Nb}_{\text{MORB}}$  is 5 to 17 ([Lassiter et al., 2002](#); [Le Roux et al., 2006](#)), but the Samoan inclusions tend to be shifted to higher  $\text{Cl}/\text{Nb}$ .

The pillow glasses for the Malumalu (AVON3-78-1) and Vailulu'u (AVON3-71-2) samples hosting the Samoan melt inclusions have  $\text{Cl}/\text{Nb}$  of 13 and 27, respectively ([Kendrick et al., 2015](#)). Both lavas have been previously suggested to have assimilated seawater-derived materials ([Kendrick et al., 2014](#); [Reinhard et al., 2018](#)). We find some evidence for assimilation in a single melt inclusion from Vailulu'u (AVON3-71-2#14), which has a high  $\text{Cl}/\text{Nb}$  ratio of 840, suggesting contamination by seawater-derived materials; however, any such assimilation does not appear to have impacted  $^{87}\text{Sr}/^{86}\text{Sr}$  ( $0.705471 \pm 0.000022$ ) as it is at the lower end of the range observed here ([Fig. 5](#)).

#### 4.2.3. S in Samoan melt inclusions versus Samoa pillow glasses

Malumalu melt inclusions have 1786–3163 ppm S, and Vailulu'u melt inclusions have 1582–2090 ppm S. The associated host pillow glasses (AVON3-78-1 and AVON3-71-2) have 831 ppm S and 2337 ppm S, respectively. The suite of pillow glasses from all Malumalu and Vailulu'u samples have S contents varying from 781 to 2391 ppm and 1010 to 4834 ppm, respectively ([Workman et al., 2006](#)).

Sulfur and Gd and Dy have similar incompatibility during melting and crystal fractionation ([Saal et al., 2002](#)) when sulfide is not present, except at lower pressures when sulfur degasses. However, Samoan basalts are known to host both sulfates and sulfides ([Labidi et al., 2015](#)). The melt inclusions in this study are sulfide-saturated ([Fig. S5](#); using the model of [Fortin et al., 2015](#) and [Smythe et al., 2017](#))—and sulfides are in fact, observed in some of the inclusions from the two samples examined here—thus S/Gd ([Fig. 5](#)) of Samoan melt inclusions



**Fig. 5.** Element ratios (volatile/incompatible trace element) are plotted against  $^{87}\text{Sr}/^{86}\text{Sr}$  for Samoan pillow glasses from Malumalu and Vailulu'u seamounts and the new melt inclusion data from Malumalu sample AVON3-78-1 and Vailulu'u sample AVON3-71-2. See Supplementary Text section S1.6 for a discussion on the calculation of the errors associated with the element ratios presented here; the error bars for the  $^{87}\text{Sr}/^{86}\text{Sr}$  of Samoan glasses and melt inclusions are smaller than the symbol size. Melt inclusions that are compromised with respect to their volatile contents (as shown in Fig. 2) are not shown. Only samples with  $\text{CO}_2\text{-H}_2\text{O}$  saturation pressures  $\geq 0.1$  kbar are shown (just one Samoan glass with  $^{87}\text{Sr}/^{86}\text{Sr}$  data has a saturation pressure  $< 0.1$  kbar, sample AVON3-68-03 Rpt, so its removal does not significantly impact the dataset).

in this study may have been influenced by sulfide saturation and are likely not representative of the Samoan primary melts.

#### 4.2.4. F in Samoan melt inclusions versus Samoa pillow glasses

Malumalu melt inclusions have 1161–1293 ppm F, and Vailulu'u melt inclusions have 777–1145 ppm F. The associated host pillow glasses (AVON3-78-1 and AVON3-71-2) have 1254 ppm F and 903 ppm F, respectively. The suite of pillow glasses from all Malumalu and Vailulu'u samples have F contents varying from 1173 to 1409 ppm, and 843 to 1188 ppm F, respectively (Workman et al., 2006).

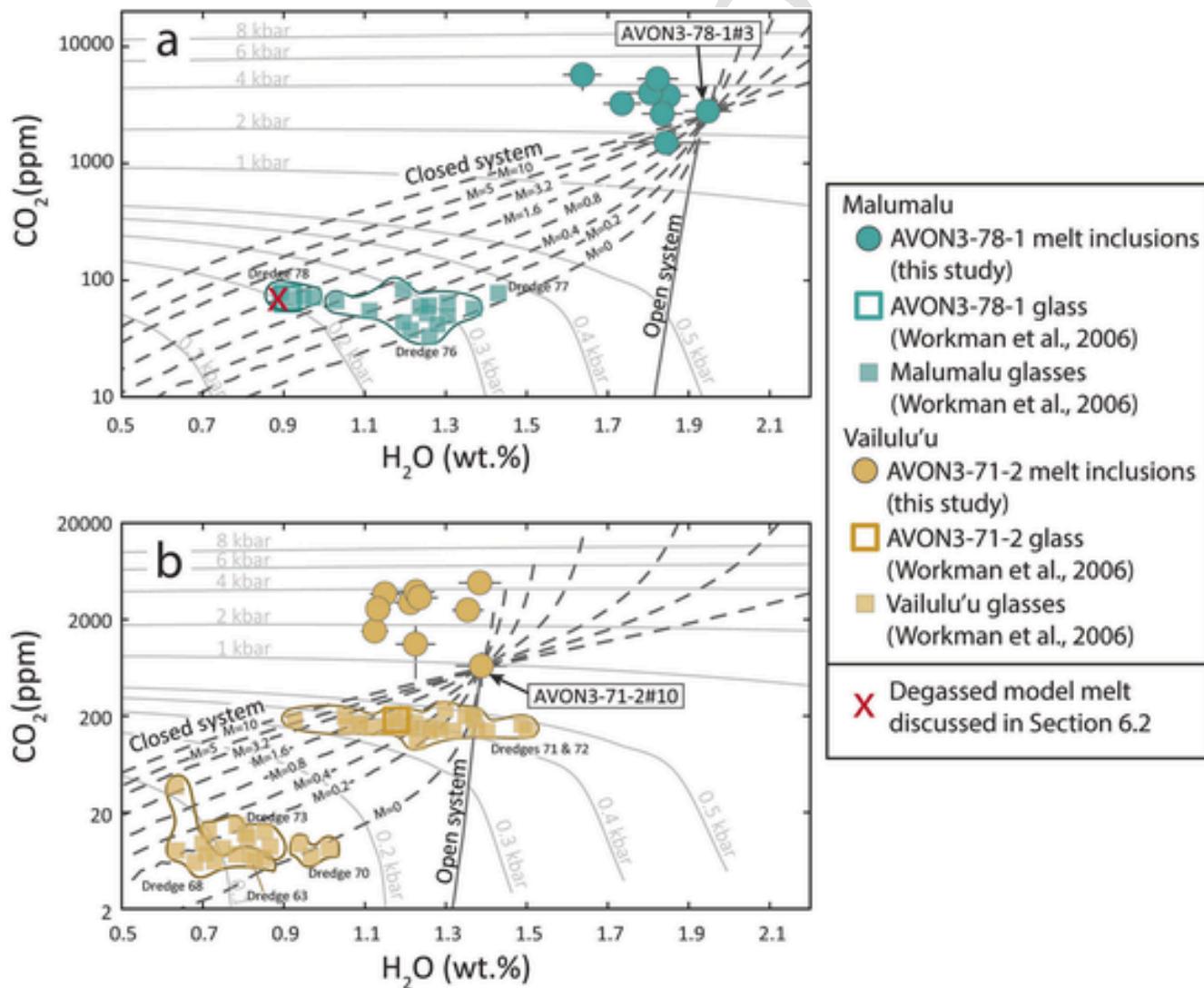
F/Nd in melt inclusions and glasses from global datasets of MORB and OIB are suggested to be relatively constant because there is no significant F/Nd fractionation during crystallization and melting. The F/

Nd values of global OIB and MORB have been estimated to be  $20.1 \pm 5.8$  (Workman et al., 2006), but there is evidence for greater F/Nd variability (Koleszar et al., 2009; Shimizu et al., 2016; Jackson et al., 2015; Lassiter et al., 2002; Métrich et al., 2014; Cabral et al., 2014; Rose-Koga et al., 2012). The average F/Nd for Malumalu melt inclusions ( $30 \pm 6$  2SD,  $N = 8$ ) is similar to the average F/Nd for Vailulu'u melt inclusions ( $29 \pm 14$  2SD,  $N = 10$ ), but Vailulu'u melt inclusions have more variability in F/Nd (Fig. 5). Three Vailulu'u melt inclusions (AVON3-71-2#7, #17, #22) have  $\sim 200$  ppm lower F contents than the other Samoan melt inclusions, but similar Nd contents, resulting in lower F/Nd (21–23). The Samoan melt inclusion have F/Nd ( $30 \pm 11$ , 2SD,  $N = 18$ ) tends to be higher than the canonical F/Nd ( $\sim 21$ ) for fresh global OIB and MORB, and higher than the average pillow glass F/

Nd from Malumalu ( $22 \pm 2$ , 2SD,  $N = 17$ ; [Workman et al., 2006](#)) and Vailulu'u ( $22 \pm 3$ , 2SD,  $N = 48$ ; [Workman et al., 2006](#)). However, the average F/Nd of Samoan melt inclusions overlaps within two standard deviations of the canonical F/Nd value and the F/Nd of Samoan pillow glasses. Nonetheless, the Vailulu'u melt inclusions tend to very high F/Nd up to 41. Elevated F may be characteristic of the HIMU mantle source ([Hauri and Hart, 1993](#); [Rose-Koga et al., 2017](#); [Jackson et al., 2015](#)), and Vailulu'u is thought to contain a small contribution from a HIMU component ([Workman et al., 2004](#); [Jackson et al., 2014](#)). However, we do not have an explanation for the higher F/Nd in the Vailulu'u melt inclusions relative to the glasses from this seamount.

#### 4.2.5. $H_2O$ in Samoan melt inclusions versus Samoa pillow glasses

Malumalu melt inclusions have 1.56–1.95 wt%  $H_2O$ , and Vailulu'u melt inclusions have 1.13–1.39 wt%  $H_2O$ . The associated host glasses from these two submarine samples tend to have lower  $H_2O$  contents than the melt inclusions with 0.91 wt%  $H_2O$  (AVON3-78-1) and 1.18 wt%  $H_2O$  (AVON3-71-2), respectively. The low  $H_2O$  in the AVON3-78-1 Malumalu glass sample lies within the range of  $H_2O$  for pillow glass samples from Malumalu (0.89 to 1.43 wt%), and the same applies for the AVON3-71-2 Vailulu'u glass sample as it also lies in the range of pillow glasses from this seamount (0.63 to 1.50 wt%) ([Fig. 6](#)). When the melt inclusions and pillow glasses are corrected for olivine fractionation to be in equilibrium with mantle olivine ( $Fo_{90}$ ), a figure of  $H_2O_{Fo_{90}}$  (wt%) versus  $Ce_{Fo_{90}}$  (ppm) shows that the melt inclusions and pillow glasses from Malumalu and Vailulu'u seamounts have similar



**Fig. 6.**  $CO_2$  (ppm) versus  $H_2O$  (wt%) for (a) melt inclusions from Malumalu sample AVON3-78-1 and all pillow glasses from Malumalu seamount and (b) melt inclusions from Vailulu'u sample AVON3-71-2 and all pillow glasses from Vailulu'u seamount. Melt inclusion data are from this study and Samoan pillow glass data are from [Workman et al. \(2006\)](#) and the data are shown in fields separated by the dredge number (which is provided next to the relevant field). Isobars are calculated using MagmaSat for an average melt inclusion composition of each sample at 1200 °C. Closed-system degassing curves are calculated with varying values for M (where "M" represents mass fraction of equilibrium fluid in the magma but here is given as a percentage; e.g., M = 1.6 means there was 1.6 wt% of fluid in the magma before degassing). Degassing trends assume an oxygen fugacity (QFM) and temperature (1200 °C) and are calculated in rhyolite-MELTS v1.2 using  $H_2O$  and  $CO_2$  compositions that match the highest  $H_2O$  melt inclusions in the respective panels (i.e., Malumalu melt inclusion AVON3-78-1#3 in the upper panel, and Vailulu'u melt inclusion AVON3-71-2#10 in the lower panel); other compositions used in the degassing calculation are in Table S9. Melt inclusion  $CO_2$  and  $H_2O$  concentrations shown are values that have been corrected to be in equilibrium with the host olivine. Volatile-compromised melt inclusions (as shown in [Fig. 2](#)) are not included in this figure. Other parental melt compositions are possible as initial values for the degassing model, but for simplicity, we use the highest  $H_2O$  inclusions because they allow the degassing model to encompass the pillow glasses with the highest  $H_2O$  concentrations.

$\text{Ce}_{\text{Fo90}}$ , but the inclusions have  $\text{H}_2\text{O}$  that is higher than the pillow glasses (Fig. S6c). We also show the same plot but instead provide the measured  $\text{H}_2\text{O}$  and Ce in the glass glasses—due to uncertainties associated with the olivine correction—and find that it does not change our conclusions.

Consistent with this observation, the  $\text{H}_2\text{O}/\text{Ce}$  range of the Malumalu AVON3-78-1 melt inclusions (180–232, average  $202 \pm 33$ , 2SD,  $N = 8$ ) and the Vailulu'u AVON3-71-2 melt inclusions (161–275, average  $190 \pm 78$ , 2SD,  $N = 10$ ) are higher than the host pillow glasses, 59 (for pillow glass from sample AVON3-78-1) and 123 (for pillow glass from sample AVON3-71-2), respectively. Critically, the average  $\text{H}_2\text{O}/\text{Ce}$  ( $197 \pm 58$  2SD,  $N = 15$ ) in the melt inclusion dataset from Malumalu and Vailulu'u seamounts is nearly twice as high as the same ratio in the Samoan pillow glasses from these two seamounts ( $106 \pm 51$ , 2SD,  $N = 65$ ). This is notable because the inclusions and pillow glasses are from the same two volcanoes, span the same range of  $^{87}\text{Sr}/^{86}\text{Sr}$  (Fig. 3), and have similar ITE ratios (Fig. 4), suggesting they sample the same mantle sources. This observation that melt inclusions have higher  $\text{H}_2\text{O}/\text{Ce}$  than submarine pillow glasses may relate to the fact that the Samoan melt inclusions from Malumalu and Vailulu'u have universally higher calculated MagmaSat  $\text{CO}_2\text{-H}_2\text{O}$  vapor saturation pressures (Fig. 3) than the submarine pillow glasses, reflective of higher entrapment pressures for the melt inclusions ( $> 0.870$  kbar) than the eruption pressures experienced by the glasses ( $< 0.53$  kbar) (see Fig. 3).

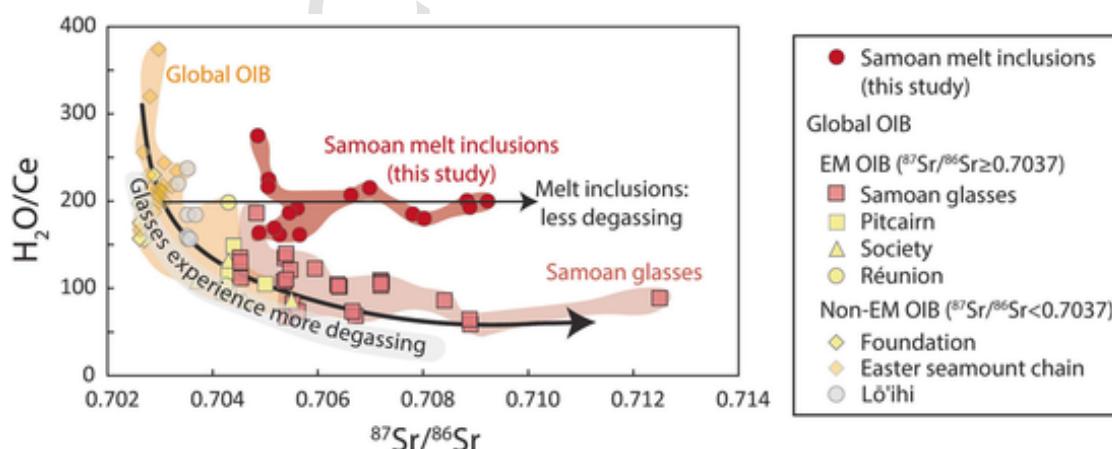
While Workman et al. (2006) found a negative correlation between  $\text{H}_2\text{O}/\text{La}$  and  $^{87}\text{Sr}/^{86}\text{Sr}$  for Samoan glasses, we find that the Samoan melt inclusions in this study exhibit no correlation between  $\text{H}_2\text{O}/\text{Ce}$  and  $^{87}\text{Sr}/^{86}\text{Sr}$  (Figs. 3, 5a, 7a), even though the melt inclusions are from two of the three volcanoes studied by Workman et al. (2006) and span the same range of  $^{87}\text{Sr}/^{86}\text{Sr}$  as their pillow glasses. In contrast to Workman et al. (2006), the  $\text{H}_2\text{O}/\text{Ce}$  of the melt inclusions is quite constant (i.e.,  $197 \pm 58$  2SD,  $N = 15$ ) over a range of  $^{87}\text{Sr}/^{86}\text{Sr}$  from 0.704858 to 0.709225, and the melt inclusions with the strongest EM2 mantle signatures (i.e.,  $^{87}\text{Sr}/^{86}\text{Sr} > 0.7080$ ,  $N = 4$ ) have elevated  $\text{H}_2\text{O}/\text{Ce}$  ( $193 \pm 19$  2SD) that is indistinguishable from the  $\text{H}_2\text{O}/\text{Ce}$  ( $202 \pm 31$ )

in melt inclusions with weaker EM2 signatures ( $^{87}\text{Sr}/^{86}\text{Sr} < 0.7080$ ,  $N = 3$ ).

While the Samoan pillow glasses and melt inclusions exhibit some overlap with each other in  $\text{H}_2\text{O}/\text{Ce}$  at the lowest  $^{87}\text{Sr}/^{86}\text{Sr}$ , they exhibit diverging trends in  $\text{H}_2\text{O}/\text{Ce}$  with increasing  $^{87}\text{Sr}/^{86}\text{Sr}$ : melt inclusions exhibit unchanging  $\text{H}_2\text{O}/\text{Ce}$  with increasing  $^{87}\text{Sr}/^{86}\text{Sr}$ , and glass exhibit decreasing  $\text{H}_2\text{O}/\text{Ce}$  with increasing  $^{87}\text{Sr}/^{86}\text{Sr}$ . Thus, the difference in  $\text{H}_2\text{O}/\text{Ce}$  between melt inclusions and glasses is largest at the highest  $^{87}\text{Sr}/^{86}\text{Sr}$  (Fig. 7). To illustrate this, we note that all four pillow glasses with the strongest EM2 signatures ( $^{87}\text{Sr}/^{86}\text{Sr} > 0.7080$ ) have  $\text{H}_2\text{O}/\text{Ce} < 90$  ( $\text{H}_2\text{O}/\text{Ce}$  average of  $75 \pm 30$ , 2SD) while the  $\text{H}_2\text{O}/\text{Ce}$  of the four melt inclusions with the highest  $^{87}\text{Sr}/^{86}\text{Sr}$  ( $> 0.7080$ ) have  $\text{H}_2\text{O}/\text{Ce}$  that is more than twice as high as that in the pillow glasses ( $\text{H}_2\text{O}/\text{Ce}$  average of  $193 \pm 19$ , 2SD). Prior studies (e.g., Workman et al., 2006) often cite that pillow glasses erupted at depths  $> 1000$  mbsl retain volatile/ITE ratios of the mantle source. All pillow glasses shown in Fig. 7 have  $\text{H}_2\text{O}\text{-CO}_2$  saturation pressures  $\geq 0.1$  kbar. Given that the melt inclusions and glasses in Fig. 7 exhibit deep saturation pressures and appear to sample the same mantle source (Fig. 4), we might expect the deeply-erupted pillow glasses should have the same  $\text{H}_2\text{O}/\text{Ce}$  as melt inclusions. These observations raise a key question that we will address in the discussion: why do melt inclusions and pillow glasses erupted at depths  $> 1000$  mbsl from the same volcanoes and with the same  $^{87}\text{Sr}/^{86}\text{Sr}$  (i.e., sampling the same mantle sources) have  $\text{H}_2\text{O}/\text{Ce}$  that differ, on average, by nearly a factor of two?

## 5. Modeling degassing processes for Samoan melts

We find that melt inclusions from the Vailulu'u and Malumalu seamounts have higher  $\text{CO}_2$  concentrations than their corresponding pillow glasses (Fig. 6), indicating that the deeply-erupted submarine pillow glasses are more degassed than the melt inclusions. An outstanding question is whether the lower  $\text{H}_2\text{O}/\text{Ce}$  in the pillow glasses reflects this higher degree of degassing via concomitant degassing of both  $\text{CO}_2$  and  $\text{H}_2\text{O}$ .



**Fig. 7.** Samoan melt inclusions and pillow glasses compared to a global dataset of submarine OIB glasses. Samoan melt inclusions—protected from the extensive degassing experienced by Samoan pillow glasses—have higher  $\text{H}_2\text{O}/\text{Ce}$  than Samoan pillow glasses, and thus fall off of the global submarine OIB pillow glass trend that shows decreasing  $\text{H}_2\text{O}/\text{Ce}$  with increasing  $^{87}\text{Sr}/^{86}\text{Sr}$ . The arrows in this figure are merely schematic to guide the reader and are not the result of a model. Only OIB glasses with  $^{87}\text{Sr}/^{86}\text{Sr}$  (to establish the level of source enrichment) and  $\text{MgO} > 4.5$  wt% (to minimize the impact of fractional crystallization) are shown. Sources of data (see Table S11): Pitcairn (Kendrick et al., 2014; Woodhead and Devey, 1993), Society (Devey et al., 1990; Kendrick et al., 2014), Réunion (Kendrick et al., 2017; Fretzdorff and Haase, 2002), Foundation (Kendrick et al., 2017), Easter seamount chain (Dixon et al., 2002), Lō'ihi (Dixon and Clague, 2001; Staudigel et al., 1984). Samples from Pitcairn, Society, Réunion, and Foundation can be found in a data compilation by Kendrick et al. (2017). Samples marked as having experienced assimilation of seawater Cl in Kendrick et al. (2017) are excluded. However, unlike Dixon et al. (2002), we do not use  $\text{H}_2\text{O}/\text{Ce}$  as a filter for assimilation. Samoan pillow glasses include submarine glasses characterized for  $^{87}\text{Sr}/^{86}\text{Sr}$  from Vailulu'u, Malumalu, Tupito (formerly known as Muli), Taumatau, and Ta'u (Kendrick et al., 2015; Workman et al., 2006). Volatile-compromised melt inclusions (as shown in Fig. 2) are not shown. Melt inclusions from other OIB localities are not shown because there are no other OIB melt inclusions characterized for both  $^{87}\text{Sr}/^{86}\text{Sr}$  and  $\text{H}_2\text{O}/\text{Ce}$ . Only samples with  $\text{CO}_2\text{-H}_2\text{O}$  saturation pressures  $\geq 0.1$  kbar are shown (just one Samoan glass with  $^{87}\text{Sr}/^{86}\text{Sr}$  data has a saturation pressure  $< 0.1$  kbar, sample AVON3-68-03 Rpt, so its removal does not significantly impact the dataset).

We show that the data are consistent with Samoan pillow glasses being related to the Samoan melt inclusions through closed-system degassing; the new insight gained here is that, when compared with genetically-related melt inclusions, it is clear that even deeply erupted glasses (with  $\text{CO}_2\text{-H}_2\text{O}$  saturation pressures  $\geq 0.1$  kbar) have lost significant  $\text{H}_2\text{O}$  and have lower  $\text{H}_2\text{O}/\text{Ce}$  compared to their primary melt (Fig. 3). This is evident in models we have run using a set of open- and closed-system degassing paths while also varying the fraction of  $\text{CO}_2 + \text{H}_2\text{O}$  fluid in equilibrium with the initial melt (Fig. 6).

There are indications that there is a volatile-bearing fluid phase coexisting with magma in magma reservoirs at depths before eruption (e.g.,  $\text{CO}_2$  fluxing: Rust et al., 2004; Blundy et al., 2010). Therefore, a free parameter  $M$  is introduced, representing the mass fraction of a volatile fluid in the magma + fluid system (e.g., VESIcal program has this parameter built-in; Iacovino et al., 2021; Wieser et al., 2022). In this case,  $[\text{CO}_2]_{\text{melt}} = (1-M)([\text{CO}_2]_{\text{melt observed}}) + M([\text{CO}_2]_{\text{equilibrium fluid}})$ . A similar equation is also used for  $\text{H}_2\text{O}$ . In the models, we define the variable  $M$  as the mass fraction of fluid (shown as a percentage in Fig. 6) in equilibrium with the melt, where  $M = 0\%$  means that no initial fluid is in equilibrium with the melt (i.e., all volatiles are dissolved in the melt, no fluid is present in excess), and where  $M = 100\%$  means that all volatiles are in the fluid (i.e., no melt is present). The higher the  $M$  value, the deeper degassing starts (Fig. S8). We generate degassing paths assuming a starting  $\text{H}_2\text{O}$  and  $\text{CO}_2$  composition equivalent to the melt inclusion with the highest  $\text{H}_2\text{O}$  content from Malumalu seamount (sample AVON3-78-1#3) to model Malumalu data, and the equivalent high- $\text{H}_2\text{O}$  melt inclusion from Vailulu'u (sample AVON3-71-2#10) to model Vailulu'u data. We use MagmaSat (Ghiorso and Gualda, 2015) to calculate the fluid composition (i.e.,  $X\text{CO}_2$  and  $X\text{H}_2\text{O}$ ) in equilibrium with the two melt inclusions (see Table S9 for model melt inclusion compositions). Additionally, we use rhyolite-MELTS v.1.2 (which incorporates the same fluid solubility model as MagmaSat) in rhyolite-MELTS to simulate degassing because this model is well-suited for high  $\text{CO}_2$  contents in the initial melt composition (Ghiorso and Gualda, 2015; Gualda et al., 2012). The models are run at 1200 °C, and start at 3–27 kbar, and run to 1 bar. The influence of oxygen fugacity on the model degassing path is negligible, so we assume QFM for all models. The compositional and physical parameters used in the MagmaSat and rhyolite-MELTS models are given in Table S9.

The degassing model results are presented in Fig. 6. While there is no single degassing path that explains the difference in pillow glass and melt inclusion  $\text{CO}_2$  and  $\text{H}_2\text{O}$  concentrations, closed-system degassing paths with  $M$  values ranging from 0% to 10% describe most of the pillow glasses from both seamounts. However, there are some Vailulu'u pillow glasses that have  $\text{H}_2\text{O}$  that is higher than the range described by the calculated degassing paths. For example, three Vailulu'u pillow glasses have higher  $\text{H}_2\text{O}$  concentrations than the highest- $\text{H}_2\text{O}$  melt inclusion. This could be because the melt inclusion suite in this study does not represent the full range of  $\text{H}_2\text{O}$  contents possible at the seamount, and higher initial  $\text{H}_2\text{O}$  concentrations may be more reflective of the primary melts.

A primary observation from this modeling exercise is that closed-system degassing from melt inclusion entrapment pressures (i.e.,  $> 0.87$  kbar) to eruption on the seafloor (pillow glasses have equilibrium pressures of 0.08 to 0.53 kbar) results in significant loss of both  $\text{CO}_2$  and  $\text{H}_2\text{O}$ , even for glasses with  $\text{CO}_2\text{-H}_2\text{O}$  saturation depths near 0.4 kbar (see closed-system degassing paths in Fig. 6). Furthermore, the presence of equilibrium fluid ( $M$ ) indicates that the magmatic system is richer in  $\text{CO}_2$ , and thus starts to degas at greater depths. Below we show that loss of  $\text{H}_2\text{O}$  during closed-system degassing to shallow levels (i.e.,  $< 0.53$  kbar) explains the lower  $\text{H}_2\text{O}/\text{Ce}$  in pillow glasses compared to the melt inclusions.

## 6. Discussion

$\text{H}_2\text{O}$  has incompatibility during mantle melting similar to Ce (Michael, 1995), thus  $\text{H}_2\text{O}/\text{Ce}$  of the melt inclusions and glasses should reflect the mantle source unless the samples have degassed. Clear negative correlations between  $\text{H}_2\text{O}/\text{Ce}$  and  $^{87}\text{Sr}/^{86}\text{Sr}$  in pillow glasses have been used to infer the  $\text{H}_2\text{O}/\text{LREE}$  (light rare earth element) ratios of the mantle source, assuming the  $\text{H}_2\text{O}$  has not been degassed. The primary observation in this study that we seek to explain is that Samoan pillow glasses and olivine-hosted melt inclusions span the same range of  $^{87}\text{Sr}/^{86}\text{Sr}$  and incompatible trace element ratios unaffected by differentiation—and thus sample the same mantle sources, yet the melt inclusions have  $\text{H}_2\text{O}/\text{Ce}$  ratios that are approximately twice as high as the pillow glasses. The second key observation is that the melt inclusion  $\text{H}_2\text{O}/\text{Ce}$  ratios are relatively constant over a wide range of  $^{87}\text{Sr}/^{86}\text{Sr}$ , in contrast to the pillow glasses whose  $\text{H}_2\text{O}/\text{Ce}$  ratios anticorrelate with  $^{87}\text{Sr}/^{86}\text{Sr}$ . This raises two key questions: Why do Samoan EM2 pillow glasses exhibit  $\text{H}_2\text{O}/\text{Ce}$  that are approximately half the values in Samoan EM2 melt inclusions at the same seamounts, and what does this mean for the  $\text{H}_2\text{O}$  content of the EM2 mantle?

### 6.1. Higher $\text{H}_2\text{O}/\text{Ce}$ in melt inclusions relative to pillow glasses: Diffusive $\text{H}_2\text{O}$ gain through host olivine and/or assimilation of seawater-derived materials?

In order to compare the volatile contents of the Samoan melt inclusions and pillow glasses in this study, it is important to assess whether the higher  $\text{H}_2\text{O}/\text{Ce}$  in the inclusions reflects processes operating in the magma during ascent, such as diffusive proton gain (i.e., addition of H) or assimilation of seawater-derived materials. Prior work has shown that, compared to large melt inclusions, small inclusions are more susceptible to diffusive loss or addition of protons, and this results in modified  $\delta\text{D}$  (Hauri, 2002; Portnyagin et al., 2008; Gaetani et al., 2012; Bucholz et al., 2013; Hartley et al., 2015). Given enough time, the  $\delta\text{D}$  of all inclusions will eventually equilibrate with the external melt. Diffusive proton loss from the inclusion can be identified by an increase in  $\delta\text{D}$ , which is associated with decreasing melt inclusion volume. Conversely, diffusive gain of protons by the inclusion is identified by a decrease in  $\delta\text{D}$ , where the magnitude of the  $\delta\text{D}$  reduction is greater in the smaller volume melt inclusions. The new  $\delta\text{D}$  analyses on melt inclusions show a lack of variation in  $\delta\text{D}$  ( $-35.9 \pm 17.4\%$ , 2SD,  $N = 18$ ) with melt inclusion size for melt inclusions with volumes  $> 10^5 \mu\text{m}^3$ , which indicates no water loss or gain in the larger inclusions (Fig. 2). However, the two smallest melt inclusions (AVON3-78-1#13 and AVON3-78-1#32), both with melt inclusion volume  $< 10^5 \mu\text{m}^3$ , have significantly higher  $\delta\text{D}$  ( $+128\%$  and  $+6\%$ , respectively) than the other melt inclusions from this lava, which indicates that the small volume melt inclusions have experienced diffusive  $\text{H}_2\text{O}$  loss, not  $\text{H}_2\text{O}$  addition (Fig. 2). These two small volume inclusions are excluded from the discussion below. While there is a lack of variation in  $\delta\text{D}$ , the variability in  $\text{H}_2\text{O}$  concentrations of the melt inclusions in both seamounts does not suggest that diffusion has fully overprinted the original  $\text{H}_2\text{O}$  concentration of the melt inclusions. Additionally, the variability in  $\text{H}_2\text{O}$  concentrations across the melt inclusions from a single sample (AVON3-78-1 or AVON3-71-2) does not relate to melt inclusion size, which indicates that the variability in  $\text{H}_2\text{O}$  is independent of H diffusive exchange with the host melt.

Assimilation of seawater-derived materials can increase  $\text{H}_2\text{O}$  in magmas (except if the magma is saturated in  $\text{H}_2\text{O}$ ), and this process can be traced by monitoring  $\text{Cl}/\text{Nb}$ , a ratio that is elevated in seawater-derived materials. One hypothesis for the higher  $\text{H}_2\text{O}/\text{Ce}$  in melt inclusions is that they have experienced more assimilation of seawater-derived materials compared to Samoan pillow glasses, and this hypothesis can be tested by measurement of  $^{87}\text{Sr}/^{86}\text{Sr}$ . A plot of  $\text{Cl}/\text{Nb}$  versus

$^{87}\text{Sr}/^{86}\text{Sr}$  shows that Samoan pillow glasses and melt inclusions have an overlapping range of  $\text{Cl}/\text{Nb}$  at any given  $^{87}\text{Sr}/^{86}\text{Sr}$ , so the offset to higher  $\text{H}_2\text{O}/\text{Ce}$  in the melt inclusions is not explained by greater assimilation experienced by melt inclusions (Fig. 5). We do acknowledge that one melt inclusion with exceptionally high  $\text{Cl}/\text{Nb}$  (840)—a ratio consistent with assimilation of seawater-derived materials (Table S1)—also has elevated  $\text{H}_2\text{O}/\text{Ce}$  (366), suggesting a role for assimilation in increasing the  $\text{H}_2\text{O}/\text{Ce}$  in this particular inclusion. However, this single inclusion has been excluded from the discussion below owing to its very high  $\text{Cl}/\text{Nb}$ .

In short, diffusive gain of  $\text{H}_2\text{O}$  and preferential assimilation of seawater-derived materials by the melt inclusions relative to the pillow glasses do not explain the higher  $\text{H}_2\text{O}/\text{Ce}$  in the melt inclusions relative to the pillow glasses. Instead, we argue in the following section that closed-system degassing plays a key role in lowering the  $\text{H}_2\text{O}$  and the  $\text{H}_2\text{O}/\text{Ce}$  of pillow glasses. In contrast, limited  $\text{H}_2\text{O}$  degassing has occurred in the melt inclusions in this study, leaving the melt inclusion  $\text{H}_2\text{O}/\text{Ce}$  relatively unchanged (although there may have been some  $\text{H}_2\text{O}$  loss prior to entrapment in crystallizing olivines), thereby explaining why melt inclusions have higher  $\text{H}_2\text{O}/\text{Ce}$  than pillow glasses. This contrasts with prior studies (e.g., Workman et al., 2006) suggesting that pillow glasses erupted at depths  $> 1000$  mbsl retain  $\text{H}_2\text{O}/\text{LREE}$  ratios of the mantle source.

## 6.2. Higher $\text{H}_2\text{O}/\text{Ce}$ in melt inclusions relative to pillow glasses: $\text{H}_2\text{O}$ loss in Samoan glasses caused by degassing

We argue that the difference in  $\text{H}_2\text{O}/\text{Ce}$  between Samoan EM2-flavored melt inclusions and pillow glasses from the same seamounts relates to greater degassing of the pillow glasses relative to the melt inclusions. Unlike the pillow glasses, which can experience degassing from great depth all the way to eruption and quenching on the seafloor, melt inclusions trap melts at magma chamber depths and act as pressure vessels, which inhibits degassing at shallower levels during magma ascent. To evaluate this hypothesis, we build on the indistinguishable  $^{87}\text{Sr}/^{86}\text{Sr}$  and nonvolatile incompatible trace element ratios between pillow glasses and melt inclusions and argue that, at the depth of melt inclusion entrapment, the  $\text{H}_2\text{O}/\text{Ce}$  of the melts that later erupted as Samoan pillow glasses was *also* the same as the Samoan melt inclusions (i.e., nonvolatile ITE ratios—which remain unmodified from primary melt compositions during fractional crystallization of magmas with  $\text{MgO}$  as high as identified in the data reported here (Fig. S1)—and  $^{87}\text{Sr}/^{86}\text{Sr}$  data obtained on the melt inclusions indicate that they derive from the same parental melts as the pillow glasses and thus should have started with the same  $\text{H}_2\text{O}/\text{Ce}$ ). The  $\text{H}_2\text{O}/\text{Ce}$  of the pillow glasses, but not the melt inclusions, was then lowered by  $\text{H}_2\text{O}$  degassing during ascent.

The hypothesis of  $\text{H}_2\text{O}$  loss from the matrix melts by degassing is supported by closed-system degassing models in Fig. 6. For example, the degassing model for Malumalum AVON3-78-1 starts with a melt inclusion composition like AVON3-78-1#3 which has an elevated  $\text{CO}_2$  concentration of 2810 ppm, the highest  $\text{H}_2\text{O}$  concentration from this lava (1.95 wt%), and a high entrapment pressure of 2.63 kbar (i.e., the  $\text{CO}_2\text{-H}_2\text{O}$  saturation pressure calculated for the melt inclusion using MagmaSat). Closed-system degassing of this melt composition from 2.63 kbar to 0.2 kbar—the approximate  $\text{CO}_2\text{-H}_2\text{O}$  saturation pressure for the pillow glass from this lava, sample AVON3-78-1—at  $M = 1.6$  lowers the water content by a factor of ~2.1 to 0.90 wt%, and the  $\text{CO}_2$  content by a factor of ~40 to 69 ppm. The red “x” in Fig. 6a defines the  $\text{CO}_2$  and  $\text{H}_2\text{O}$  composition of the model melt following degassing, and the degassed model melt is very similar in composition to the pillow glass sample AVON3-78-1 and helps explain the low  $\text{H}_2\text{O}$  (0.91 wt%) and  $\text{CO}_2$  (70 ppm) of AVON3-78-1. This is important because this particular pillow glass sample anchors the highest  $^{87}\text{Sr}/^{86}\text{Sr}$  (0.7089) and lowest  $\text{H}_2\text{O}/\text{Ce}$  (59) portion of the  $^{87}\text{Sr}/^{86}\text{Sr}$  versus  $\text{H}_2\text{O}/\text{Ce}$  array produced by Workman et al. (2006) who interpreted that the low  $\text{H}_2\text{O}/\text{Ce}$

in this EM2 glass reflects melting of a water-poor (“dry”) EM2 mantle. In contrast, we show that dramatic  $\text{H}_2\text{O}$ -loss during closed-system degassing of an ascending EM2 melt with an initially high  $\text{H}_2\text{O}/\text{Ce}$  ratio ( $\text{H}_2\text{O}/\text{Ce} = 200$ ) is responsible for the low  $\text{H}_2\text{O}/\text{Ce}$  (59) in the AVON3-78-1 glass. This model contrasts with prior studies (e.g., Workman et al., 2006) suggesting that pillow glasses erupted at depths  $> 1000$  mbsl ( $\geq 0.1$  kbar) retain the  $\text{H}_2\text{O}/\text{LREE}$  ratios of the mantle source.

We observe a relatively constant and elevated  $\text{H}_2\text{O}/\text{Ce}$  ratio in melt inclusions over a wide range of  $^{87}\text{Sr}/^{86}\text{Sr}$ —melt inclusions with strong EM2 mantle signatures (i.e.,  $^{87}\text{Sr}/^{86}\text{Sr} > 0.7080$ ) have elevated  $\text{H}_2\text{O}/\text{Ce}$  ( $193 \pm 19$  2SD,  $N = 4$ ) that is indistinguishable from  $\text{H}_2\text{O}/\text{Ce}$  ( $202 \pm 31$ ,  $N = 3$ ) in melt inclusions with weaker EM2 signatures ( $^{87}\text{Sr}/^{86}\text{Sr} < 0.7080$ ). This also contrasts with Workman et al.’s (2006) observation that  $\text{H}_2\text{O}/\text{LREE}$  is lower in the Samoan EM2 pillow glasses with higher  $^{87}\text{Sr}/^{86}\text{Sr}$ . Unlike Workman et al. (2006), who argued that the low  $\text{H}_2\text{O}/\text{LREE}$  in Samoan EM2 glasses is a source feature, we show that the low ratio in these pillow glasses relative to that of the melt inclusions can be explained by closed-system degassing of magmas during ascent to the seafloor. If this degassing model is correct, and if Samoan high  $^{87}\text{Sr}/^{86}\text{Sr}$  and low  $^{87}\text{Sr}/^{86}\text{Sr}$  melts have the same initial  $\text{H}_2\text{O}/\text{Ce}$ —as supported by observations from melt inclusions in Fig. 5a—then the negative correlation between  $^{87}\text{Sr}/^{86}\text{Sr}$  in  $\text{H}_2\text{O}/\text{Ce}$  in the pillow glasses in Workman et al. (2006) (Fig. 7) leads us to an important conclusion: the Samoan EM2 pillow glasses have degassed more  $\text{H}_2\text{O}$  (to achieve lower  $\text{H}_2\text{O}/\text{Ce}$ ) than Samoan pillow glasses with weaker EM2 signatures (which have higher  $\text{H}_2\text{O}/\text{Ce}$ ). Therefore, we next focus on identifying the mechanism that results in greater  $\text{H}_2\text{O}$  degassing in more extreme EM melts relative to non-EM melts.

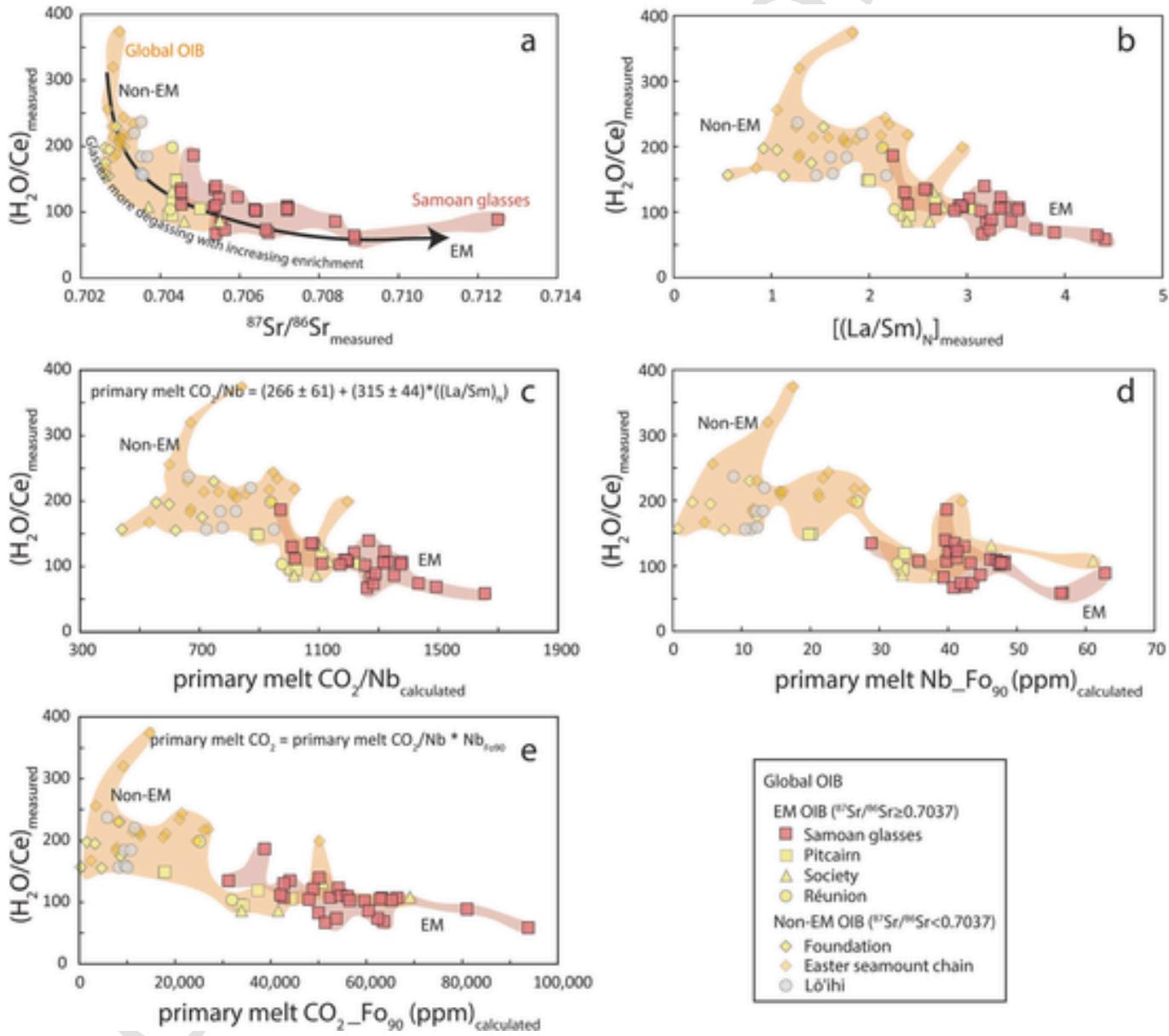
## 6.3. Explaining the inverse relationship between $^{87}\text{Sr}/^{86}\text{Sr}$ and $\text{H}_2\text{O}/\text{Ce}$ in global pillow glasses by degassing

The global OIB dataset on submarine pillow glasses exhibits an inverse relationship between  $^{87}\text{Sr}/^{86}\text{Sr}$  and  $\text{H}_2\text{O}/\text{Ce}$  (Fig. 7) that is similar to the inverse correlation observed in Samoan submarine glasses, and it has long been known that submarine glasses with low  $^{87}\text{Sr}/^{86}\text{Sr}$  tend to have higher  $\text{H}_2\text{O}/\text{Ce}$  than glasses with high  $^{87}\text{Sr}/^{86}\text{Sr}$  (e.g., Dixon et al., 2002; Workman et al., 2006; Kendrick et al., 2014). The Samoan melt inclusions noticeably stray from this trend. The Samoan melt inclusions with high  $^{87}\text{Sr}/^{86}\text{Sr}$  are shifted to higher, relatively constant  $\text{H}_2\text{O}/\text{Ce}$  ( $197 \pm 58$  2SD,  $N = 15$ ) compared to the  $\text{H}_2\text{O}/\text{Ce}$  ( $108 \pm 58$ , 2SD,  $N = 41$ ) in a global database of OIB glasses (including Samoa) that have similarly high  $^{87}\text{Sr}/^{86}\text{Sr}$  (i.e.,  $> 0.7037$ ) (Fig. 7). In fact, the high  $\text{H}_2\text{O}/\text{Ce}$  of Samoan inclusions ( $197 \pm 58$  2SD,  $N = 15$ ) is similar to the  $\text{H}_2\text{O}/\text{Ce}$  ( $209 \pm 92$ , 2SD,  $N = 32$ ) of pillow glasses characterized by low  $^{87}\text{Sr}/^{86}\text{Sr}$  ( $< 0.7037$ , the threshold value used to distinguish between EM and non-EM melts). If the Samoan melt inclusions represent less degassed versions of the melts sampled by Samoan submarine glasses, then the similarity in  $\text{H}_2\text{O}/\text{Ce}$  between low  $^{87}\text{Sr}/^{86}\text{Sr}$  ( $< 0.7037$ ) OIB pillow glasses and Samoan melt inclusions ( $> 0.7037$ ) raises the possibility that non-EM OIB melts and EM OIB melts initially have the same  $\text{H}_2\text{O}/\text{Ce}$ , but the latter suffer more degassing of  $\text{H}_2\text{O}$  than the former during magma ascent. If the inverse correlation between  $\text{H}_2\text{O}/\text{Ce}$  and  $^{87}\text{Sr}/^{86}\text{Sr}$  in Samoan, and global, OIB glasses is driven by greater degassing of the most geochemically enriched lavas compared to geochemically depleted lavas, what mechanism causes geochemically enriched glasses from OIB mantle sources to have suffered more  $\text{H}_2\text{O}$  degassing than geochemically depleted glasses?

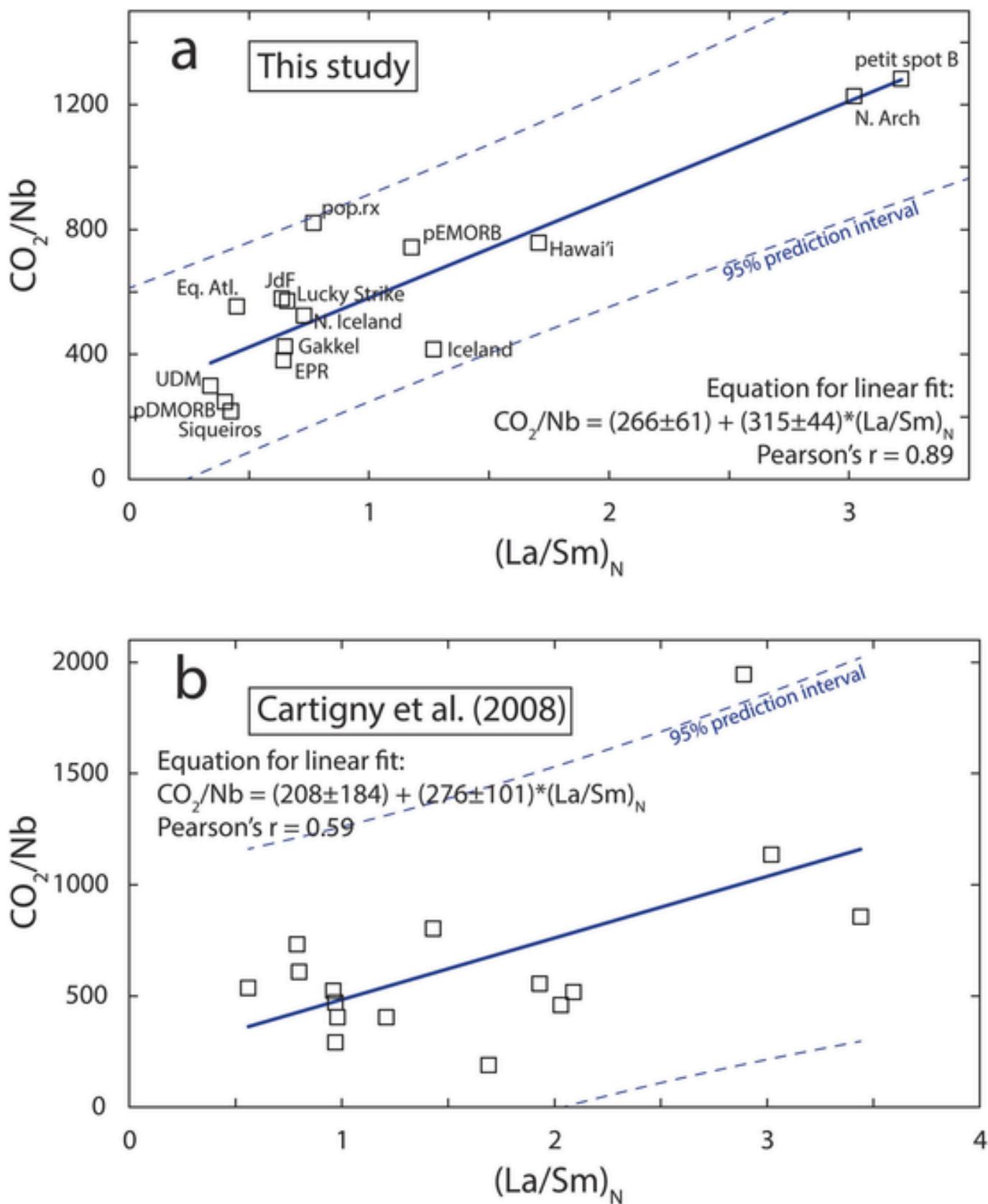
Previous work suggests that EM mantle sources have higher  $\text{CO}_2$  and generate primary melts with higher  $\text{CO}_2$  than non-EM mantle sources (Burnard et al., 2014; Cartigny et al., 2008; Taracsák et al., 2019; Hauri et al., 2018; Miller et al., 2019; Matthews et al., 2021; Michael and Graham, 2015; Shimizu et al., 2023). We show that primary melts with higher  $\text{CO}_2$  degas more  $\text{CO}_2$  and more  $\text{H}_2\text{O}$  by closed-system degassing during ascent than primary melts with the same  $\text{H}_2\text{O}$

but lower  $\text{CO}_2$  concentrations. As a result, EM lavas that were initially more  $\text{CO}_2$ -rich erupt on the seafloor with lower  $\text{H}_2\text{O}/\text{Ce}$  than their geochemically-depleted non-EM counterparts that started with lower primary melt  $\text{CO}_2$ , even though both EM and non-EM primary melts started with the same  $\text{H}_2\text{O}$  concentrations and  $\text{H}_2\text{O}/\text{Ce}$ . At the extreme end, non-EM (i.e., geochemically depleted) melts erupted at, for example, MORB settings, can have such low primary melt  $\text{CO}_2$  that they never saturate and neither  $\text{CO}_2$  nor  $\text{H}_2\text{O}$  are lost by degassing (Saal et al., 2002; Michael and Graham, 2015; Hauri et al., 2018).

By contrast, we argue that  $\text{CO}_2$ -rich primary melts, typical of geochemically-enriched OIB settings, saturate  $\text{CO}_2$  at great depths and lose significant  $\text{CO}_2$  and  $\text{H}_2\text{O}$  by closed-system degassing. In order to show this, we first estimate primary melt  $\text{CO}_2$  for a suite of global OIB glasses using a set of relationships shown in Figs. 8 and 9. Because OIB glasses are degassed in  $\text{CO}_2$  to variable degrees, we estimate primary melt (undegassed)  $\text{CO}_2$  for OIB glasses in the following way. First, using a new correlation between undegassed  $\text{CO}_2/\text{Nb}$  and  $(\text{La}/\text{Sm})_N$  in MORB and OIB (see equation in Fig. 9a and discussion in Supplementary Information section S1.8), together with the measured  $(\text{La}/\text{Sm})_N$  available in



**Fig. 8.** Samoan pillow glasses compared to a global dataset of submarine OIB glasses, which show a trend of decreasing  $\text{H}_2\text{O}/\text{Ce}$  with increasing  $^{87}\text{Sr}/^{86}\text{Sr}$ . The exact same OIB dataset is shown in all panels: only OIB with  $^{87}\text{Sr}/^{86}\text{Sr}$  data (to establish the level of source enrichment) and  $\text{MgO} > 4.5$  wt% (to minimize the impact of fractional crystallization) are shown. (a)  $\text{H}_2\text{O}/\text{Ce}$  versus  $^{87}\text{Sr}/^{86}\text{Sr}$ . (b)  $\text{H}_2\text{O}/\text{Ce}$  versus  $(\text{La}/\text{Sm})_N$ . (c)  $\text{H}_2\text{O}/\text{Ce}$  versus primary melt  $\text{CO}_2/\text{Nb}$ , where  $\text{CO}_2/\text{Nb}$  is calculated from the measured  $(\text{La}/\text{Sm})_N$  ratios using a linear fit between  $\text{CO}_2/\text{Nb}$  and  $(\text{La}/\text{Sm})_N$  (see equation shown in panel c) that is produced from a global data compilation of least degassed glasses and melt inclusions from OIB and MORB in Fig. 9. (d)  $\text{H}_2\text{O}/\text{Ce}$  versus calculated primary melt Nb for each sample, where primary melt Nb is calculated by olivine addition (or subtraction) from melt—following methods in text—until the melt is in equilibrium with  $\text{Fo}_{90}$  olivine. (e)  $\text{H}_2\text{O}/\text{Ce}$  versus calculated primary melt  $\text{CO}_2$ , where the  $\text{CO}_2$  is calculated for each lava by multiplying the calculated  $\text{CO}_2/\text{Nb}$  (panel c) with the calculated primary melt Nb (panel d) (see equation in panel e). The same sources of data and data treatment as that described in the Fig. 7 caption are used here. The calculated primary melt  $\text{CO}_2/\text{Nb}$  and  $\text{CO}_2\text{-Fo}_{90}$  (ppm) values for the global OIB data set presented here are given in Table S11. Only samples with  $\text{CO}_2\text{-H}_2\text{O}$  saturation pressures  $\geq 0.1$  kbar are shown (just one Samoan glass with  $^{87}\text{Sr}/^{86}\text{Sr}$  data has a saturation pressure  $< 0.1$  kbar, sample AVON3-68-03 Rpt, so its removal does not significantly impact the dataset).



**Fig. 9.** (a) Linear fit (blue line) of global MORB and OIB glasses and melt inclusions with respect to  $\text{CO}_2/\text{Nb}$  and  $(\text{La}/\text{Sm})_N$ . Data compilation is the same as what is presented in Fig. 1 of Hirschmann (2018). The values for  $\text{CO}_2/\text{Nb}$  and  $(\text{La}/\text{Sm})_N$  in panel (a) are given in Table S10. The blue dashed lines represent the 95% prediction interval. The 95% confidence interval of the linear fit is given with the equation of the linear fit. The linear relationship calculated in panel a is used to estimate

the suite of OIB glasses examined here, we can calculate the primary

**Fig. 9.**—continued

the  $\text{CO}_2/\text{Nb}$  values for glasses and melt inclusions in **Fig. 8**. (b) Linear fit (blue line) to the Mid-Atlantic Ridge  $\text{CO}_2/\text{Nb}$  versus  $(\text{La}/\text{Sm})_{\text{N}}$  data from [Cartigny et al. \(2008\)](#). The blue dashed lines are the 95% prediction bands for the linear fit using [Cartigny et al. \(2008\)](#) data. Sources of data for panel a: L. Strike (Lucky Strike): [Wanless et al. \(2015\)](#); JdF, EPR (Juan de Fuca, East Pacific Rise): [Wanless and Shaw \(2012\)](#); Gakkel: [Wanless et al. \(2014\)](#); Siqueiros: [Saal et al. \(2002\)](#); pEMORB and PDMORB (Pacific enriched and depleted MORB): [Shimizu et al. \(2016\)](#); UDM (ultra-depleted MORB): [Michael and Graham \(2015\)](#); pop.rx (North Atlantic popping rocks): [Cartigny et al. \(2008\)](#); Eq. Atl. (Equatorial Atlantic): [Le Voyer et al. \(2017\)](#); Iceland: [Hartley et al. \(2014\)](#); N. Iceland (Borgarhraun): [Hauri et al. \(2018\)](#); N. Arch (North Arch, Hawai'i): [Dixon et al. \(1997\)](#), [Frey et al. \(2000\)](#); Hawai'i (Kilauea): [Anderson and Poland \(2017\)](#); Petit Spot B (Western Pacific seamounts): [Machida et al. \(2015\)](#). (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

melt  $\text{CO}_2/\text{Nb}$  in the OIB glasses (**Fig. 8c**). (It is worth noting that the correlation between  $\text{CO}_2/\text{Nb}$  and  $(\text{La}/\text{Sm})_{\text{N}}$  in **Fig. 9a** is similar to a relationship previously suggested for MORB by [Cartigny et al. \(2008\)](#) (**Fig. 9b**.) We then calculate the primary melt Nb concentrations for each of the OIB glasses by correcting for olivine addition or subtraction so the melts are in equilibrium with mantle ( $\text{Fo}_{90}$ ) olivine (**Fig. 8d**). By multiplying calculated primary melt  $\text{CO}_2/\text{Nb}$  (**Fig. 8c**) by the calculated primary melt Nb concentrations (**Fig. 8d**), we obtain primary melt  $\text{CO}_2$  concentrations for the OIB glasses (**Fig. 8e**). These calculations suggest that the most extreme EM OIB glasses (which have the lowest  $\text{H}_2\text{O}/\text{Ce}$ )—all from the Samoa, Societies, and Pitcairn hotspots—tend to have the highest primary melt  $\text{CO}_2$  (**Fig. 8e**), with  $\text{CO}_2$  concentrations commonly as high as 5 to 6 wt% in Samoa and Societies lavas (and higher than 80,000 ppm  $\text{CO}_2$  in two Samoan lavas). These high OIB primary melt  $\text{CO}_2$  concentrations calculated with our new model are in broad agreement with  $\text{CO}_2$  concentrations calculated using the petrologic model of [Sun and Dasgupta \(2020\)](#) (see supplementary Fig. S9 for direct comparison).

Based on these calculations, Samoan primary melts inferred from the pillow rim glass data have  $\text{CO}_2$  concentrations between ~28,000 ppm and ~94,000 ppm, with most values clustering near ~60,000 ppm  $\text{CO}_2$  (**Fig. 8e**). With few exceptions, the non-EM OIB (which have the highest erupted  $\text{H}_2\text{O}/\text{Ce}$ )—Lō'ihi, Easter, and Foundation—have lower calculated primary melt  $\text{CO}_2$  than the EM OIB—Samoa, Societies, Pitcairn, and Réunion (**Fig. 8e**). This is consistent with prior suggestions that geochemically enriched mantle domains with higher  $^{87}\text{Sr}/^{86}\text{Sr}$  generate primary melts with higher  $\text{CO}_2$  than geochemically depleted mantle domains ([Burnard et al., 2014](#); [Cartigny et al., 2008](#); [Le Voyer et al., 2019](#)). Like [Burnard et al. \(2014\)](#), we argue that melts with higher initial  $\text{CO}_2$ —characteristic of EM melts (**Fig. 8e**)—undergo greater degrees of degassing than melts with lower initial  $\text{CO}_2$ . We then show that, everything else being the same, higher initial  $\text{CO}_2$  melts experience greater degassing of  $\text{H}_2\text{O}$  (and thus have lower erupted  $\text{H}_2\text{O}/\text{Ce}$ ) than melts with lower initial  $\text{CO}_2$  (**Fig. 8**).

In order to demonstrate this quantitatively, we use the calculated primary melt  $\text{CO}_2$  of the OIB glasses in **Fig. 8e**, as well as estimated primary melt  $\text{H}_2\text{O}$  (see below), as inputs to a closed-system degassing model. We then show that the  $\text{CO}_2$ -rich (60,000 ppm) model primary melt degasses by closed-system degassing to have  $\text{H}_2\text{O}$  concentration and  $\text{H}_2\text{O}/\text{Ce}$  similar to values measured in Samoan EM pillow glasses. To test our hypothesis that high primary melt  $\text{CO}_2$  concentrations result in more degassing of  $\text{H}_2\text{O}$ , and thus, lower  $\text{H}_2\text{O}/\text{Ce}$  in the final erupted and degassed melt, we examine how  $\text{H}_2\text{O}$  and  $\text{CO}_2$  concentrations change with decreasing pressure starting with two hypothetical primary melts that are identical in all respects except for the initial  $\text{CO}_2$  concentrations: the two melt endmembers are given the same starting  $\text{H}_2\text{O}$  (2 wt%),  $\text{H}_2\text{O}/\text{Ce}$  (200), and major and trace element chemistry (see compositions in Table S9). The first model melt is assigned a higher  $\text{CO}_2$  concentration of 60,000 ppm, meant to approximate the elevated  $\text{CO}_2$  concentration of EM glasses in **Fig. 8e** (see previous paragraph), and comes from our new model for calculating  $\text{CO}_2$  in OIB primary melts (**Fig. 8c, 9a**) for which calculated Samoan primary melt  $\text{CO}_2$  concentrations tend to cluster near 60,000 ppm  $\text{CO}_2$ . The second model melt is assigned an initial  $\text{CO}_2$  concentration of 5000 ppm, and is meant to approximate the  $\text{CO}_2$  concentration of the non-EM glasses from Foundation, Easter, and Lō'ihi in **Fig. 8e**. In order to illustrate the impact of

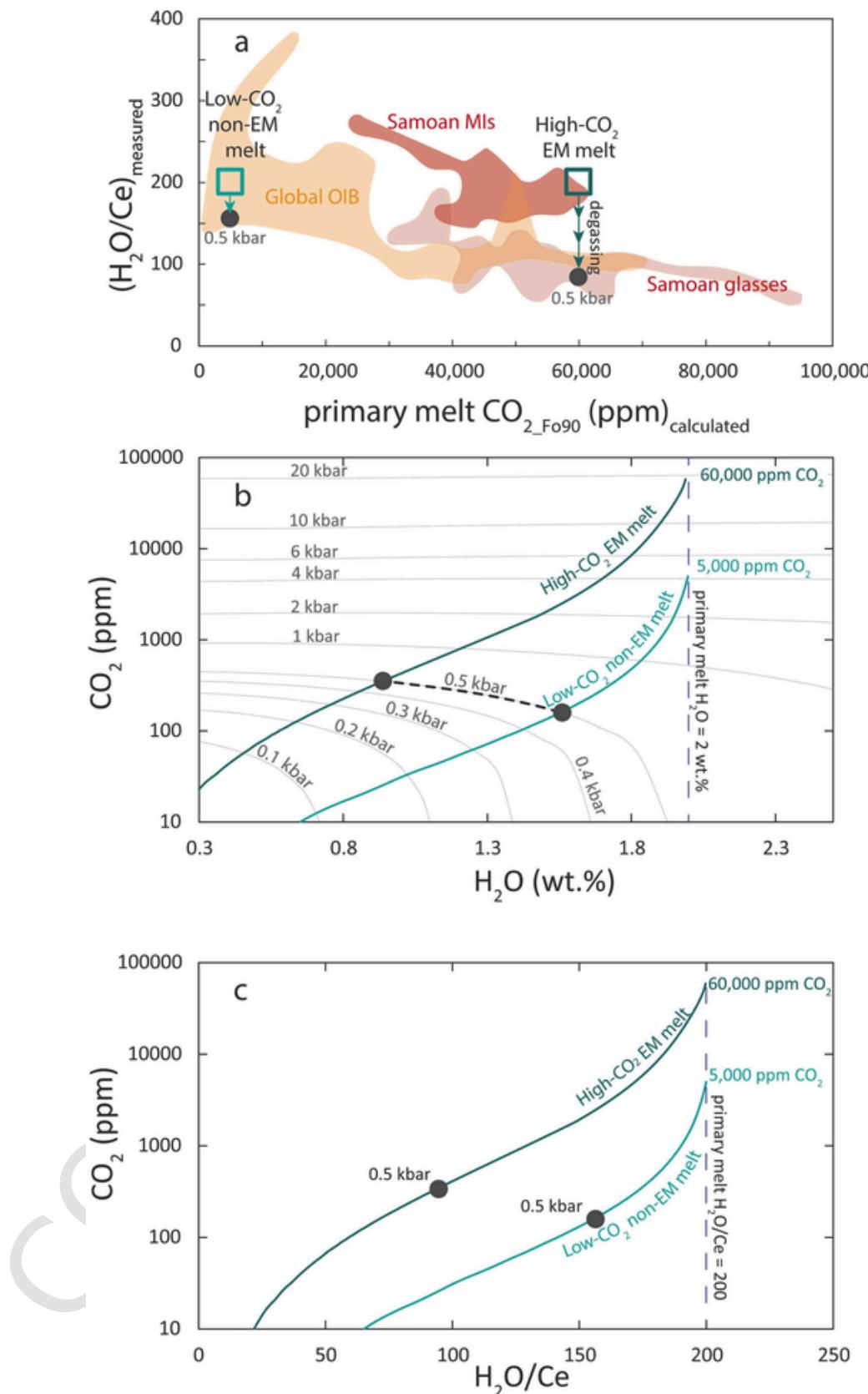
$\text{H}_2\text{O}$  loss during degassing as a function of initial primary melt  $\text{CO}_2$  concentration, we model the closed-system degassing paths of the two melts using rhyolite-MELTS. Degassing of the high  $\text{CO}_2$  (60,000 ppm) melt starts at 26 kbar (i.e., where  $\text{CO}_2$  first saturates; see dark green line in **Fig. 10**), which is higher than the 4 kbar pressure where the lower  $\text{CO}_2$  melt saturates (see light green line in **Fig. 10**). After both the high- and low- $\text{CO}_2$  melts degas to the same low eruptive pressure (**Fig. 10**)—0.5 kbar, meant to represent pressure for eruption on the seafloor at ~5000 mbsl—the high  $\text{CO}_2$  melt has a significantly lower  $\text{H}_2\text{O}$  concentration and lower  $\text{H}_2\text{O}/\text{Ce}$  (~0.94 wt% and 94, respectively) compared to the low  $\text{CO}_2$  melt (~1.56 wt% and 156, respectively), even though both melts started with the same  $\text{H}_2\text{O}$  (2 wt%) and  $\text{H}_2\text{O}/\text{Ce}$  (200). In summary, a melt with initial  $\text{CO}_2$  of 60,000 ppm degases nearly 53% of its  $\text{H}_2\text{O}$  (and 99.4% of its  $\text{CO}_2$ ) during ascent to 0.5 kbar, but the melt with 5000 ppm initial  $\text{CO}_2$  degases just 22% of its  $\text{H}_2\text{O}$  (and 96.8% of its  $\text{CO}_2$ ) during ascent to 0.5 kbar. Thus, everything else being the same, high  $\text{CO}_2$  melts degas more  $\text{H}_2\text{O}$  and have lower  $\text{H}_2\text{O}/\text{Ce}$  than low  $\text{CO}_2$  melts.

The simple degassing model in **Fig. 10** helps explain the observation that EM pillow glasses erupted on the seafloor have lower  $\text{H}_2\text{O}/\text{Ce}$  compared to non-EM pillow glasses: EM primary melts, which are associated with higher initial  $\text{CO}_2$  (**Fig. 8e**), saturate in volatiles deeper in the crust/mantle and degas more  $\text{CO}_2$  and  $\text{H}_2\text{O}$  prior to eruption than non-EM melts with lower primary melt  $\text{CO}_2$ . The Samoan melt inclusions in this study exhibit high  $\text{H}_2\text{O}/\text{Ce}$ , similar to non-EM melts, because  $\text{CO}_2$  and  $\text{H}_2\text{O}$  degassing was arrested following melt inclusion entrapment in olivine at high pressures (0.870 to 5.13 kbar). Although the Samoan melt inclusions lost significant amounts of  $\text{CO}_2$  prior to being entrapped by olivine, these melt inclusions still preserve higher  $\text{H}_2\text{O}/\text{Ce}$  than the erupted pillow glasses due to having degassed less  $\text{CO}_2$  and, thus, less  $\text{H}_2\text{O}$ . However, we acknowledge that, prior to entrapment, even the melt inclusions may have already lost some  $\text{H}_2\text{O}$  due to concomitant loss of both  $\text{H}_2\text{O}$  and  $\text{CO}_2$  during closed-system degassing. Nonetheless, melt inclusions preserve higher  $\text{H}_2\text{O}$  and  $\text{H}_2\text{O}/\text{Ce}$  than pillow glasses, and thus preserve values closer to the primary melts.

#### 6.4. Comparison with prior models for the origin of low $\text{H}_2\text{O}/\text{Ce}$ in EM lavas, and implications for a “damp” EM mantle

Previous studies have suggested that the negative correlation between  $\text{H}_2\text{O}/\text{Ce}$  and  $^{87}\text{Sr}/^{86}\text{Sr}$  is the result of dehydration ( $\text{H}_2\text{O}$  depletion) in the EM source, which has low  $\text{H}_2\text{O}/\text{Ce}$  that gives rise to EM primary melts with low  $\text{H}_2\text{O}/\text{Ce}$ . In these models, the EM mantle source has low  $\text{H}_2\text{O}/\text{Ce}$  because subducted materials contributing to the EM mantle lose  $\text{H}_2\text{O}$  during subduction ([Dixon et al., 2002](#)) or via diffusion during residence in the mantle ([Workman et al., 2006](#)), or because the pyroxenite—which has low  $\text{H}_2\text{O}/\text{Ce}$ —contributes to the EM mantle ([Bizimis and Peslier, 2015](#)). However, if these models explained the origin of the low  $\text{H}_2\text{O}/\text{Ce}$  in high  $^{87}\text{Sr}/^{86}\text{Sr}$  glasses from Samoa, then the Samoan melt inclusions with high  $^{87}\text{Sr}/^{86}\text{Sr}$  should also exhibit lower  $\text{H}_2\text{O}/\text{Ce}$ , but this is not the case.

Thus, an important implication is that submarine glasses cannot be reliably used to estimate the  $\text{H}_2\text{O}/\text{Ce}$  content of OIB primary melts, particularly glasses for EM OIB that appear to experience greater degassing (owing to higher initial  $\text{CO}_2$ ) than non-EM OIB. Our data show that, unlike Samoan submarine glasses, Samoan melt inclusions have similar



**Fig. 10.** Figure illustrating how submarine glasses from enriched mantle (EM) sources acquire low  $\text{H}_2\text{O}/\text{Ce}$  relative to glasses from non-EM sources. We test two melt compositions that are identical in every way (major and trace element compositions are identical and  $\text{H}_2\text{O}$  is 2 wt%) except for  $\text{CO}_2$ , where one melt starts with 60,000 ppm  $\text{CO}_2$  (meant to represent an EM melt) and the other melt starts with 5000 ppm  $\text{CO}_2$  (meant to represent the non-EM melt). (a)  $\text{H}_2\text{O}/\text{Ce}$  versus

**Fig. 10.**—continued

primary melt  $\text{CO}_2$ , reproduced in cartoon form from Fig. 8e. The black circle indicates the  $\text{H}_2\text{O}/\text{Ce}$  value for melts at 0.5 kbar after degassing. (b) In a plot of  $\text{CO}_2$  versus  $\text{H}_2\text{O}$ , modeled closed system degassing paths show the impact of varying initial  $\text{CO}_2$  on a melt that has the same initial  $\text{H}_2\text{O}$  content of 2 wt%. While both melts start with the same initial  $\text{H}_2\text{O}$ , those with higher initial  $\text{CO}_2$  degas more  $\text{H}_2\text{O}$  and have lower  $\text{H}_2\text{O}$  at a given isobar compared to a melt that has lower initial  $\text{CO}_2$ , demonstrating how higher initial  $\text{CO}_2$  results in lower  $\text{H}_2\text{O}$  (and lower  $\text{H}_2\text{O}/\text{Ce}$ ) in the erupted lavas. Degassing trends assume oxygen fugacity (QFM) and temperature (1200 °C) and are calculated in rhyolite-MELTS v1.2 using AVON3-78-1#3 major element melt inclusion composition. Bottom panel shows the same degassing paths as the upper panels but shows  $\text{H}_2\text{O}/\text{Ce}$  instead of  $\text{H}_2\text{O}$ . Calculation of  $\text{H}_2\text{O}/\text{Ce}$  along the degassing paths assumes 1) an initial  $\text{H}_2\text{O}/\text{Ce}$  value of 200 (similar to the average  $\text{H}_2\text{O}/\text{Ce}$  value of Samoan melt inclusions examined here) and 2) a constant melt Ce concentration (100 ppm) that is calculated using the initial  $\text{H}_2\text{O}/\text{Ce}$  ratio (i.e., 200) and the initial  $\text{H}_2\text{O}$  (2 wt%). Similar to the relationship between  $\text{CO}_2$  and  $\text{H}_2\text{O}$  for high and low  $\text{CO}_2$  melts, at a given isobar the melt with high initial  $\text{CO}_2$  has significantly lower  $\text{H}_2\text{O}/\text{Ce}$  compared to the melt with lower initial  $\text{CO}_2$ . Isobars are calculated using MagmaSat for the average Malumalum melt inclusion composition at 1200 °C. We consider only the case of  $M = 0$  to highlight the role of initial  $\text{CO}_2$  and saturation pressure on the  $\text{H}_2\text{O}$  content of a glass, and non-zero  $M$  values would result in even more rapid  $\text{H}_2\text{O}$  loss.

$\text{H}_2\text{O}/\text{Ce}$  ( $197 \pm 58$ , 2SD,  $N = 15$ ) that does not vary with  $^{87}\text{Sr}/^{86}\text{Sr}$ . In fact, the  $\text{H}_2\text{O}/\text{Ce}$  ratio ( $193 \pm 19$ , 2SD,  $N = 4$ ) for extreme Samoan EM2 ( $^{87}\text{Sr}/^{86}\text{Sr} > 0.708$ ) melt inclusions overlaps with  $\text{H}_2\text{O}/\text{Ce}$  suggested for depleted mantle domains that are sampled by non-EM OIB ( $^{87}\text{Sr}/^{86}\text{Sr} < 0.7037$ ;  $\text{H}_2\text{O}/\text{Ce} = 209 \pm 92$ ), average Pacific MORB ( $180 \pm 20$ ; Dixon et al., 2017), average depleted N. Atlantic MORB ( $230 \pm 20$ ; Dixon et al., 2017), and PREMA (Prevalent Mantle) Pacific and Atlantic OIB ( $215 \pm 30$  and  $220 \pm 30$ , respectively, where PREMA-type OIB do not bear EM or HIMU signatures; Dixon et al., 2017). Thus, the major implication of this work is that, instead of being “dry”, the EM2 mantle sampled by Samoa is just as “damp” as non-EM reservoirs, but this can only be seen by examining deeply entrapped melt inclusions that preserve higher  $\text{CO}_2$  and  $\text{H}_2\text{O}$  than submarine pillow glasses. Due to pronounced  $\text{CO}_2$  and concomitant  $\text{H}_2\text{O}$  degassing that appears to impact enriched lavas more than depleted ones—owing to higher primary melt  $\text{CO}_2$  in the former relative to the latter—future melt inclusion studies will be important for determining the initial  $\text{H}_2\text{O}/\text{Ce}$  in EM1 (Pitcairn) and other EM2 (Societies) lavas, where existing low  $\text{H}_2\text{O}/\text{Ce}$  values measured in high  $^{87}\text{Sr}/^{86}\text{Sr}$  Pitcairn and Societies submarine glasses (Kendrick et al., 2014) may also be due to the same degassing mechanism that lowers  $\text{H}_2\text{O}/\text{Ce}$  in Samoan OIB. If so,  $\text{H}_2\text{O}/\text{Ce}$  in the mantle is not likely to be as variable as previously supposed, and much of the mantle sampled by MORB and OIB have very similar  $\text{H}_2\text{O}/\text{Ce}$  that is, on average, close to a value of 200, and not as low as values found in pillow glasses of EM lavas (down to 59 in Samoan pillow glasses, 86 in Societies glasses, and 95 in Pitcairn glasses). Thus, like the  $\text{Pb}/\text{Ce}$  ratio, which is similar in OIB and MORB globally (Hofmann et al., 1986), the  $\text{H}_2\text{O}/\text{Ce}$  ratio may also be similar in the sources of plume-derived lavas (Michael, 1995).

A second implication of “damp”  $\text{H}_2\text{O}/\text{Ce}$  in EM Samoan lavas is that, in spite of hosting subducted continental materials which have presumably lost  $\text{H}_2\text{O}$  by degassing at an ancient subduction zone (White and Hofmann, 1982; Farley et al., 1992; Jackson et al., 2007), Samoan EM2 melt (as sampled by EM melt inclusions) do not have lower  $\text{H}_2\text{O}/\text{Ce}$  than non-EM OIBs. This is a surprising result, and suggests that recycled continental materials may not play a large role in governing the  $\text{H}_2\text{O}$  budgets of mantle domains. This could be because the amount of recycled continental crust in the mantle sources of OIB is small, and thus has a diminished impact on the overall  $\text{H}_2\text{O}$  and  $\text{H}_2\text{O}/\text{Ce}$  of the mantle source. For example, Samoan lava AVON3-78-1 examined in this study, which has a strong EM2 signature ( $^{87}\text{Sr}/^{86}\text{Sr}$  of 0.7089), is estimated to have only ~1% continental crust (Reinhard et al., 2018) that is added to a depleted plume component that constitutes the rest (99%) of the mantle source. In this model, the  $\text{H}_2\text{O}$  and Ce budgets of the EM2 mantle source is primarily controlled by the depleted plume component, even if the recycled continental crust component has low  $\text{H}_2\text{O}/\text{Ce}$ . In this way, both depleted mantle and EM OIB can have similar  $\text{H}_2\text{O}/\text{Ce}$ .

#### CRediT authorship contribution statement

**O.E. Anderson:** Conceptualization, Formal analysis, Investigation, Project administration, Visualization, Writing – original draft, Writing – review & editing, Methodology. **M.G. Jackson:** Conceptualization,

Funding acquisition, Writing – review & editing, Project administration, Resources, Supervision. **A.S. Pamukçu:** Investigation, Supervision, Writing – review & editing. **E.F. Rose-Koga:** Investigation, Writing – review & editing. **V. Le Roux:** Resources, Writing – review & editing. **F. Klein:** Methodology, Resources, Writing – review & editing. **K.T. Koga:** Formal analysis, Methodology, Writing – review & editing. **G.A. Gaetani:** Resources, Supervision, Writing – review & editing. **A.A. Price:** Investigation, Supervision, Writing – review & editing.

#### Uncited references

Le Voyer et al., 2019  
Ryan et al., 2009

#### Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

#### Data availability

The data is shared as a supplementary Excel spreadsheet.

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#### Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.chemgeo.2024.121979>.

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