



# Corals in ocean acidification and the role of calcium ion homeostasis to maintain calcification

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## Abstract

Coral calcification is essential to provide the structural foundation for coral reefs and is integral in supporting marine biodiversity reliant on reef ecosystems. The drivers for calcification in corals are undoubtedly highly complex and require several perspectives to identify vulnerabilities in the context of environmental change. Specifically, ocean acidification (OA) resulting from the rise of anthropogenic carbon dioxide (CO<sub>2</sub>) emissions poses a potential threat to the physiological mechanisms that drive calcification in corals. Therefore, this report goes beyond environmental seawater chemistry to examine the physiological mechanism of calcium ion homeostasis. Calcium's role in calcification physiology is well established, but how calcium homeostasis could shift under acidification has little been considered a significant driver in reduced calcification. Calcium is potentially the most actively transported substrate in coral calcification, though in high chemical abundance in seawater, corals are likely utilizing the most energy to concentrate calcium at the site of calcification. We argue for increased consideration of the calcium ion in the context of OA when identifying sensitivities. The concepts proposed here are justified through a combination of results from novel RAMAN spectroscopy and molecular work that provides insight into shifts in calcium homeostasis when exposed to acidification. We speculate that future work incorporating methodologies considering calcium dynamics in OA could benefit by narrowing in on what physiological mechanisms are potentially vulnerable. It is imperative that we identify what drives lower calcification in corals under OA to inform efficient directives in identifying species sensitivities to future climate change.

**Keywords:** coral reef; calcium carbonate; aragonite saturation state ( $\Omega$ ); pCO<sub>2</sub>; climate change; organismal physiology

## Environmental perspective on tropical coral calcification

Colonial corals are animals that provide the primary structural framework for coral reefs, which support some of the world's most productive and diverse communities (Smith 1978, Hoegh-Guldberg 2011, Al-Horani 2015). The biogenic growth process of synthesizing calcium carbonate (CaCO<sub>3</sub>), termed calcification, is fundamental to this framework, and without it we could see rapid declines in reef building corals ability to support marine communities through their structures (Hoegh-Guldberg et al. 2007, Graham and Nash 2013, Bahr et al. 2016, Jokiel 2016). Corals and coral reefs will face global ocean change, including ocean warming and acidification, in the future as atmospheric carbon dioxide (CO<sub>2</sub>) increases, threatening the existence of reefs worldwide (Kleypas et al. 1999, Caldeira and Wickett 2003, Sabine et al. 2004, Jokiel 2016). Predictions for the years 2081–2100 show that pH levels are *virtually certain* to decrease by 0.28–0.29 pH units (RCP8.5) and *very likely* to warm 2–4 times (low emission RCP2.6; IPCC 2022). There has been growing interest in understanding how lowered pH will impact biological functions, with 1500 documented publications in marine taxa between 2008 and 2023 (OA-ICC 2023). Specifically for coral growth, calcification responses broadly vary and currently range from +45% to –100% relative to ambient conditions in tropical corals (Reynaud et al. 2003, Langdon and Atkinson 2005, Andersson and Mackenzie 2012, Chan and Connolly 2013). Global net ecosystem calcification due to lowered pH

was predicted to decline on average between 14% and 30% (Kleypas et al. 1999, Chan and Connolly 2013). Ocean acidification (OA) will not be the sole driver for coral reef declines in the future (Klein et al. 2024). However, experimental work in OA allows us to tease apart the physiological burden of maintaining calcification in chemically stressful but sublethal conditions. This is particularly valuable for understanding the mechanisms behind calcification and creating a baseline to apply toward future interactive stressors.

Coral calcification is a function of precipitating CaCO<sub>3</sub> and thus requires both calcium (Ca<sup>2+</sup>) and dissolved inorganic carbon (DIC; i.e. CO<sub>2</sub>, HCO<sub>3</sub><sup>–</sup>, and CO<sub>3</sub><sup>2–</sup>). Initially, the conventional theory used to connect the seawater environment's interactions with the drivers of calcification is the utilization of aragonite saturation state ( $\Omega_{\text{arag}}$ ) values (Kleypas et al. 1999, Marubini et al. 2001, Langdon and Atkinson 2005, Castillo Alvarez et al. 2024).

$$\Omega_{\text{arag}} = [\text{Ca}^{2+}] [\text{CO}_3^{2-}] / K_{\text{sp}} \quad (1)$$

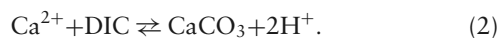
Equation 1 shows concentrations of calcium and carbonate ([Ca<sup>2+</sup>] and [CO<sub>3</sub><sup>2–</sup>], respectively) and  $K_{\text{sp}}$  as the solubility constant of aragonite (Dickson et al. 2007). Aragonite is a polymorph of CaCO<sub>3</sub> primarily used by corals, which has distinct constants and chemical behaviors in seawater compared to other polymorphs found in calcifying organisms (Dickson et al. 2007). In the context of OA, the  $\Omega_{\text{arag}}$  parameter decreases as a function of carbonate ion concentrations, resulting in a hypothesized substrate limitation of calcification

(Langdon and Atkinson 2005, Castillo Alvarez et al. 2024). However, in the present day, this theory has been confounded by reports identifying the  $\Omega_{\text{arag}}$  parameter as dependent on processes such as calcification and production, not the inverse (Jokiel et al. 2014, DeCarlo et al. 2017a, Davis et al. 2021). Expanding on the influence of seawater chemistry, some studies suggest that bicarbonate ion concentrations ( $[\text{HCO}_3^-]$ ) drive calcification sensitivity in corals (Jury et al. 2010, Bach 2015). However, molecular research indicates that corals possess only physiological pathways for  $\text{HCO}_3^-$  ion transporters (Zoccola et al. 2004), challenging the  $\text{CO}_3^{2-}$  based perspective. Furthermore, Bach et al. (2015) and Jokiel (2016) also present the seawater ratios of bicarbonate or dissolved inorganic carbon to protons ( $[\text{HCO}_3^-]$  or  $[\text{DIC}]: [\text{H}^+]$ ) as having greater explanatory power compared to  $\Omega_{\text{arag}}$  under OA. Lastly, it has been discussed that net ecosystem marine heterotrophy and productivity could drive net ecosystem calcification independent of pH or  $\Omega_{\text{arag}}$  and thus the ratios of inorganic carbon to protons (Kealoha et al. 2019). Therefore, this report goes beyond environmental seawater chemistry to examine the physiological mechanisms that drive calcification in corals (Chan and Connolly 2013).

To expand our perspective on calcification in a future acidified ocean, our objective is to propose the factor of calcium ion dynamics as a potential driver for reduced calcification in corals. Specifically, when exposed to low pH, little focus has been put on the calcium ion's role in calcification's response to acidification (DeCarlo et al. 2018). We acknowledge that OA does not chemically limit the calcium ion concentration in seawater. Therefore, our stance is derived purely from a physiological perspective. We are confident of calcium ion's direct importance to calcification through foundational studies manipulating seawater availability (Gattuso 1998) and within coral tissues (Marshall 1996, Tambutté et al. 1996). However, if the physiological burden of maintaining calcium homeostasis is too steep, then calcification in corals could suffer. Therefore, we argue that the physiological dynamics of calcium flux should be further examined under changing ocean chemistry and considered as a potential driver of lower calcification in corals.

## What is physiologically driving calcification?

Calcification occurs when calcium combines with dissolved inorganic carbon (i.e.  $\text{CO}_2(\text{aq})$ ,  $\text{HCO}_3^-$ , or  $\text{CO}_3^{2-}$  as DIC), synthesizing  $\text{CaCO}_3$  and two protons ( $\text{H}^+$ ) in excess (Jokiel 2011a, 2011b, 2016).



Jokiel (2011a) proposed the boundary layer limitation—proton flux model, arguing for increased representation of calcification rates away from  $\Omega_{\text{arag}}$  (i.e. ratios of DIC to  $\text{H}^+$ ; Bach 2015). Ocean acidification causes the hydrogen ion concentration ( $[\text{H}^+]$ ) to increase (decreasing pH) and thus could increase the diffusive gradient following Fick's First Law of diffusion on the flux of protons (Eq. 2; Jokiel et al. 2014, Jokiel 2016). Protons would act as calcification inhibitors. In support of proton flux, aspects in Comeau et al. (2019) and Venn et al. (2022) show that the boundary layer's pH correlates with the extracellular calcifying fluid's (ECF) pH and shows a reduction in interstitial pH under OA. It is worth noting, however, that the calcifying fluid's pH does not always explain reduced calcification under OA, resulting in a heterogeneity

of responses across species (Comeau et al. 2019). However, other contributions to the variation of calcification responses have been shown i.e. life stages (Albright and Langdon 2011), growth rate (Comeau et al. 2014), methodology (Chan and Connolly 2013), and heterotrophy (Kealoha et al. 2019). Despite these influences, aragonite crystal nucleation in tropical corals occurs within the ECF, where the chemical environment facilitates coral skeleton precipitation (Allemand et al. 2011, Al-Horani 2015).

The ECF is mainly isolated from the external seawater and thus relies greatly on biological regulation to support calcification (Allemand et al. 2004, 2011). This biological regulation consists of ion transporters, such as those that concentrate calcium and bicarbonate (Marshall 1996, Allemand et al. 2011, Hohn and Merico 2015, Reymond and Hohn 2021) and/or export protons (Allemand et al. 2011). This regulation is usually energy-dependent, supported by metabolism from heterotrophy or photosynthesis (Allemand et al. 2011). Hohn and Merico (2015) modeled physiological ion-flux energy budgets for calcification in detail. However, briefly, the most likely pathway for all ion flux (transcellular and paracellular as “H4”) requires a positive energy budget of 0.676 mol ATP (mol  $\text{CaCO}_3$ ) to precipitate 1.11 mmol  $\text{CaCO}_3 \text{ m}^{-2} \text{ d}^{-1}$ . Where ions transcellularly concentrated are calcium at 1.30 mmol  $\text{m}^{-2} \text{ d}^{-1}$  (active or passive), bicarbonate at 0.0001 mmol  $\text{m}^{-2} \text{ d}^{-1}$  (active), and  $\text{CO}_2$  at 1.26 mmol  $\text{m}^{-2} \text{ d}^{-1}$  (passive) along with paracellularly concentrated/transported bicarbonate at 0.0163 mmol  $\text{m}^{-2} \text{ d}^{-1}$  (passive). Conversely, transcellularly effluxed are protons at  $-2.59 \text{ mmol m}^{-2} \text{ d}^{-1}$  (active; Hohn and Merico 2015). Thus, in a substrate comparison, calcium is concentrated transcellularly (active or passive) in the ECF 195.02% greater than bicarbonate and one-to-one with  $\text{CO}_2$  (passive; calculated from Hohn and Merico 2015). Leading calcium to be the highest transported substrate in coral calcification, though in high abundance in seawater, corals are likely utilizing the most energy to concentrate calcium at the ECF (Eq. 1). Protons are thus the leading active effluxed material from the ECF, and in concert with calcium regulation, we begin to define the dynamics of energetic maintenance of substrate to waste products in the physiology of coral calcification.

Transcellular active transport requires the hydrolysis of ATP to actively concentrate materials across the diffusion gradient, which is an energetically costly process (Allemand et al. 2011, Al-Horani 2015). Where high-energy budgets exist, disruptions in homeostasis caused by environmental change could impact the energy-dependent physiological pathways. Therefore, calcium (substrate) and proton (waste) regulation in calcification could be the most influential physiological mechanisms for regulating calcification in environmental change. There is growing evidence on the impact of ecological change regarding proton flux (Jokiel 2011a, Bach 2015, Bahr et al. 2018, Comeau et al. 2019, Venn et al. 2022), little work has focused on calcium flux as a function of environmental change, specifically under OA or ocean warming (DeCarlo et al. 2018).

## The calcium ion and coral physiology

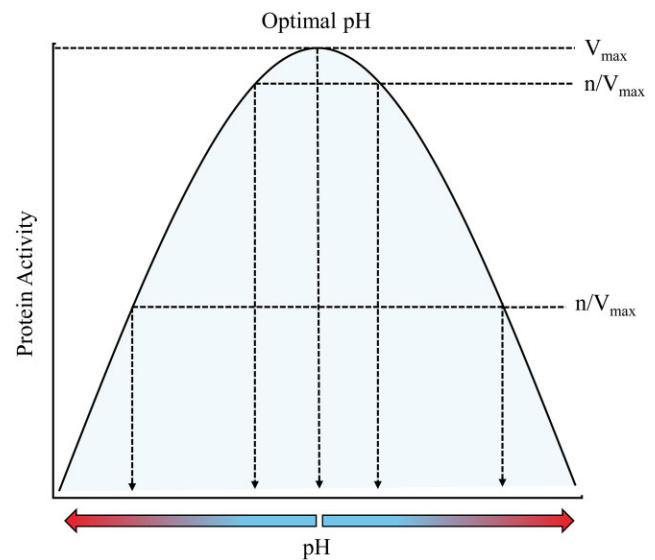
The quantification of energy budgets for transcellular calcium transport related to calcification by Hohn and Merico (2015) can be supported by studies addressing coral ion transport dynamics. Briefly, transcellular calcium transport occurs via

L-type voltage-gated channel proteins (Marshall 1996, Zoccola and Failla et al. 1999) and plasma membrane  $\text{Ca}^{2+}/\text{H}^{+}$  ATPase (PMCA) proteins (Zoccola et al. 2004). The proteins here are essential for concentrating calcium in the ECF, promoting  $\text{CaCO}_3$  nucleation. In Reymond and Hohn (2021), the ratios of calcium and carbonate ( $\text{Ca}^{2+}:\text{CO}_3^{2-}$ ) were assessed in a mimicked ECF of corals and found a 3x increase in calcification in a high  $\text{Ca}^{2+}:\text{CO}_3^{2-}$  (higher  $[\text{Ca}^{2+}]$ ) compared to a low  $\text{Ca}^{2+}:\text{CO}_3^{2-}$  (lower  $[\text{Ca}^{2+}]$ ). Indicating greater nucleation of aragonite with a higher concentration of calcium in the ECF specifically (Reymond and Hohn 2021). Conversely, they suggest magnesium removal mechanisms play a significant role in mineralization, though mentioning overall the mechanism for calcium concentration and magnesium removal to explain taxon-specific responses to OA in low  $\Omega_{\text{arag}}$  (Reymond and Hohn 2021). Using an Ussing chamber and studying electrophysical properties in coral ion transport, Taubner et al. (2017) showed tight regulation of the ECF through active ion transport. They concluded that active transport of calcium was required to maintain calcification rates in corals (Taubner et al. 2017). Therefore, the active biological regulation of calcium concentration and proton export at the ECF is essential for calcification in corals.

Focusing on PMCA, calcium concentration and proton export are simultaneous, and it has been shown that pH increases in the ECF along with calcium concentrations (Al-Horani et al. 2003). Though this correlation is not only a function of PMCA (i.e. several other active proteins used in the regulation of ions independently), based on energy conservation, PMCA would be the most efficient mechanism in regulating calcium and protons influencing substrate dynamics and waste products in coral calcification. From a stoichiometric perspective, PMCA consumes ATP at a ratio of 2:4:1 ( $\text{Ca}^{2+}:\text{H}^{+}:\text{ATP}$ , respectively) compared to singularly pumping protons at 4:1 ( $\text{H}^{+}:\text{ATP}$  respectively; Hohn and Merico 2015). Thus, potentially, being more energetically favorable to utilize PMCA by corals in regulating the ECF rather than separate ion transport mechanisms. Therefore, considering the impact, calcium concentrations in the ECF have on calcification (Reymond and Hohn 2021) and the need for active ion transport of calcium to sustain calcification (Taubner et al. 2017), PMCA is a valuable mechanism for corals to sustain calcification (Al-Horani et al. 2003, Hohn and Merico 2015). This concept is well understood. However, most of the attention PMCA has received as impacted by OA lies in the export of protons from the ECF, not its ability to supply calcium (Jury et al. 2010, Cyronak et al. 2016, Allison et al. 2018). Theoretically, both processes are vulnerable to acidification, as is the entire PMCA enzyme, and potentially shifting the focus to calcium may provide insight into coral sensitivity to calcification.

### What we know: ocean acidification and calcium dynamics

OA has been shown to decrease the pH of the immediate environment (defined as the mesoglea) above the cells that carry out ion transport for the ECM, decreasing by 0.58–0.71 pH units relative to ambient conditions (Venn et al. 2022). Thus, essential transport mechanisms, i.e. PMCA and proton pumps responsible for the regulation of the ECF, are exposed to decreased pH from OA. There has been no work on the direct measures of kinetic reaction differences between pH environ-



**Figure 1.** Theoretical schematic of protein function dependent on pH. Where the highest activity ( $V_{\text{max}}$ ) is shown, as pH increases (right) or decreases (left), the resulting  $V_{\text{max}}$  is a fraction ( $n$ ) of optimal values (Bisswanger 2014).

ments in marine invertebrates concerning the two transport mechanisms. However, from a crude biochemical perspective, both transporters (similarly with all enzymes) have optimal pH ranges for reaction efficiency, recently addressed in Bisswanger (2014) and visually represented in Fig. 1 (Allison et al. 2018). All enzymes rely on similar tolerance thresholds, not limited to pH, where temperature also plays a significant role (Bisswanger 2014, Allison et al. 2018).

The  $V_{\text{max}}$  of PMCA in marine invertebrates involved in calcification has not been characterized, and we are unaware of the rate limitations ( $n/V_{\text{max}}$ ) imposed by acidification. This assay has also not been done for any calcification supporting ion-transport mechanisms in marine invertebrates. A rate-limiting disequilibrium could disrupt calcium transport and, in concert, proton export from the ECF, potentially impacting calcification in corals. Molecular work that could be used to support this concept was shown recently by Radice et al. (2024), which utilized RNA-seq methodologies to highlight the phenotypic plasticity of corals in an acidified environment. We find results from Radice et al. (2024) extremely valuable as corals were exposed to low seawater pH for ~2 years *in situ*, compensating for a significant source of variability in *ex-situ* examination. They found significant gene upregulation of P-type calcium ATPases (GO:0005388) and the downregulation of genes associated with V-type proton ATPases (GO:0046961, GO:0008553, GO:0016471)—the difference herein being a higher priority of transporting calcium and exporting protons over exporting protons alone (Fig. 2). As discussed, stoichiometric energy conservation would favor PMCA over singular proton pumps. However, if proton concentrations were the sole driver for this stress response, we could assume the animal also upregulates or maintains normal expression of proton ATPases (potentially a more efficient/rapid mechanism). We speculate that instead, the observation of downregulation by Radice et al. (2024) could point toward prioritizing calcium, as the major difference between both transporters is the movement of calcium to the ECF. Not only was PMCA upregulated, but





have received significant attention in direct measures or manipulation studies (DeCarlo et al. 2018).

We feel that the arguments here provide sufficient justification to investigate calcium dynamics under OA and interacting stressors, combining methods in DeCarlo et al. (2018) utilizing RAMAN spectroscopy and Radice et al. (2024) as an omics approach. Proper quantification in concentrations of calcium and carbonate in the ECF, correlated with phenotypic plasticity as a function of gene expression, would provide valuable information on calcium's role in calcification under OA. If calcium is a significant driver in lower calcification in corals, then the animal's ability to maintain calcium homeostasis in the ECF would determine coral species-specific responses to OA (DeCarlo et al. 2018). We also suggest building from these data in transgenerational studies, with targeted directives in calcium dynamics and other physiological mechanisms that regulate the ECM. Assessing how gene expression for ion-regulatory mechanisms changes over generations could define potentially vulnerable species concerning the adaptive capacities of these individuals (Yuan et al. 2019, Radice et al. 2024). For example, if calcium dynamics drive lower calcification in corals under OA, assessments of the transgenerational ability to maintain calcium homeostasis would define a coral's adaptive capacity (Figs 1 and 2). However, we need to first identify the driver(s) for lower calcification in corals, and we argue that calcium dynamics should be included.

Concerning big picture climate change, OA is not occurring independently, and corals will face many stressors where interactive effects could add complexities to predicting their response in the future (Bahr et al. 2018, Leung et al. 2022, Klein et al. 2024). It is important to isolate the predicted stressors to understand baseline sensitivities, but these data provide a limited context to the state of future coral reefs (Klein et al. 2024). Utilizing the methodologies described above with interacting stressors could provide a valuable comparison for the calcification of corals in the future. The state of reefs will ultimately be determined by the ability of sessile calcifiers such as corals to overcome the effects of climate change. We would suggest that when the drivers for lower calcification can be confidently identified in an OA context, these studies should be furthered with predicted disease, local threats, temperature, sedimentation, and population shifts to understand the state of future coral reefs. As for the sessile benthic calcifier with limited physiological tolerance ranges, it is imperative to identify if adaptation is feasible in a future ocean.

In conclusion, expanding on calcium dynamics and coral calcification under OA could be crucial to identify what could be driving lower calcification in corals. By integrating novel geochemical methods (i.e. RAMAN spec) with molecular analyses of ion-transporter genes, we can isolate how corals allocate energy to maintain ECF homeostasis in our future ocean. As anthropogenic CO<sub>2</sub> emissions continue to drive climate change, knowledge of the drivers for calcification is essential for safeguarding the future of coral reefs and the ecosystems they support.

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## Author contributions

David A. Armstrong (Conceptualization [lead], Data curation [lead], Investigation [lead], Writing – original draft [lead], Writing – review & editing [lead]), and Keisha D. Bahr (Funding acquisition [lead], Investigation [supporting], Project administration [lead], Resources [lead], Supervision [lead], Writing – review & editing [equal]).

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## Data availability

There are no new data associated with this article.

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